# Feasibility studies on the utilization of fresh water microalgal species in Carbon dioxide sequestration

Thesis submitted to the University of Calicut in partial fulfillment of the requirement for the degree of

DOCTOR OF PHILOSOPHY IN BOTANY

by

# **KARTHIKA S. MENON**



DIVISION OF ENVIRONMENTAL SCIENCE DEPARTMENT OF BOTANY UNIVERSITY OF CALICUT KERALA - 673635

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Dr. C.C. Harilal

Associate Professor

UNIVERSITY OF CALICUT DEPARTMENT OF BOTANY Division of Environmental Science P.O. Calicut University, Tenhipalam Malappuram District, Kerala – 673 635 Contact: 09447956226 Mail: ccharilal22@gmail.com

Coordinator / Head, Department of Environmental Science, University of Calicut

# CERTIFICATE

This is to certify that the thesis entitled **"Feasibility studies on the utilization of fresh water microalgal species in carbon dioxide sequestration"**, submitted to the University of Calicut by Mrs. Karthika S. Menon, in partial fulfillment of the award of the degree of Doctor of Philosophy in Botany is a bonafide record of the research work carried out by her under my guidance and supervision.

No part of the present work has formed the basis for the award of any other degree or diploma, previously.

University of Calicut 10<sup>th</sup> June 2018

**Dr. C. C. Harilal** (Supervising Teacher)

# **DECLARATION**

The thesis entitled **"Feasibility studies on the utilization of fresh water microalgal species in carbon dioxide sequestration"** submitted by me in partial fulfillment of the requirements for the award of the degree of Doctor of Philosophy in Botany of the University of Calicut is an original research work carried out by me under the guidance and supervision of Dr. C.C. Harilal, Associate Professor, Department of Botany, University of Calicut. No part of the work has formed the basis for the award of any other Degree or Diploma of any University.

University of Calicut 10<sup>th</sup> June 2018 Karthika S. Menon

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# **ABBREVIATIONS**

$CO_2$	:	Carbon dioxide
BB medium	:	Bolds Basal medium
PBR	:	Photo bio reactor
DO	:	Dissolved oxygen
EDTA	:	Ethylenediaminetetraacetic acid
HCl	:	Hydrochloric acid
NTU	:	Nephelometric Turbidity Unit
°C	:	Degree Celsius
μs	:	micro seimens
kΩ	:	kilo ohm
μm	:	micrometer
ml	:	milliliter
ppm	:	parts per million
mg/l	:	milligram per litre
mg/ml	:	milligram per millilitre
nm	:	Nanometer

Uncontrolled greenhouse gas emissions due to human activities have contributed substantially to global warming and climate change. The greenhouse gases mainly include Carbon dioxide ( $CO_2$ ), Methane ( $CH_4$ ) and oxides of Nitrogen (NOx). Of these,  $CO_2$  is of major concern today, owing to its higher concentration in the atmosphere.

Among various strategies for  $CO_2$  sequestration, biological sequestration using photosynthetic microalgae have received considerable attention in recent times. Microalgae, one of the most important living resources of both fresh and marine systems can be employed for  $CO_2$  sequestration, as they have higher photosynthetic efficiency, higher biomass production and faster growth rate, compared to other energy crops. They can readily be incorporated into engineered systems.

The present study is an attempt to assess the potentialities of indigenous freshwater microalgal species in carbon dioxide sequestration. The specific objectives outlined in the present study in this direction include:

- Maintenance of pure cultures of microalgal species using standard methods and selection of microalgal species which are active under culture conditions.
- Monitoring the responses of selected microalgal species under varying dosages of carbon dioxide supply.
- Determining the optimum pH favouring maximized growth of selected micro algal members.

• Assessment of the CO<sub>2</sub> assimilation efficiency of selected micro algal members in PBR under controlled conditions.

For a meaningful elucidation of the objectives, the study has been undertaken in three stages and their outcomes are depicted in three chapters. They include:

- I. Screening of microalgae for CO<sub>2</sub> assimilation efficiency
- II. pH specific modification of culture medium for growth maximization of *Chlamydomonas globosa* and *Acutodesmus obliquus*
- III. PBR based feasibility studies on the carbon sequestration efficiency of selected microalgal members

Chapter I is dealing with microalgal culturing and subsequent assessment of their CO<sub>2</sub> assimilation efficiency. In this study, microalgal samples were collected from heterogeneous fresh water environments and were subjected to culturing in Bolds Basal medium. Accordingly pure cultures of *Chlamydomonas grovei, Chlamydomonas globosa, Desmodesmus opliensis, Monoraphidium contortum* and *Acutodesmus obliquus* were maintained in the laboratory. Screening studies were carried out on these microalgal members with respect to their efficiencies in assimilating varying levels of carbon dioxide supply like 10, 15, 20, 25, 30, 40, 60, 80 and100 bubbles/2 hours. The results were analyzed and the microalgal members, which were effective in  $CO_2$  assimilation, were listed out.

Chapter II is dealing with studies which were carried out to maximize the biomass production of selected microalgal members by altering culture conditions. As pH was noted to be an important factor influencing the growth of microalgal members, an experimental approach has been initiated with an objective to evaluate the optimum pH favouring the maximized growth of C.

*globosa* and *A. obliquus* in Bolds Basal medium which has identified as excellent candidates for carbon sequestration. For experimentation, treatments were maintained using Bolds Basal medium in which the pH of culture medium was adjusted from 3-12, with a gradation of 0.5. The results were analyzed and the optimum pH influencing the growth and multiplication of micro algal members like *C. globosa* and *A. obliquus* were worked out.

As *C. globosa* and *A. obliquus* exhibited better growth performances in higher dosages of carbon dioxide, an attempt has been carried out to assess their efficiencies in carbon dioxide accumulation using a proto type Photo Bio Reactor (PBR). For satisfying this objective, a laboratory scale closed vertical bubble column PBR having a size of 10 x 10 x 45 cm and a volume of 4.5 L was designed using acrylic material. Here separate experimentations were carried out for *C. globosa* and *A. obliquus* and the results concerning their efficiency are depicted in Chapter III.

# Chapter I Screening of Microalgae for CO<sub>2</sub> Assimilation Efficiency

#### Introduction

Unprecedented greenhouse gas emissions due to human activities have contributed substantially to global warming and climate change. The greenhouse gases mainly include Carbon dioxide (CO<sub>2</sub>), Methane (CH<sub>4</sub>) and oxides of Nitrogen (NOx). Of these,  $CO_2$  is of major concern today, owing to its higher concentration in the atmosphere (Ramanathan, 1988; Cheng et al., 2013). Industrial revolution of the 20<sup>th</sup> century has pumped huge volumes of carbon dioxide into the atmosphere and the magnitude of such emissions have grown exponentially from 280ppm before industrialization to 367ppm in 1999, 379 ppm in 2005 and to the current levels of 403.64 ppm (2017), as reported by Mauna Loa Observatory, Hawaii, USA. According to Cape Grim Baseline Air Pollution Station, a joint venture of the Bureau of Meteorology and the Commonwealth Scientific and Industrial Research Organization (CSIRO), the levels of atmospheric carbon dioxide has noticed to be 403.26 ppm in 2017. The situation will worsen that by the end of the 21st century, we could expect to see carbon dioxide concentrations of anywhere from 490 to 1260 ppm, 75-350% above the pre-industrial concentration (Ebi et al., 2003). The release of carbon dioxide into the atmosphere is far more rapidly than it is being removed, and this imbalance causes increased carbon dioxide concentrations in the atmosphere. This unbalanced emission of atmospheric carbon dioxide constitutes a major challenge to global sustainability.

Both natural and anthropogenic sources contribute to elevated CO<sub>2</sub> emissions the atmosphere. The natural sources include ocean release, into decomposition of organic matters, respiration process, forest fires, emissions pertaining to volcanic eruption etc. Anthropogenic releases are said to be due the dependency on fossil fuels to meet energy needs in various sectors (Chang and Yang, 2003; Hansen et al., 2008). Other reasons include deforestation, unscientific agricultural practices, unsustainable land use changes etc. First World Climate Conference by UNEP and WMO (1979) observed anthropogenic activities as contributing factors to increased carbon dioxide concentrations in the atmosphere. The summit by WMO, UNEP and the International Council of Scientific Unions (ICSU) in Austria (1985) also accepted the reality of anthropogenic global warming due to accelerated release of carbon dioxide. The Kyoto Protocol (1997) and Paris Agreement (2016) also called for a reduction in GHG emissions and boosted climate change discussions.

The ill effects of increased levels of carbon dioxide in the atmosphere and its resultant complications are many. Higher emission may result in increased concentration of  $CO_2$  in our atmosphere, where it remains for 100 to 200 years, leading to a raise in temperature. The temperature increase has been unequivocally proven and is occurring at an unprecedented rate. As per studies of NASA on Earth's global surface temperature, year 2017 has been declared as the second warmest year since 1880. It is further stated that atmospheric temperatures could rise by 1.1 to 6.4 °C (2.0 to 11.5 °F) during the 21<sup>st</sup> century and sea levels would probably rise by 18 to 59 cm (IPCC, 2007).

There are global concerns on the release of GHGs by various nations. International initiatives are in progress over decades to check  $CO_2$  emissions into the atmosphere. In 2016, the largest  $CO_2$  emitting countries/regions

showed reduced emission trends, notably in China (-0.3%), United States (-2.0%), Russian Federation (-2.1%), Brazil (-6.1%) and the United Kingdom (-6.4%). Conversely an increased emission trend was noticed in India (+4.7%) and some of the developing countries (Olivier et al., 2015). This is pointing to the urgent measures, which are to be undertaken by the nation to control its greenhouse gas emissions. If the current emission trend continues, it will results in probable adverse impacts including changes in the quantity, timing and distribution of rain; sea level rise; increased frequency and intensity of wildfires; floods; droughts; storms and disturbances pertaining to coastal, marine and other ecosystems; expanding deserts; heat waves; increasingly severe weather; loss of agricultural productivity etc.

Considering the ill effects of elevated levels of  $CO_2$  in the atmosphere, identification of technically, economically and eco-friendly feasible strategies to reduce  $CO_2$  emissions are of great urgency. Recently, many research and development efforts are on to mitigate  $CO_2$  emissions, worldwide, and the most recent alternative for managing carbon dioxide is carbon sequestration.

Carbon sequestration refers to all means that is natural or artificial, by which  $CO_2$  is either confiscated from the atmosphere or redirected from emission sources and stored in oceans, terrestrial environments and geologic formations. It is the process by which the atmospheric carbon is captured and converted to various forms that are unable to contribute to global warming (Nogia et al., 2013). The goal of deliberate carbon sequestration is to reduce the net flux of  $CO_2$  emissions to the atmosphere by trapping/sequestering carbon dioxide in heterogeneous environments. This itself is the agenda of Kyoto Protocol, to reduce the greenhouse gases emissions by 18%, below 1990 level, by 37 European Community and industrialized countries, within the period of 2013 - 2020.

Various physical, chemical and biological methods have been applied to capture and sequester  $CO_2$  (Abu-Khader, 2006). Physical methods include capturing of  $CO_2$  using PSA (Pressure Swing Adsorption) or PTSA (Pressure and Temperature Swing Adsorption) method (Ohta, 1997). A "molecular basket" using nanoporous solid adsorbent was also used to collect  $CO_2$  in the condensed form. Polyethylenimine (PEI)-modified mesoporous molecular sieve of MCM-41 type (MCM-41-PEI) has also been tested as a  $CO_2$  adsorbent (Xu et al., 2002; Abu-Khader, 2006).

Chemical reaction-based methods include cyclic carbonation/de-carbonation reaction (Gupta and Fan, 2002) and gas-absorption process in which the separation of carbon dioxide from a gas mixture is effected by washing with aqueous amine solution (Resnik, et al., 2004). The chemical reaction-based CO2 sequestration includes processes like separation, transportation and sequestration, which are expensive. Both the technologies mentioned above are not only costly and energy-consuming, but also offers marginal mitigation benefits (Wang et al., 2008). These methods, also requires considerable space for storage, coupled with elevated costs for monitoring, maintenance and operation and create serious concerns regarding CO<sub>2</sub> leakage (Bilanovic et al., 2009). Although the capturing and transportation of carbon dioxide is a feasible strategy and are technically proven, further investigation pertaining to the reliability and safety in terms of long-term storage remains necessary (Bonenfant et al., 2003; Yeh et al., 2001). Thus it is compulsory to develop cost-effective, environment friendly, sustainable methodologies /alternatives to curb the soaring emission rate. Many research works have been, and are being conducted to address this crisis.

Biosequestration is the capture and storage of atmospheric carbon dioxide by biological processes. This can be by increased photosynthesis (through practices such as reforestation / preventing deforestation and genetic

engineering techniques); by enhanced soil carbon trapping in agriculture; or by the use of algal bio sequestration (through aquatic systems or by algal bioreactors) to mitigate the carbon dioxide emissions from coal, petroleum (oil) or natural gas-fired electricity generation. Biological CO<sub>2</sub> fixation through the production of biomass energy by the process of photosynthesis has attracted much attention as an alternative strategy (Ragauskas et al., 2006; Kondili and Kaldellis, 2007). Many researchers consider photosynthesis as the best near intermediate-term solution to the problem of carbon emissions (Bayless, 2001).

Among several strategies of biological sequestration, utilizing the photosynthetic efficiencies of microalgae has been receiving significant attention as it offers energy-saving and eco-friendly technology (Skjanes et al., 2007; Wang et al., 2008). Apart from the dual role in biofuel production by controlling the GHG emissions and sequestering carbon, the terrestrial plants are expected to contribute 3-6% reduction in  $CO_2$  emissions (Ho et al., 2011; Kao et al., 2014). However, the capabilities of photosynthetic microalgae to fix  $CO_2$  are 10–50 times more efficient than terrestrial plants (Cheng et al., 2013; Lam et al., 2012). It was also proven that the algal species confining to fresh water and marine environments are responsible for 50% of total photosynthetic primary production (Giordano et al., 2005).

Thus microalgae are of particular interest to researchers seeking methods to mitigate carbon due to their rapid growth rates, compared to other energy crops (Dote, 1994; Minowa, 1995; Miao and Wu, 2006; Mutanda et al., 2011), potential for higher-efficiency solar conversion than terrestrial plants (Herzog et al., 1997), higher photosynthetic efficiency, higher biomass production,  $CO_2$  fixation ability, production of biodiesel and other by-products through biomass process technologies, better ability to handle in extreme environments that favour them for an easy incorporation into

engineered systems, shorter growth cycle and weight doubling time of about three to five days etc. (Chen et al., 2011). According to Chisti (2007), during every 24 hours, algal species double their biomass; but during the exponential phase, the doubling time can be as short as three-and-a-half hours. Apart from these, microalgae seem to be the most available eco-friendly alternative for all finite oil and food resources. Approximately one kilogram of algal dry cell utilizes around 1.83 kg of  $CO_2$  (Kumar et al., 2011), which in other words imply that per kilogram microalgae could capture nearly 1.83 kg  $CO_2$  (Brennan and Owende, 2010).

In this scenario, the strategy of employing potential microalgal members seems to be more economically competitive and environment friendly, compared to other short-term carbon capture and storage technologies. However, most of the reports on the responses of microalgae to higher dosages of CO<sub>2</sub> under laboratory conditions confines to a limited number of species like *Chlorella, Scenedesmus* and *Spirulina* etc. Several other members like *Botryococcus, Chlorobotrys, Chlamydomonas, Chlorococcum, Chlorogleopsis, Dunaliella, Emiliania, Eudorina, Haematococcus, Oocystis, Nannochloropsis, Synechococcus, Thermosynechococcus* etc. have been employed in carbon dioxide sequestration and have reported to exhibit efficiencies in various capabilities.

In Kerala, heterogeneous aquatic systems including kole wetlands are known for its immense micro algal diversity. The unique climatic conditions, water quality and availability of adequate nutrients provide an appropriate habitat for the profound growth and multiplication of microalgae, of which the major fraction remains still unexplored. Hence an issue based or point source based carbon dioxide mitigation can be made possible through aquatic sequestration using natively isolated, potential microalgal members. The capabilities of such isolated members can be employed in other engineered systems too. The present study is an attempt to bring/ isolate indigenous freshwater microalgal members that can be employed in this direction.

The objectives of the present study are thus outlined as:

- Maintenance of pure cultures of microalgal species using standard methods and selection of microalgal species which are active under culture conditions.
- Monitoring the responses of selected micro algal species under varying dosages of carbon dioxide supply.

# **Review of literature**

The potential for efficient photosynthesis (Miao and Wu, 2006; Basu et al., 2013), rapid multiplication (Mutanda et al., 2011; Cheng et al., 2013), higher growth rates compared to energy crops (Dote, 1994; Minowa, 1995; Miao and Wu, 2006; Wang et al., 2008), direct capturing of atmospheric gases from point sources (Kadam, 1997; Packer 2009; Chiu et al., 2011; Lam et al., 2012), conversion of atmospheric gases to glucose for growth requirements (Kurano et al.,1995; Ho et al., 2011; Ho et al., 2014; Maity et al., 2014), synthesis of a wide range of biomolecules (Tang et al., 2011; Maity et al., 2014; Cheah et al., 2015), wide tolerance to extreme environments (Kurano et al., 1995; Demidov et al., 2000; Zeng et al., 2011; Zhao and Su, 2014) etc. make microalgae more interesting in their utilization for a wide range of purposes. Biosequestration using microalgae thus received much attention as an eco-friendly approach in recent times.

Microalgae were attempted to fix  $CO_2$  from different sources including atmosphere, industrial flue gases and soluble carbonates (NaHCO<sub>3</sub> / Na<sub>2</sub>CO<sub>3</sub>) (Wang et al., 2008; Vidyasankar et al., 2013). Previous studies have established the ability of microalgae employed in the management of flue gases emissions pertaining to municipal waste incinerators (Douskova et al., 2009), gas boilers (Doucha et al., 2005), simulated flue gases (Lee et al., 2000), industrial heater (Chae et al., 2006), coal-fired power plants (McGinn et al., 2011) etc. The mechanism of passing of flue gases directly through aqueous medium, in which the cultured microalgae seems to fix the  $CO_2$  is an efficient method in capturing process (Kadam, 1997). Moreover, for the growth and development of many microalgal species, the high purity carbon dioxide is not essential, which in turn reduces the pre-treatment cost (Olaizola et al., 2003), though it imposes extreme conditions to the microalgae employed.

The selection of suitable microalgal species for carbon dioxide mitigation has an important role in the efficiency and cost reduction of the selected bio mitigation mechanism (Brennan and Owende, 2010). Biomass production seems to be the most significant/ critical factor in the selection of appropriate microalgal species for the successful bioconversion of carbon dioxide (Cheng et al., 2006; Cheah et al., 2015). Most of the research works quantified carbon fixation in terms of biomass production (Chae et al., 2006).

In addition to tolerance to higher levels of  $CO_2$  species with higher assimilation ability, increased tolerance to toxic components, adaptability with respect to changing temperature, pH and nutrient conditions seems to be ideal characteristics of an adaptive species (Singh and Ahluwalia, 2013). Fulke et al. (2010) and Eloka-Eboka and Inambao (2017) reported increased carbon fixation ability and lipid production as promising candidates for sequestration. In the case of large scale process, the species that can utilize the  $CO_2$  present in flue gas and exhibit increased growth under the natural daynight cycle are preferred (Benemann, 1993; Stewart and Hessami, 2005).

The  $CO_2$  acclimation of microalgae to higher concentrations is a complex process comprising various adaptation techniques (Sergeenko et al., 2000;

Muradyan et al., 2004). Depending on the microalgal species and  $CO_2$  concentrations, presence of lag period was noticed in some tolerant species (Satoh et al., 2002) like *Chlamydomonas* (Baba et al., 2011) and *Botryococcus braunii* (Yoo et al., 2010). For the reduction of the lag phase during the carbon sequestration studies and improving the carbon fixation efficiencies under higher dosages, acclimatization of inoculums with  $CO_2$  lower dosages under same conditions were performed (Yun et al., 1997).

The microalgae also exhibit higher  $CO_2$  concentration tolerance by redistribution of certain cellular organelles and adjusting their structural anatomy (Papazia et al., 2008). In species of *Chlorella* and *Scenedesmus obliquus*, Miyachi et al. (1986) observed developed pyrenoid surrounded by thick starch granules, in low  $CO_2$  cells than the high  $CO_2$  cells. Similar observations were also reported by Tsuzuki et al. (1986) in *Dunaliella tertiolecta* and the chloroplast in low- $CO_2$  cells was located near the plasma membrane while that in high-  $CO_2$  cells was positioned in the inner region of the cells.

Structural malformations like decrease in the cell volume (Jian-Rong and Kun-shan, 2002), increase in vacuole size and disintegration of thylakoid membranes (Khairy et al., 2014), increase in volume (Pronina et al., 1993; Faria et al., 2012; Khairy et al., 2014) and vacuolization of the cytoplasm (Sasaki et al.,1999) with response to elevated CO<sub>2</sub> concentrations was also reported. Morphological alterations in *Scenedesmus* occur in response to stress was also reported by Hangata et al. (1992).

In the  $CO_2$  tolerant *Chlorococcum littorale*, as an adaptation technique, Demidov et al. (2000) noticed a rapid shift of the PSA, when subjected to elevated  $CO_2$  state. This transition of the PSA from state I to II was observed to enhance the cyclic electron transport above (PS) I and furthermore to maintain the pH homeostasis, the generation of additional ATP was mandatory. By using the additionally generated ATP via cyclic electron transport, the V-ATPase maintains the pH homeostasis (Miyachi et al., 2003).

The physiological responses of microalgae to elevated levels of  $CO_2$  may results in the accumulation several other value added by-products including lipids, carbohydrates, proteins, pigments, carotenoids and vitamins, which are active ingredients in pharmaceuticals, food additives, feed supplements (Ho et al., 2010; Wijffels and Barbosa, 2010; Kumar et al., 2011; Milledge, 2011; Šoštaric et al., 2012; Toledo-Cervantes et al., 2013) etc. The carotenoids like lutein, canthxanthin, astaxanthin and  $\beta$ -carotene are of commercial importance and mainly used as ingredients of fish / poultry feed to enhance the reddish color of fish and yellowish color of egg yolk (Lorenz and Cysewski, 2000). Due to the increasing demand of  $\beta$ -carotene as pro-vitamin A, it is used in multivitamin preparations and in healthy food formulations (Krinsky and Johnson, 2005; Spolaore et al., 2006). Their antioxidant activity can safeguard the algal cells from photooxidative damage (Siefermann-Harms, 1985), which in turn can inhibit carcinogenesis in other organisms (Kumar et al., 2014). The fatty acids like oleic acid has significant role in the reduction of cholesterol content and thereby checking cardio vascular diseases (Beyhan et al., 2011) and the lauric, palmitic, linoleic, oleic, stearic, and antibacterial myristic acids have their and antifungal properties (Agoramoorthy et al., 2007).

There are several reports regarding the tolerance of species and strains toward elevated  $CO_2$  concentrations. It has been noticed that the  $CO_2$  tolerance efficiency and optimum  $CO_2$  concentration of micro algae differ significantly (Singh and Ahluwalia, 2013; Toledo-Cervantes et al., 2013; Cheah et al., 2015) within and between species. There were some general conclusions regarding tolerance efficiency that the  $CO_2$  concentrations above 5% may impart toxic effects to microalgal growth (Ramanan et al., 2010; Zhao and Su,

2014). However several authors also claimed that  $CO_2$  concentrations above 5% can improve the growth of microalgal strains (Silva et al., 1984; Lee and Tay, 1991).

Among the promising candidates, *Chlorella* and *Scenedesmus* have been the most studied species to date. Scenedesmus was better able to tolerate very high CO<sub>2</sub> concentrations than *Chlorella* (Hanagata, 1992). The species and strains of *Chlorella* employed in carbon sequestration includes *Chlorella* sp. (Chiu et al., 2008; Velea et al., 2009; Fluke et al., 2010; Devgoswami et al., 2011; Kao et al., 2012; Zhao et al., 2015; Suali et al., 2017), Chlorella sorokiniana (Kumar et al., 2014), Chlorella HA-l (Watanabe et al., 1992), Chlorella PY-ZU1 (Cheng et al., 2013), Chlorella sp. NTUH15 and Chlorella sp. NTUH25 (Chang and Yang, 2003), Chlorella vulgaris (Brown, 1996; Yun and Park, 1997; Keffer and Kleinheinz, 2002; Jeong et al., 2003; Chen et al., 2010; Anjos et al., 2013), Chlorella vulgaris ARC-1 (Chinnasamy et al., 2009), Chlorella vulgaris LEB-104 (Sydney et al., 2010), Chlorella kessleri (de Morais and Costa, 2007a,b), Chlorella pyrenoidosa (Yang and Gao, 2003), Chlorella pyrenoidosa SJTU-2 (Tang et al., 2011), Chlorella vulgaris BEIJ 1890 (Kastanek et al., 2010), Chlorella vulgaris LEB-104 (Sydney et al., 2010), Chlorella sp. UK001 (Hirata et al., 1996), Chlorella minutissima (Papazia et al., 2008) and Chlorella ZY-1 (Yue and Chen, 2005).

Similarly, the species and strains of *Scenedesmus* include general *Scenedesmus* sp. (Hanagata et al., 1992; De Morais and Costa, 2007a,b; Velea et al., 2009; Ho et al., 2010; Yoo et al., 2010; Devgoswami et al., 2011; Maraskolhe et al., 2012; Nayak et al., 2013), *Scenedesmus acuminatus* (Minillo et al., 2013) *Scenedesmus dimorphus* and *Scenedesmus quadricauda* (Jiang et al., 2013; Vidyashankar et al., 2013), *Scenedesmus obliquus* (Yang and Gao, 2003; Tang et al., 2011), *Scenedesmus obliquus* SA1 (Basu et al., 2013), *Scenedesmus obliquus* CNW-N (Ho et al., 2010), *Scenedesmus* 

*obliquus* WUST4 (Li et al., 2011), *Scenedesmus bajacalifornicus* BBKLP-07 (Patil and Kaliwal1, 2017) and *Scenedesmus* sp. ISTGA1 (Tripathi et al., 2 015).

Several attempts were also been carried out in this direction using other members like Acutodesmus sp. (Varshney et al., 2016), Ankistrodesmus (Salim, 2013), Botryococcus braunii (Sawayama et al., 1995; Yoo et al., Botrvococcus braunii 2010). SAG-30.81 (Sydney et al., 2010). Chlamydomonas sp. MGA 161 (Miura et al., 1993), Chlamydomonas reinhardtii (Yang and Gao, 2003; Packer, 2009), Chlorocuccum littorale (Kodama et al., 1993; Kurano et al., 1995; Ota et al., 2009), Chlorobotrys sp. (Velea et al., 2009), Chlorogleopsis sp. (Ono and Cuello, 2007), Cyanidium caldanum (Seckbach et al., 1970), Desmodemus sp. F2 (Ho et al., 2014), Dunaliella sp. (Eloka-Eboka and Inambao, 2017), Dunaliella tertiolecta (Nagase et al., 1998), Dunaliella tertiolecta SAD-13.86 (Sydney et al., 2010), Emiliania huxleyi (Takano and Matsunaga, 1995), Eudorina sp. K17 20 (Hanagata et al., 1992), Haematococcus (Devgoswami et al., 2011), Limnothrix redekei and Geitlerinema sulphureum (Manjre and Deodhar, 2013), Monoraphidium minutum (Zeiler et al., 1995), Nannochloropsis (Negoro et al., 1991; Chiu et al., 2009; Jiang et al., 2011), Nannochloropsis oculata (Hsueh et al., 2009), Oocystis sp. (Takeuchi et al., 1992), Spirulina sp. (De Morais and Costa, 2007b), Spirulina platensis (Watanabe and Hall, 1995), Spirulina platensis LEB-52 (Sydney et al., 2010), Synechococcus sp. (Miyairi, 1995) and *Tetraselmis* sp. (Matsumoto et al., 1995),

There are several challenges behind the idea of microalgal sequestration becoming practical. An important task in this direction is the identification of a species that can cope up with very high  $CO_2$  concentrations, as their growth rates are negatively influenced by increasing levels of  $CO_2$  (Lee and Lee, 2003). As  $CO_2$  supply contributes to the lowering of pH of the culture

medium, due to carbonic acid formation (Lam and Lee, 2011), the species that survive excellently in acidic environments are generally considered for  $CO_2$ sequestration studies. The strains suitable for  $CO_2$  sequestration must also have low risk of contamination and should produce high value products (Lopez et al., 2009). The selection of high-performance microalgal species with increased tolerance/adaptability to  $CO_2$ , rapid growth rate, high photosynthetic efficiency, higher lipid content etc. maximizes microalgal  $CO_2$ fixation and biomass production (Zhao and Su, 2014). In this light, the present study has been carried out with the objective of screening indigenous microalgal members with respect to their efficiencies for assimilating carbon dioxide.

#### **Materials and Methods**

In the present study, pure cultures of microalgae were prepared, maintained in the laboratory and screened for their efficiencies in assimilating external supply of carbon dioxide. Changes in growth pattern of the algae due to carbon dioxide supply were assessed through micrometry, culture turbidity, cell count and biomass. Similarly variations in gaseous components like Dissolved Oxygen (DO), free carbondioxide and other attributes of the culture media like pH, conductivity, resistivity, alkalinity and temperature were also monitored at regular intervals during the treatment period. The comparative efficiencies of algal members were worked out and the results are interpreted. The entire study has been consolidated into (1) collection (2) culturing and (3) screening of microalgae.

# **Collection of microalgae**

The present study has been carried out during March 2014 to December 2016. Microalgal samples were collected using plankton nets from heterogeneous fresh water environments of Malappuram, Palakkad and Thrissur districts of Kerala state. The collected organisms were identified using standard manuals, literatures (Ramos et al., 2012; Philipose, 1967; Prescott, 1951; Snow, 1902) and also with the aid of experts.

# Culturing of microalgae

The collected microalgal samples were subjected to sub culturing and pure cultures of various organisms were maintained in Bolds Basal medium (Bold, 1949; Bischoff and Bold, 1963). The constituents of Bolds Basal medium is given below:

	400 mL	1 Liter	Added quantity	Molar
Components	stock	Stock	per liter of	<b>Concentration in</b>
	solution	Solution	medium	final medium
	-	Major Stock	Solutions	
NaNO	10 g L <sup>-1</sup>	25.00 g L <sup>-1</sup>	10 ml	$2.04 \times 10^{-3}$ M
IndinO <sub>3</sub>	dH <sub>2</sub> O	dH <sub>2</sub> O	10 111	2.94 X 10 IVI
$C_{2}C_{1} \bullet 2HO$	1 g L <sup>-1</sup>	2.50 g L <sup>-1</sup>	10 ml	$1.70 \times 10^{-4} M$
	dH <sub>2</sub> O	dH <sub>2</sub> O	10 111	1./0 x 10 101
$M_{0}SO. \bullet 7H_{0}O$	3 g L <sup>-1</sup>	$7.50 \text{ g L}^{-1}$	10 ml	$3.04 \times 10^{-4} M$
Wig504 - 71120	dH <sub>2</sub> O	dH <sub>2</sub> O	10 111	J.04 X 10 IVI
K <sub>2</sub> HPO	3 g L-1	7.50 g L <sup>-1</sup>	10 ml	$4.31 \times 10^{-4} M$
<b>R</b> <sub>2</sub> <b>III O</b> <sub>4</sub>	dH <sub>2</sub> O	dH <sub>2</sub> O	10 111	4.51 X 10 WI
KH2PO4	7 g L-1	17.50 g L <sup>-1</sup>	10 ml	1 29 x 10 <sup>-3</sup> M
11121 04	dH <sub>2</sub> O	dH <sub>2</sub> O	10 111	1.29 X 10 101
NaCl	1 g L-1	2.50 g L <sup>-1</sup>	10 ml	$4.28 \times 10^{-4} M$
i tuei	dH <sub>2</sub> O	dH <sub>2</sub> O	10 111	1.20 / 10 / 11
Alkaline EDTA			added 1 ml of this	
Stock Solution			solution per liter	
Stock Solution			of medium	
EDTA anhydrous		50 g L <sup>-1</sup>		$4.28 \times 10^{-4} M$
22 111 4111 9 41 0 415		dH <sub>20</sub>		
КОН		31 g L <sup>-1</sup>		1 38 x 10 <sup>-3</sup> M
		dH <sub>2</sub> O		1.001110 1.1
			added 1 ml of this	
Acidified Iron			solution per liter	
Stock Solution		1 0 0 T - 1	of medium	
FeSO <sub>4</sub> • 7H <sub>2</sub> O		4.98 g L <sup>-1</sup>		4.48 x 10 <sup>-5</sup> M
		dH <sub>2</sub> O		
$H_2SO_4$		1.0 mL		
			added I ml of this	
Boron Stock			solution per liter	

Table 1.1. Constituents of BB medium

Components	400 mL	1 Liter Stock	Added quantity	Molar Concentration in
Components	solution	Solution	medium	final medium
Solution			of medium	
H <sub>3</sub> BO <sub>3</sub>		11.42 g L <sup>-1</sup> dH <sub>2</sub> O		4.62 x 10 <sup>-4</sup> M
Trace Metal Stock Solution			added 1 ml of this solution per liter of medium	
$ZnSO_4 \bullet 7H_2O$		8.82 g L <sup>-1</sup> dH <sub>2</sub> O		7.67 x 10 <sup>-5</sup> M
$MnCl_2 \bullet 4H_2O$		1.44 g L <sup>-1</sup> dH <sub>2</sub> O		1.82 x 10 <sup>-5</sup> M
MoO <sub>3</sub>		$0.71 \text{ g L}^{-1} \text{ dH}_2\text{O}$		1.23 x 10 <sup>-5</sup> M
$CuSO_4 \bullet 5H_2O$		$1.57 \text{ g } \text{L}^{-1}$ $\text{dH}_2\text{O}$		1.57 x 10 <sup>-5</sup> M
$Co(NO_3)_2 \bullet 6H_20$		0.49 g L <sup>-</sup> <sup>1</sup> dH <sub>2</sub> O		4.21 x 10 <sup>-6</sup> M

# **Preparation of Bolds Basal medium (BB medium)**

For the preparation of medium, 10 ml of the first six stock solutions together with 1 ml each of alkaline EDTA, acidified iron, boron and trace metals solutions were added and the resultant solution was made upto 1 litre with glass distilled water and autoclaved. The pH of the medium was maintained at 6.6.

The cultures were periodically monitored through an image analyzer (Leica DFC 295 and Olympus BX43) and their photographs were taken (Olympus DP27). The pure cultures were maintained in fresh water micro algal culture collection, maintained in the Department of Botany, University of Calicut. Viability of cultures was maintained through adequate supply of culture medium.

Of various microalgal members maintained in culture collection, Chlamydomonas grovei, Chlamydomonas globosa, Desmodesmus opliensis, Monoraphidium contortum and Acutodesmus obliquus were taken for further study. Their selection was based on their viability and rapid growth and multiplication under culture conditions. The photographs concerning the micro algal members are given in plate 1. The descriptions of microalgal members taken for study are given below:

# 1. Chlamydomonas grovei G.S.West

Division	-	Chlorophyta
Class	-	Chlorophyceae
Order	-	Chlamydomonadales
Family	-	Chlamydomonadaceae
Genus	-	Chlamydomonas
Species	-	grovei

## **General environment**

This is a freshwater species found in all forms of stagnant water such as tanks, ditches, damp soil and pools. It is also adapted to moist soil and habitats rich in salts of ammonium.

# Morphology

Motile unicellular algae. Cells almost spherical, 2.5-4.5  $\mu$ m long and 2.5-4  $\mu$ m broad, with a pair of anterior flagellum of 6-5-10  $\mu$ m long. Chloroplast parietal, cup shaped, hollowed out at the anterior end which does not contain pyrenoids and nucleus, but contains bands composed of a variable number of photosynthetic thylakoids, which are not organized into grana-like structures.

# Reproduction

Asexual reproduction is by longitudinal division of protoplast into two, four or eight protoplast and liberation occurs by a gelatinization of the parent cell wall. The sexual reproduction occurs under unfavourable conditions. *Chlamydomonas* may also form aplanospores and akinetes.

# 2. Chlamydomonas globosa J.W.Snow

Division	-	Chlorophyta
Class	-	Chlorophyceae
Order	-	Chlamydomonadales
Family	-	Chlamydomonadaceae
Genus	-	Chlamydomonas
Species	-	globosa

# **General environment**

This is a freshwater species seen in stagnant water bodies including tanks, puddles, pond and lakes.

#### Morphology

In the motile form, the cells are slightly ellipsoidal or spherical, 5.2 to 7.8  $\mu$ m in diameter, anterior beak absent. Presence of two flagella which are longer or slightly longer than the cells. Pyrenoid is enveloped by a thick layer of starch. Chloroplast is thickened at the posterior end and extends to the extreme anterior end of the protoplast. A single lively vacuole can be distinguished in the anterior end. Presence of numerous oil globules in the anterior section of the cells seen.

# Reproduction

Asexual reproduction takes place mainly by zoospores under favourable conditions. The unfavourable conditions like desiccation, scarcity of nitrogen etc. leads to sexual reproduction, which involves gametogenesis, syngamy, zygospore formation and germination.

# 3. Desmodesmus opoliensis (P.G.Richter) E. Hegewald

Division	-	Chlorophyta
Class	-	Chlorophyceae
Order	-	Sphaeropleales
Family	-	Scenedesmaceae
Genus	-	Desmodesmus
Species	-	opoliensis

# **General environment**

This is a freshwater species. The distribution is ubiquitous and mainly seen in the paddy fields, lakes, shallow still waters, ponds and tanks.

# Morphology

Coenobia with a single row of linearly arranged 2,4 or 8 cells. Cells 9-18.5  $\mu$ m long and 3-7 $\mu$ m broad, oblong-cylindrical with round ends. Cell wall smooth, without ridges. Poles of terminal cells with a long, more or less straight or curved spines. Inner cells oblique to straight with straight walls with short spines.

# Reproduction

Reproduction occurs by non-motile autospores. The parental protoplast undergoes longitudinal and transverse division to form non-motile autospores.

# 4. Monoraphidium contortum (Thuret) Komárková-Legnerová

Division	-	Chlorophyta
Class	-	Chlorophyceae
Order	-	Sphaeropleales
Family	-	Selenastraceae
Genus	-	Monoraphidium
Species	-	contortum

## **General environment**

*Monoraphidium* is usually found within the plankton groups of fresh water lakes and ponds and sometimes inhabits artificial ponds and even waterfalls.

# Morphology

Cells 7.5 - 12  $\mu$ m length and 1.5 - 2  $\mu$ m width, uninucleate, solitary, irregularly curved with long, narrow, spindle shaped, gradually tapered apices, pointed ends, sigmoid, undulate to helicoidally twisted. Single parietal chloroplast without pyrenoids. Presence of 2,4,8 autospores in each cells.

# Reproduction

Asexual reproduction occurs by autospores released by longitudinal or transverse rupturing of parental wall. Flagellated stages and sexual reproduction are unknown.

# 5. Acutodesmus obliquus (Turpin) Hegewald & Hanagata

Division	-	Chlorophyta
Class	-	Chlorophyceae
Order	-	Sphaeropleales
Family	-	Scenedesmaceae
Genus	-	Acutodesmus
Species	-	obliquus

# **General environment**

This is a freshwater species seen floating in sluggish streams and lakes.

# Morphology

Coenobium of 2,4 and 8 cells with linear or slightly alternating irregular arrangement. Cells 7 to 35  $\mu$ m length, 1.5 to 8  $\mu$ m in width, fusiformly arranged with long axes parallel, adjoined throughout the lateral walls with truncated tapering ends. Chlorophyll is distributed throughout the cytoplasm. Outer free walls concave; cell poles narrowed and directed away from the centre of the colony.

# Reproduction

Reproduction always takes place by the formation of autocolonies within the mother cell wall. The formation of zoospores or motile gametes was never reported in this species.

#### **Experimental layout of screening studies**

The screening of microalgal members with regard to their efficiencies in assimilating carbon dioxide at various bubbling ratios like 10, 15, 20, 25, 30, 40, 60, 80 and 100 bubbles/2 hours, were worked out. The general layout of the experiment is given in Plate 2.

For experimentation, each time, 12 litres of BB medium was prepared and to this, 4 litres of culture medium containing pure cultures of the test organism was added and kept for incubation for a period of 24 hours. 16 litres of the resultant microalgal culture medium was then transferred to 16 conical flasks, each with a capacity of one litre, after through mixing. These were then separated into three sets of five conical flasks each. The first set of five flasks was treated as control set, and was maintained as such. To the second set containing 5 flasks, ambient air was supplied in the form of bubbles at regular intervals of two hours and this was treated as aerated set. To the third set containing five conical flasks, carbon dioxide from a cylinder was bubbled at a regular interval of two hours and was considered as  $CO_2$  treated set.

The culture contained in  $16^{th}$  conical flask was treated as the initial control set and was used to work out all parameters concerning the initial day of treatment. All the three sets were kept at illumination during the daytime from 6 am to 6pm. The experimentation was started at 6 am on the initial day. The control set was kept idle and to the other sets (aerated and CO<sub>2</sub> treated) air and CO<sub>2</sub> has been supplied at specific flow rates from 6am to 6 pm at an interval of 2 hours. Sampling and analyzing of cultures for pH, conductivity, resistivity, micrometry (cell size), turbidity, cell count, biomass content, dissolved Oxygen, free carbon dioxide and alkalinity were carried out at 6 am, every day. Monitoring of the cultures was carried out for 5 days (120 hours).
The significance of various parameters studied and methods of estimation are given below.

# pН

Estimation of the pH of culture media gives information regarding the metabolic status of organisms contained within it. It is also significant to have adequate information pertaining to the impact of various gaseous influxes on to the culture media. In the present study, pH of the treatments sets was analyzed using a digital pH meter (Systronics, 6373).

# Conductivity

Conductivity is the measure of the ability of an aqueous solution to conduct electric current. This ability is directly related to the presence, total concentration, mobility and valence of dissolved ions present in the culture medium. The conductivity values of the treatments sets were estimated using a digital water quality analyzer (Eutech Cyberscan, PCD 650) and the results are reported in micro Siemens ( $\mu$ S).

#### Resistivity

Resistivity is the opposition of the solution to the flow of an electrical current over a distance, which is directly related to the amount of additives; usually in the form of dissolved salts, alkaline substances, chlorides, sulphates and carbonate compounds. Resistivity values of the treatments sets were recorded using digital water quality analyser (Eutech, Cyberscan - PCD 650) and represented in k $\Omega$ .

# **Dissolved oxygen (DO)**

Estimation of Dissolved Oxygen (DO) of culture medium gives valid information on the nature, type and metabolic state of organisms contained within it. In the present study, DO is estimated using Winkler's method as given below:

Reagents used:

- 1. Winkler A: 100g of KOH and 50g of KI dissolved in 200ml of preboiled distilled water.
- Winkler B: 100g of MnSO<sub>4</sub>.4 H<sub>2</sub>O was dissolved in 200ml of distilled water. This has been heated to dissolve maximum salt; filtered after cooled.
- 3. Concentrated  $H_2SO_4$
- 4. Sodium thiosulphate: 0.025N
- Starch: 1.0g of starch powder was dissolved in 100ml distilled water. This was then heated for complete dissolution.

#### **Procedure**:

Each sample was taken in DO bottles without bubbling and added with 1.0 ml each of Winkler A and B solution, using separate pipettes. Extra care was taken to ensure that no air bubble is trapped while adding the reagents. After placing the stopper, the bottle was shaken and kept undisturbed for allowing the precipitate to settle down. Then the stopper was carefully removed and added with 1-2ml of concentrated  $H_2SO_4$ . The stopper was replaced and the bottle was shaken to dissolve the precipitate. 50ml sample from the DO bottle was transferred to conical flask for titration. The sample was then titrated against 0.025N Sodium thiosulphate, using starch as the indicator. The end point was recorded when initial dark brown / blue colour changes to colourless. The titration was repeated until concordant values were obtained.

DO was then calculated using the following equation and the results are reported in mg/L:

DO (mg/l) = 
$$\frac{\text{ml of sodium thiosulphate used x N of sodium thiosulphate x 8x 1000}}{\text{ml of sample used}}$$

# Free carbon dioxide

Free carbon dioxide in water accumulates due to microbial activity and respiration of organisms. Like DO, estimation of free carbon dioxide gives reliable estimate of the photosynthetic /respiratory status of organisms contained within it. In the present study, free carbon dioxide was estimated titrimetrically as follows.

Reagent used:

- 1. 0.05 N NaOH
- 2. Phenolphthalein reagent

#### **Procedure**:

A definite quantity of the sample was titrated with 0.05 N NaOH using Phenolphthalein as indicator. The end point was characterized by the appearance of a pink color. Free carbon dioxide was calculated from the following equation and the results are reported in mg/L:

Free carbondioxide(mg/l) =  $\frac{(ml x normality) of NaOH x 1000 x 44}{ml of sampleused}$ 

# Alkalinity

Alkalinity is the capacity to neutralize a strong acid and is characterized by the presence of all hydroxyl ions capable of combining with the hydrogen ion. Reagent used:

- 1. 0.01 N HCl
- 2. Phenolphthalein reagent
- 3. Methyl orange reagent

#### **Procedure**:

To 100 ml of the sample, drops of phenolphthalein were added. If the colour changed to pink, it was titrated with 0.1 N HCl until the colour disappeared at end point. This was kept as phenolphthalein alkalinity (PA). Now added 2-3 drops of methyl orange to the same sample and continued the titration, until the yellow colour changed to pink at the end point. The total alkalinity (TA) was then calculated from the following equation. The results are reported in mg/L:

$$PA mg/L = \frac{ml of HCl used with phenolphtdein x N of HCl x 1000 x 50}{ml of sampleused}$$
$$TA mg/L = \frac{ml of HCl used with phenolphtdein and methylorange x N of HCl x 1000 x 50}{ml of sampleused}$$

# Temperature

The measurement of temperature associated with the culture medium was important for assessing its effects on the biochemistry, rate of photosynthesis and biochemical reactions of micro algae. A mercury thermometer having a scale marked for every 0.1°C was taken and immersed in conical flask up to the level of mercury in the capillary column. The results are expressed in degree Celsius (°C).

#### **Micrometry (cell size)**

Micrometry involves the measurement of minute objects with a microscope using stage micrometer and ocular micrometer. In the present study, micrometry is used to assess the changes in cell size of micro algae in the culture media due to external gaseous inputs. In Micrometry, the stage micrometer, which is mounted on the stage, has linear scale of length 1 mm, which is precisely divided into 100 equal divisions. One small division is equal to 0.01mm. The ocular micrometer, which is placed on the top of the field diaphragm of eyepiece has 100 equal divisions. The magnification of the microscope is calibrated by aligning both micrometers parallel to each other. Stage micrometer is positioned so that its initial division coincides with initial divisions of ocular micrometer. The number of divisions on both scales was counted up to the next point, where the divisions of two scales again coincide. The size of organisms was estimated by the following equation and the results are reported in  $\mu$ m.

Value of one ocular micrometer =  $\frac{\text{Number of divisions on the stage micrometer}}{\text{Number of divisions on the ocular micrometer}} \times 10$ 

#### Turbidity

Turbidity is the measure of the amount of suspended particles in the sample solution. Assessment of the turbidity gives information about the density of the microalgal culture medium. Turbidity readings were taken using Nephelometer (EI-341) using distilled water as blank and the results are represented in NTU. Calibration of the instrument was done using standards prepared using Hydrazine sulphate and Hexa methylene tetramine.

# Cell count

Estimation of cell count provides information regarding the growth status of algal cells in culture medium. The cell count of algal members in the present

study was carried out using a Haemocytometer. For estimation of cell count, a drop of well agitated sample was placed on to the counting chamber and mounted using a cover glass. Using a compound microscope the cell count was taken and represented in cells x  $10^{4}$ .

#### Biomass

Algal biomass production is directly proportional to the efficiency with which the algal cells assimilate through photosynthesis. The biomass of microalgae was estimated by centrifuging 100ml of culture media containing sample in a pre -weighed centrifuge tube at 4000 rpm for 5 minutes. After centrifugation, the tube was blotted with a filter paper. After discarding the supernatant, it was kept undisturbed for 15 minutes for air drying. The weight of the tube with microalgae was then estimated. The weight of the microalgae was then calculated from the differences in weight of the centrifuge tube with microalgae and the weight of the centrifuge tube alone. The results are reported in g.

#### **Results and Discussion**

Currently, biological fixation of  $CO_2$  through photosynthesis has received much attention due to simultaneous production of biomass energy (Kondili and Kaldellis, 2007; Ragauskas et al., 2006). The rapid growth rate of microalgae (up to 10 times that of higher plants) and potential for higherefficiency solar conversion than land plants has accelerated their search for  $CO_2$  sequestration activities. The main challenge in this regard is the identification of ideal species that can thrive under extreme conditions, owing to extraneous supply of  $CO_2$ .

In the present study, an attempt has been carried out to assess the potentialities of selected microalgal species like *Chlamydomonas grovei*, *Chlamydomonas globosa*, *Desmodesmus opliensis*, *Monoraphidium contortum* and *Acutodesmus obliquus* belonging to Chlorophyceae, with regard to their efficiencies in carbon dioxide assimilation. Preliminary studies

were carried out on individual species for determining their optimum levels carbon dioxide assimilation.

Accordingly, 10, 15, 25 bubbles / 2 hours were supplied for *C. grovei* (Table 1.2) and 10, 15, 20, 40, 60 and 80 bubbles / 2 hours for *C. globosa* (Table 1.3). The potentials of *D. opoliensis* were monitored under 4 bubbling frequencies like 20, 25, 30 and 40 bubbles / 2 hours (Table 1.4) and the bubbling frequencies of 10, 15 and 20 bubbles / 2 hours (Table 1.5) were attempted for *M. contortum*. The efficiencies of *A. obliquus* were worked out under CO<sub>2</sub> supply of 20, 40, 60, 80 and 100 bubbles / 2 hours (Table 1.5).

Results pertaining to each treatment sets were analyzed and the final median values were worked out. Consolidated results concerning *C* .grovei, *C*. globosa, *D*. opliensis, *M*. contortum and *A*. obliquus with respect to control and various bubbling frequencies of air and  $CO_2$  are depicted in Tables 1.2 – 1.6, respectively. Statistical analyses of the data were carried out and the results are given in Tables 1.7 and 1.8.

An attempt has also been carried out to standardize the bubbling frequencies (10, 15, 20, 25, 30, 40, 60, 80, 100 bubbles / 2 hours) in accordance with the extent of  $CO_2$  dissolved per litre of Bolds Basal medium. This has been carried out by bubbling  $CO_2$  gas from the source at specific frequencies through the medium (1.0 litre) taken in a conical flask and closed with a cotton plug. The resultant free  $CO_2$ content associated with the medium was then analyzed titrimetrically using phenolphthalein and 0.05N NaOH and the results are represented in mg/L.

Accordingly, it has been noticed that a bubbling frequency of 10, 15, 20, 25, 30, 40, 60, 80 and 100 bubbles / 2 hours in a litre of Bolds Basal medium retained 66.0, 72.6, 79.2, 85.8, 92.4, 105.6, 132.0, 158.4 and 184.8 mg/L of free CO2 content, respectively, within BB medium.

Table1.2. Consolidated results on the responses of *Chlamydomonas grovei* to various dosages of CO<sub>2</sub> (Results are represented as median values of respective treatment sets).

No.         FARAMETERS AVALUSED         10         15         25           1         pH	SI.	DADAMETEDS ANALVSED	CO <sub>2</sub> supply in bubbles / 2 hours of interval				
1         pH           Control $6.74$ $6.74$ $6.74$ $6.72$ $6.77$ CO2 treated $6.74$ $6.72$ $6.77$ $6.81$ 2         CONDUCTIVITY( $\mu$ s)	No.	I ARAMETERS ANALTSED	10	15	25		
Control $6.74$ $6.74$ $6.7$ CO2 treated $6.75$ $6.74$ $6.77$ CO2 treated $6.75$ $6.74$ $6.81$ CONDUCTIVITY( $\mu$ s)	1	рН					
Aerated $6.74$ $6.72$ $6.77$ CO <sub>2</sub> treated $6.75$ $6.74$ $6.81$ 2         CONDUCTIVITY(µs) $$		Control	6.74	6.74	6.8		
CO2 treated $6.75$ $6.74$ $6.81$ C ONDUCTIVITY( $\mu$ s)		Aerated	6.74	6.72	6.77		
2         CONDUCTIVITY( $\mu$ s)           Control         764         774.9         754.7           Aerated         762         773.9         755           COy treated         763         773.4         754.6           3         RESISTIVITY( $k\Omega$ )		CO <sub>2</sub> treated	6.75	6.74	6.81		
Control         764         774.9         754.7           Aerated         762         773.9         755           CO2 treated         763         773.4         754.6           3 RESISTIVIY(kΩ)	2	CONDUCTIVITY(µs)					
Aerated         762         773.9         755           CO2 treated         763         773.4         754.6           3         RESISTIVITY(kΩ)		Control	764	774.9	754.7		
CO <sub>2</sub> treated         763         773.4         754.6           3         RESISTIVITY(kΩ)		Aerated	762	773.9	755		
3         RESISTIVITY(kΩ)           Control         1.28         1.262         1.295           Aerated         1.283         1.264         1.295           CO2 treated         1.283         1.264         1.295           CO2 treated         1.283         1.264         1.295           CO1trol         36         31.5         33           Aerated         37.5         31.5         31.9           CO2 treated         36         31.5         31.9           CO2 treated         36         10.45         4.9           CO2 treated         4.55         10.5         3.4           Aerated         3.6         10.45         4.9           CO2 treated         4.55         10.45         4.65           6         CELL COUNT(× 10 <sup>4</sup> cells/ml)		CO <sub>2</sub> treated	763	773.4	754.6		
Control         1.28         1.262         1.295           Aerated         1.283         1.264         1.295           CO2 treated         1.283         1.264         1.295           CO2 treated         1.283         1.264         1.295           CO2 treated         36         31.5         33           Aerated         37.5         31.5         31.9           CO2 treated         36         31.5         31.9           CO2 treated         3.6         10.45         4.9           CO2 treated         3.6         10.45         4.9           CO2 treated         3.6         10.45         4.9           CO2 treated         4.55         10.45         4.9           CO2 treated         4.55         10.45         4.9           CO2 treated         9.67         5.75         7.5           7         BIOMASS (gm)	3	RESISTIVITY(kΩ)					
Aerated         1.283         1.264         1.295 $CO_2$ treated         1.283         1.264         1.296           4         MICROMETRY(µm) $$		Control	1.28	1.262	1.295		
CO2 treated         1.283         1.264         1.296           4         MICROMETRY( $\mu$ m)		Aerated	1.283	1.264	1.295		
4         MICROMETRY( $\mu$ m)           Control         36         31.5         33           Aerated         37.5         31.5         31.9           CO2 treated         36         31.5         31.9           CO2 treated         36         31.5         31.9           Control         4.25         10.5         3.4           Aerated         3.6         10.45         4.9           CO2 treated         4.55         10.45         4.65           6         CELL COUNT(× 10 <sup>4</sup> cells/ml)		CO <sub>2</sub> treated	1.283	1.264	1.296		
Control         36         31.5         33           Aerated         37.5         31.5         31.9           CO2 treated         36         31.5         31.9           CO2 treated         36         31.5         31.9           CO2 treated         36         10.5         3.4           Aerated         3.6         10.45         4.9           CO2 treated         4.55         10.45         4.9           CO2 treated         4.55         10.45         4.65           6         CELL COUNT(× 10 <sup>4</sup> cells/ml)	4	MICROMETRY(µm)					
Aerated $37.5$ $31.5$ $31.9$ CO2 treated       36 $31.5$ $31.9$ <b>5 TURBIDITY(NTU)</b>		Control	36	31.5	33		
CO2 treated         36         31.5         31.9           5         TURBIDITY(NTU)		Aerated	37.5	31.5	31.9		
5         TURBIDITY(NTU)           Control         4.25         10.5         3.4           Aerated         3.6         10.45         4.9           CO2 treated         4.55         10.45         4.65           6         CELL COUNT(× 10 <sup>4</sup> cells/ml)             Control         8.25         6.38         6.55           Aerated         6.91         4.38         8.1           CO2 treated         9.67         5.75         7.5           7         BIOMASS (gm)              Control         0.008         0.007         0.01           Aerated         0.011         0.008         0.017           Control         0.008         0.007         0.017           Aerated         0.011         0.008         0.018           CO2 treated         0.012         0.007         0.017           8         DISSOLVED OXYGEN(mg/L)         Standard         T         R.6           Control         7.7         8.2         6.7           9         FREE CO2(mg/L)              Control         55         52.8         68.2 <td></td> <td>CO<sub>2</sub> treated</td> <td>36</td> <td>31.5</td> <td>31.9</td>		CO <sub>2</sub> treated	36	31.5	31.9		
Control         4.25         10.5         3.4           Aerated         3.6         10.45         4.9 $CO_2$ treated         4.55         10.45         4.65           6         CELL COUNT(× 10 <sup>4</sup> cells/ml)         8.25         6.38         6.55           Aerated         6.91         4.38         8.1 $CO_2$ treated         9.67         5.75         7.5           7         BIOMASS (gm)	5	TURBIDITY(NTU)					
Aerated $3.6$ $10.45$ $4.9$ CO <sub>2</sub> treated $4.55$ $10.45$ $4.65$ 6         CELL COUNT(× 10 <sup>4</sup> cells/ml) $8.25$ $6.38$ $6.55$ Aerated $6.91$ $4.38$ $8.1$ CO <sub>2</sub> treated $9.67$ $5.75$ $7.5$ 7         BIOMASS (gm) $           Control         0.008 0.007 0.01           Aerated         0.011 0.008 0.007 0.01           Aerated         0.011 0.008 0.017 0.017           Aerated         0.012 0.007 0.017           MISSOLVED         0.012 0.007 0.017           BISSOLVED         0.012 0.007 0.017           Control         7.9 8.7 8.6           Aerated         7.7 8.2 6.7           9         FREE CO2(mg/L)         $		Control	4.25	10.5	3.4		
CO2 treated $4.55$ $10.45$ $4.65$ 6         CELL COUNT(× 10 <sup>4</sup> cells/ml) $$		Aerated	3.6	10.45	4.9		
6         CELL COUNT(× 10 <sup>4</sup> cells/ml)           Control $8.25$ $6.38$ $6.55$ Aerated $6.91$ $4.38$ $8.1$ CO <sub>2</sub> treated $9.67$ $5.75$ $7.5$ 7         BIOMASS (gm) $$		CO <sub>2</sub> treated	4.55	10.45	4.65		
Control $8.25$ $6.38$ $6.55$ Aerated $6.91$ $4.38$ $8.1$ CO <sub>2</sub> treated $9.67$ $5.75$ $7.5$ 7         BIOMASS (gm) $0.008$ $0.007$ $0.01$ Aerated $0.011$ $0.008$ $0.007$ $0.01$ Aerated $0.011$ $0.008$ $0.017$ $0.017$ B         DISSOLVED $0.012$ $0.007$ $0.017$ B         OXYGEN(mg/L) $7.9$ $8.7$ $8.6$ Aerated $7$ $7.8$ $7$ Control $7.7$ $8.2$ $6.7$ 9         FREE CO <sub>2</sub> (mg/L) $-7.7$ $8.2$ $6.7$ 9         FREE CO <sub>2</sub> (mg/L) $-7.7$ $8.2$ $6.7$ 9         FREE CO <sub>2</sub> (mg/L) $-7.7$ $8.2$ $6.6$ Aerated $55$ $52.8$ $52.8$ $68.2$ CO <sub>2</sub> treated $57.2$ $52.8$ $74.8$ 10         ALKALINITY(mg/L) $-$	6	CELL COUNT(× 10 <sup>4</sup> cells/ml)					
Aerated $6.91$ $4.38$ $8.1$ CO <sub>2</sub> treated $9.67$ $5.75$ $7.5$ 7         BIOMASS (gm) $ 0.008 0.007 0.01           Aerated         0.011 0.008 0.007 0.01           Aerated         0.011 0.008 0.007 0.018           CO2 treated         0.012 0.007 0.017           8         DISSOLVEDOXYGEN(mg/L)         0.012 0.007 0.017           6         Aerated         7 7.8 7           Control         7.9 8.7 8.6           Aerated         7 7.8 7           CO2 treated         7.7 8.2 6.7           9         FREE CO2(mg/L)         $		Control	8.25	6.38	6.55		
$CO_2$ treated         9.67         5.75         7.5           7         BIOMASS (gm)		Aerated	6.91	4.38	8.1		
7         BIOMASS (gm)           Control         0.008         0.007         0.01           Aerated         0.011         0.008         0.018           CO2 treated         0.012         0.007         0.017           8         DISSOLVED OXYGEN(mg/L) $            Control         7.9         8.7         8.6           Aerated         7         7.8         7           CO2 treated         7.7         8.2         6.7           9         FREE CO2(mg/L)        $		CO <sub>2</sub> treated	9.67	5.75	7.5		
$\begin{tabular}{ c c c c c c } \hline Control & 0.008 & 0.007 & 0.01 \\ \hline Aerated & 0.011 & 0.008 & 0.018 \\ \hline CO_2 treated & 0.012 & 0.007 & 0.017 \\ \hline \textbf{BSSOLVED} & & & & & & & & & & & & & & & & & & &$	7	BIOMASS (gm)					
Aerated $0.011$ $0.008$ $0.018$ CO2 treated $0.012$ $0.007$ $0.017$ BISSOLVED OXYGEN(mg/L) $0.012$ $0.007$ $0.017$ Control $7.9$ $8.7$ $8.6$ Aerated $7$ $7.8$ $7$ CO2 treated $7.7$ $8.2$ $6.7$ FREE CO2 (mg/L) $$		Control	0.008	0.007	0.01		
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		Aerated	0.011	0.008	0.018		
8         DISSOLVED OXYGEN(mg/L)           Control         7.9         8.7         8.6           Aerated         7         7.8         7           CO2 treated         7.7         8.2         6.7           9         FREE CO2(mg/L)         52.8         52.8         66           Aerated         55         52.8         68.2           CO2 treated         57.2         52.8         74.8           10         ALKALINITY(mg/L)         7         75         100           Control         75         75         100           Aerated         80         67.5         90           CO2 treated         75         72.5         80		CO <sub>2</sub> treated	0.012	0.007	0.017		
OXYGEN(mg/L)         7.9         8.7         8.6           Control         7.9         8.7         8.6           Aerated         7         7.8         7           CO2 treated         7.7         8.2         6.7           9         FREE CO2(mg/L) $$	0	DISSOLVED					
Control         7.9         8.7         8.6           Aerated         7         7.8         7 $CO_2$ treated         7.7         8.2         6.7           9         FREE CO <sub>2</sub> (mg/L) $ $	0	OXYGEN(mg/L)					
Aerated       7       7.8       7 $CO_2$ treated       7.7       8.2       6.7         9       FREE CO <sub>2</sub> (mg/L) $$		Control	7.9	8.7	8.6		
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		Aerated	7	7.8	7		
9         FREE CO <sub>2</sub> (mg/L)           Control $52.8$ $52.8$ $66$ Aerated $55$ $52.8$ $68.2$ CO <sub>2</sub> treated $57.2$ $52.8$ $74.8$ 10         ALKALINITY(mg/L)         V         V           Control $75$ $75$ $100$ Aerated $80$ $67.5$ $90$ CO <sub>2</sub> treated $75$ $72.5$ $80$		CO <sub>2</sub> treated	7.7	8.2	6.7		
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	9	FREE CO <sub>2</sub> (mg/L)					
Aerated         55         52.8         68.2 $CO_2$ treated         57.2         52.8         74.8           10         ALKALINITY(mg/L) $V$ $V$ Control         75         75         100           Aerated         80         67.5         90 $CO_2$ treated         75         72.5         80		Control	52.8	52.8	66		
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		Aerated	55	52.8	68.2		
10         ALKALINITY(mg/L)           Control         75         75         100           Aerated         80         67.5         90           CO <sub>2</sub> treated         75         72.5         80		CO <sub>2</sub> treated	57.2	52.8	74.8		
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	10	ALKALINITY(mg/L)					
Aerated         80         67.5         90           CO2 treated         75         72.5         80		Control	75	75	100		
CO2 treated         75         72.5         80		Aerated	80	67.5	90		
		CO <sub>2</sub> treated	75	72.5	80		

(Temperature of the medium during experimentation ranged from 29 – 32°C)

Table 1.3. Consolidated results on the responses of *Chlamydomonas* globosa to various dosages of  $CO_2$  (Results are represented as median values of respective treatment sets).

Sl.	PARAMETERS	CO <sub>2</sub> supply in bubbles / 2 hours of interval					
No.	ANALYSED	10	15	20	40	60	80
1	pH						
	Control	6.91	6.87	6.82	6.6	7.34	6.93
	Aerated	6.88	6.85	6.79	6.6	7.34	6.94
	CO <sub>2</sub> treated	6.86	6.84	6.83	5.71	5.73	5.62
2	CONDUCTIVITY(µs)						
	Control	784.5	779	753.6	821	946.6	870.7
	Aerated	782.6	777.65	753.3	821.3	945.7	870.1
	CO <sub>2</sub> treated	782.7	776.35	753.2	807.6	934.95	855.55
3	<b>RESISTIVITY(kΩ)</b>						
	Control	1.248	1.255	1.297	1.192	1.033	1.124
	Aerated	1.25	1.257	1.298	1.19	1.034	1.124
	CO <sub>2</sub> treated	1.249	1.26	1.298	1.21	1.047	1.143
4	MICROMETRY(µm)						
	Control	23.25	22	33	33.75	31.88	35.6
	Aerated	24.5	22	32.7	37.5	31.88	37.5
	CO <sub>2</sub> treated	23.25	27.5	34.5	37.5	33.75	37.5
5	TURBIDITY(NTU)				n.		
	Control	4.2	4.55	4.35	14.1	9.3	9.55
	Aerated	3.95	4.5	3.9	13.6	9.25	9.75
	CO <sub>2</sub> treated	4.2	4.5	4.15	15.25	9.55	9.65
(	CELL COUNT				1		
6	(× 10 <sup>4</sup> cells/ml)						
	Control	34	23.5	29.5	277.5	150.5	186.5
	Aerated	29	22.5	25	267	157	146
	CO <sub>2</sub> treated	34.5	28	33.75	302.5	179	109.5
7	BIOMASS(gm)		•	•		•	•
	Control	0.011	0.015	0.013	0.047	0.042	0.032
	Aerated	0.014	0.015	0.013	0.047	0.042	0.028
	CO <sub>2</sub> treated	0.018	0.018	0.019	0.052	0.05	0.028
0	DISSOLVED						
ð	OXYGEN(mg/L)						
	Control	9.2	8	8.8	11	10.4	10.3
	Aerated	8.5	8.2	7.7	10.5	10.3	9.9
	CO <sub>2</sub> treated	8.7	7	8.6	12.8	12.3	10.2
9	FREE CO <sub>2</sub> (mg/L)						
	Control	48.4	52.8	50.6	11	8.8	22
	Aerated	48.4	52.8	52.8	17.6	8.8	26.4
	CO <sub>2</sub> treated	48.4	55	50.6	63.8	88	118.8
10	ALKALINITY(mg/L)			•			
	Control	95	107.5	112.5	127.5	102.5	102.5
	Aerated	92.5	100	107.5	122.5	100	110
			i	1	1	1	

(Temperature of the medium during experimentation ranged from 29.5 – 31°C)

Table 1.4. Consolidated results on the responses of *Desmodesmus* opoliensis to various dosages of  $CO_2$  (Results are represented as median values of respective treatment sets).

CI No.	PARAMETERS ANALYSED	CO <sub>2</sub> supply in bubbles / 2 hours of interval					
SI. NO.		20	25	30	40		
1	рН		•		•		
	Control	6.76	7.29	6.61	6.68		
	Aerated	6.74	7.21	6.6	6.69		
	CO <sub>2</sub> treated	6.76	7.24	6.65	6.67		
2	CONDUCTIVITY(µs)						
	Control	762.3	805.8	775.4	777.5		
	Aerated	760.9	803.3	774.4	776.4		
	CO <sub>2</sub> treated	761.5	803.2	774.4	774.7		
3	RESISTIVITY(kΩ)						
	Control	1.284	1.213	1.26	1.258		
	Aerated	1.287	1.216	1.264	1.259		
	CO <sub>2</sub> treated	1.285	1.218	1.262	1.263		
4	MICROMETRY(µm)						
	Control	36	40.5	41.25	38.25		
	Aerated	38.3	33.75	41.25	33.75		
	CO <sub>2</sub> treated	40.5	36	39.88	36		
5	TURBIDITY(NTU)						
	Control	0.85	6.6	8.6	9.65		
	Aerated	0.75	6.05	8.4	9.6		
	CO <sub>2</sub> treated	0.9	7.65	8.9	10.35		
6	CELL COUNT(× 10 <sup>4</sup> cells/ml)		•		•		
	Control	36	37.5	23.63	33.5		
	Aerated	30	37	21.13	39		
	CO <sub>2</sub> treated	41	47.5	22.46	38.13		
7	BIOMASS (gm)			•			
	Control	0.029	0.039	0.029	0.036		
	Aerated	0.03	0.035	0.032	0.043		
	CO <sub>2</sub> treated	0.032	0.043	0.033	0.039		
8	DISSOLVED OXYGEN(mg/L)						
	Control	8.7	8.4	8.3	7.2		
	Aerated	8.3	7.3	7.2	7.8		
	CO <sub>2</sub> treated	8.9	9.6	8	7.4		
9	FREE $CO_2$ (mg/L)						
	Control	66	57.2	61.6	110		
	Aerated	63.8	70.4	59.4	101.2		
	CO <sub>2</sub> treated	63.8	79.2	59.4	103.4		
10	ALKALINITY(mg/L)						
	Control	80	100	65	77.5		
	Aerated	70	90	70	72.5		
	CO <sub>2</sub> treated	72.5	92.5	60	72.5		
				-	-		

(	(Temperature	of the	medium	during e	experiment	ation ra	inged from	- 29.5 -	- 34°C	2)

Table 1.5. Consolidated results on the responses of *Monoraphidium* contortum to various dosages of  $CO_2$  (Results are represented as median values of respective treatment sets).

Sl. No.	PARAMETERS ANALYSED	CO <sub>2</sub> supply in bubbles/2 hours of interv				
		10	15	20		
1	рН					
	Control	6.74	6.85	6.76		
	Aerated	6.75	6.86	6.77		
	CO <sub>2</sub> treated	6.76	6.81	6.78		
2	CONDUCTIVITY(µs)					
	Control	772.6	768.2	769.4		
	Aerated	772.4	766.6	770.3		
	CO <sub>2</sub> treated	772.5	767.2	770.6		
3	RESISTIVITY(kΩ)					
	Control	1.265	1.273	1.27		
	Aerated	1.267	1.276	1.27		
	CO <sub>2</sub> treated	1.266	1.274	1.27		
4	MICROMETRY(µm)			•		
	Control	2.25	72	32.63		
	Aerated	1.875	76.5	32.5		
	CO <sub>2</sub> treated	4	76.5	31.5		
5	TURBIDITY(NTU)					
	Control	7.65	8.45	3.65		
	Aerated	6.9	8.2	3.95		
	CO <sub>2</sub> treated	8	8.75	3.8		
6	CELL COUNT(× 10 <sup>4</sup> cells/ml)					
	Control	17	18.5	12		
	Aerated	14	17	14		
	CO <sub>2</sub> treated	21	21	13.5		
7	BIOMASS(gm)					
	Control	0.009	0.007	0.01		
	Aerated	0.008	0.008	0.013		
	CO <sub>2</sub> treated	0.01	0.01	0.012		
8	DISSOLVED OXYGEN(mg/L)		1			
	Control	8.2	7.3	8.1		
	Aerated	7.8	7	7.8		
	CO <sub>2</sub> treated	8.2	8.1	7.9		
9	FREE CO <sub>2</sub> (mg/L)		1			
	Control	52.8	52.8	57.2		
	Aerated	52.8	55	52.8		
	CO <sub>2</sub> treated	52.8	52.8	57.2		
10	ALKALINITY(mg/L)		I	I		
	Control	80	95	85		
	Aerated	80	87.5	80		
	CO <sub>2</sub> treated	75	82.5	80		

(Temperature of the medium during experimentation ranged from 30 – 31°C)

Table 1.6. Consolidated results on the responses of *Acutodesmus obliquus* to various dosages of  $CO_2$  (Results are represented as median values of respective treatment sets).

SI.	DADAMETEDS ANALVSED	CO <sub>2</sub>	hours of in	nterval		
No.		20	40	60	80	100
1	рН					
	Control	6.99	6.38	5.63	7.17	6.97
	Aerated	7.01	6.39	5.77	7.13	6.84
	CO <sub>2</sub> treated	7.02	6.43	5.49	5.82	5.04
2	CONDUCTIVITY(µs)					
	Control	784.6	766.65	747.75	834.85	850.5
	Aerated	783	766.7	745.2	834.75	845.6
	CO <sub>2</sub> treated	782.5	765.4	740.35	827.65	828.5
3	RESISTIVITY(kΩ)					
	Control	1.247	1.276	1.307	1.171	1.151
	Aerated	1.248	1.276	1.313	1.171	1.157
	CO <sub>2</sub> treated	1.251	1.277	1.321	1.181	1.18
4	MICROMETRY(µm)					
	Control	45	39	46.88	45	46.88
	Aerated	45	33	46.25	46.5	52.5
	CO <sub>2</sub> treated	45	34.5	49.38	50.25	48.75
5	TURBIDITY(NTU)					
	Control	3.95	8.25	11.8	9.05	7.95
	Aerated	4.05	8.35	13.3	9	8.65
	CO <sub>2</sub> treated	4.2	8.65	14.5	10.3	8.2
6	CELL COUNT (× 10 <sup>4</sup> cells/ml)					
	Control	24	28	53	25.5	24.5
	Aerated	25.13	28.63	59.5	27.5	19.25
	CO <sub>2</sub> treated	30.67	33.75	60.38	30.75	19.75
7.	BIOMASS (gm)					
	Control	0.043	0.032	0.042	0.04	0.043
	Aerated	0.041	0.031	0.053	0.047	0.04
	CO <sub>2</sub> treated	0.045	0.034	0.061	0.051	0.034
8	DISSOLVED OXYGEN (mg/L)					
	Control	9.4	9.2	8.5	10	10.5
	Aerated	9.9	9.1	10.3	10.1	9.4
	CO <sub>2</sub> treated	9.9	10.1	15	13.8	13.9
9	FREE CO <sub>2</sub> (mg/L)					
	Control	28.6	39.6	39.6	44	44
	Aerated	30.8	39.6	37.4	44	44
	CO <sub>2</sub> treated	33	35.2	48.4	41.8	129.8
10	ALKALINITY(mg/L)					
	Control	90	90	97.5	110	130
	Aerated	90	92.5	92.5	110	112.5
	CO <sub>2</sub> treated	92.5	97.5	95	100	130

(Temperature of the medium during experimentation ranged from 28 – 30°C)

For monitoring the carbon dioxide sequestration potentialities of each microalgal member, the results pertaining to major growth parameters like turbidity of the medium owing to microalgal growth, together with cell count and biomass content of micro algal members were considered.

Upon comparing the carbon dioxide assimilation efficiencies of five microalgal members, with *C. grovei*, higher turbidity (4.55 NTU), cell count (9.67 cells x 10<sup>4</sup> per ml) and biomass (0.012 gm) was noticed in the CO<sub>2</sub> treatment set of 10 bubbles / 2 hours, than control and aerated set. Comparing the growth parameters of *C. globosa*, increased turbidity (9.55 NTU), cell count (179 cells x 10<sup>4</sup> per ml) and biomass content (0.05 gm) was observed in CO<sub>2</sub> treatment set of 60 bubbles / 2 hours. With *D. opoliensis*, higher turbidity (0.9 NTU), cell count (41 cells x 10<sup>4</sup> per ml) and biomass (0.032 gm) was noticed in the CO<sub>2</sub> treatment set of 20 bubbles / 2 hours. Concerning *M. contortum*, maximum turbidity (8.75 NTU), cell count (21 cells x 10<sup>4</sup> per ml) and biomass (0.01 gm) was noticed in the CO<sub>2</sub> treatment set of 15 bubbles / 2 hours, whereas with *A. obliquus*, increased turbidity (10.3 NTU), cell count (30.75 cells x 10<sup>4</sup> per ml) and biomass content (0.051) was observed in CO<sub>2</sub> treatment set of 80 bubbles / 2 hours, compared to control and aerated set.

On an overall assessment of the growth performances of micro algal members under heterogeneous culture conditions, it has been noticed that in higher  $CO_2$ treatment sets of *A. obliquus* (bubbling frequencies of 60, 80 and 100 per 2 hours) and *C. globosa* (bubbling frequencies of 40, 60 and 80 per 2 hours), higher free  $CO_2$  content was noticed in the culture media, compared to control and aerated set.

pH is an important attribute which can affect both gas absorption and nutrient availability in aqueous systems (Vonshak, 1997). In the present study, upon comparison of the pH values pertaining to the treatments sets, it has been noticed that treatment sets of *A. obliquus* (60 bubbles/2 hours and 80 bubbles/2 hours) and *C. globosa* (40 bubbles/2 hours and 60 bubbles/2 hours) fed with intermittent  $CO_2$  supply exhibited pH in acidic ranges. It is being

stated that at higher  $CO_2$  concentration, the culture pH decreases due to the formation of high amount of bicarbonate buffer (Sobczuk et al., 2000).

Devgoswami et al. (2011) reported that when  $CO_2$  dissolve in water, depending on pH of the solution, three inorganic species can be formed. The carbonic acid formation or free  $CO_2$  molecules in pH below 4.5,  $HCO_3$ – at pH 8.5 and carbonate at pH above 8.5. Usually the  $HCO_3$  fails to diffuse across cell membrane and the carbonic anhydrase enzyme present in the cell surface, aid in conversion of  $HCO_3$  to  $CO_2$  and consequently transported to cells for assimilation process (Tsuzuki and Miyachi, 1989). The carbonic anhydrase facilitate microalgae in carbon concentrating mechanism (CCM) to satisfy the carbon requirements.

The decreasing pH due to the presence of higher CO<sub>2</sub> concentration seems to be lethal to algal species (Kumar et al., 2010). It is reported to be a limiting condition, influencing photosynthesis and inhibiting carbon dioxide biofixation (Cheng et al., 2006). In the present study, the microalgal members (A. obliquus and C. globosa) are surviving in acidic range of pH in treatment sets fed with CO<sub>2</sub> owing to their adaptabilities with respect to changing environmental conditions. The capabilities of micro algal members like Chlorella sp. and Chlorococcum littorale to tolerate acidic pH was also reported (Chang and Yang, 2003; Yue and Chen, 2005; Kodama et al., 1993; Zeng et al., 2011; Zhao and Su, 2014). The better growth performances of some of the micro algal members at acidic pHs and the maintenance of culture pH within a range are indicative of their adaptability and  $CO_2$ assimilation capabilities. The results pertaining to the research work of Olaizola et al. (2004) also indicate that the pH of the medium is a key determinant of CO<sub>2</sub> capture efficiency and the species that survive excellently in acidic environments are generally considered as more tolerant to higher CO<sub>2</sub> concentrations (Singh and Ahluwalia, 2013 and Eloka-Eboka and Inambao, 2017).

Also, upon comparing the conductivity values of the  $CO_2$  treatment sets pertaining to *A. obliquus* (60 bubbles/2 hour and 80 bubbles/2 hour) and *C.* 

*globosa* (40 bubbles/2 hour and 60 bubbles/2 hour) it was noticed to be lower than both control and aerated set. In the case of resistivity values of  $CO_2$  treatment sets, it was noticed to be higher than control and aerated set.

Responses of microalgal cells to external stimuli vary. Here in the treatment sets of *A. obliquus* (60 and 80 bubbles/2 hours) and *C. globosa* (15, 20, 40, 60 and 80 bubbles/2 hours) increased cell size was noticed in CO<sub>2</sub> treated set, compared to the cells of both control and aerated set. Hanagata et al. (1992) stated that the morphological modifications assist in imparting tolerance to elevated CO<sub>2</sub> levels. The increase in the cell size of microalgae pertaining to the CO<sub>2</sub> treated sets can be considered as the adaptation techniques with response to the elevated CO<sub>2</sub> dosages. Faria et al. (2012) noticed increased cell volumes of *Chlorella* sp. due to the accumulation of reserve substances, while cultured under CO<sub>2</sub>. An increase in size and number of vacuoles with elevated CO<sub>2</sub> levels has been reported by Pronina et al. (1993). The cell enlargement in *Chaetoceros gracilis* in response to elevated CO<sub>2</sub> concentrations was also reported (Khairy et al., 2014).

Increases in turbidity and cell count in most cases are considered to be an index of growth. In the present study, the results of turbidity and cell count was found to be increased in treatment sets of *A. obliquus* (60 and 80 bubbles/2 hour) and *C. globosa* (40 and 60 bubbles/2 hour) fed with intermittent supply of CO<sub>2</sub>. Hence it can be concluded that the intermittent supply of CO<sub>2</sub> on *A. obliquus* (80 bubbles/2 hour) and *C. globosa* (60 bubbles/2 hour) have accelerated the growth of both microalgal members.

According to literature, biomass production seems to be the most significant factor used for assessing the capabilities of microalgae in the successful bioconversion of carbon dioxide (Cheng et al., 2006; Cheah et al., 2015). Most of the research works quantified the carbon fixation in terms of biomass production (Chae et al., 2006). Here in the present study also the results pertaining to biomass content was considered mainly as the major growth parameter to ensure the optimum species and dosage in which maximum  $CO_2$ 

assimilation occurred. Results with *A. obliquus* (80 bubbles/2 hour) and *C. globosa* (60 bubbles/2 hour) are highly promising and indicating with higher growth potential, as evidenced by results of carbon dioxide treated set compared to control and aerated set.

On a comparative assessment of the DO of the culture medium, it was noticed that  $CO_2$  treated sets of *A. obliquus* (60 and 80 bubbles/2 hour) and *C. globosa* (40 and 60 bubbles / 2 hour) maintained higher levels of DO than their respective control and aerated set. The microalgae can produce oxygen as by product of photosynthesis with respect to simultaneous carbon dioxide uptake (Kumar et al., 2010; Suali et al., 2017).

Different algae exhibits varied responses to high oxygen concentrations (Pope, 1975; Salih, 2011). When  $O_2$  concentration increases, the  $CO_2$  concentration decreases successively, which in turn will hinder the photosynthetic process of microalgae and leads to the declined microalgal growth and  $CO_2$  fixation ability. Kumar et al. (2010) observed that DO concentration >35 mg/l inhibits the microalgal growth. Here in the present study, the DO recorded in the  $CO_2$  treated sets was lower than 35 mg/l and the increased DO content is an indicative of the active photosynthetic process of microalgae in the presence of available  $CO_2$ .

Alkalinity content of culture medium in the treatment sets of *A. obliquus* (60 and 80 bubbles/2 hour) was higher in the case of control and aerated set, while with *C. globosa* (40 and 60 bubbles/2 hour) increased alkalinity content in  $CO_2$  treated set was noticed.

The efficiency of  $CO_2$  removal and fixation mainly depends on the physiological conditions of microalgae, the potentiality for cell growth and ability for  $CO_2$  metabolism (Yoo et al., 2010). The optimum  $CO_2$  concentrations and the tolerance range of algae to higher carbon dioxide in the medium are species-specific (Seckbach et al., 1971; Kodama et al., 1993; Sergeenko et al., 2000). The  $CO_2$  concentration of 1–5% was reported as optimal for majority of photosynthetic microorganisms (Sergeenko et al.,

2000). Varshney et al. (2016) observed maximum growth rate for *Acutodesmus* sp. at 10 % CO<sub>2</sub> and no inhibitory effects were noticed in higher CO<sub>2</sub> concentrations. Yun et al. (2016) noticed increased growth rate of *Acutodesmus obliquus* KGE 30 at 14.1 % CO<sub>2</sub>. The maximum biomass production of *Acutodesmus obliquus* in 20 % CO<sub>2</sub> and higher tolerance to 50 % CO<sub>2</sub> concentration was also reported (Chen et al., 2016). Tang et al. (2011) observed increased growth of *Acutodesmus (Scenedesmus) obliquus* SJTU-3 under CO<sub>2</sub> concentrations of 5 to 30 % and Jiang et al. (2013) reported 10 % CO<sub>2</sub>as optimum for *Tetradesmus (Scenedesmus) dimorphus*. However there were no reports regarding the CO<sub>2</sub> fixation efficiencies of *C. globosa*.

In spite of slightly acidic pH and higher free  $CO_2$  in the culture medium, microalgal members like *A. obliquus* (80 bubbles/2 hour) and *C. globosa* (60 bubbles/2 hour) exhibited higher rate of DO production, higher turbidity, higher cell count and higher biomass production in treatment sets supplied with free  $CO_2$ , which characteristically indicates their efficiencies in mitigating gaseous  $CO_2$ . The species *C. grovei*, *M. contortum* and *D. opoliensis* also exhibited moderate efficiencies in carbon dioxide assimilation, but only next to *A. obliquus* and *C. globosa*.

# **Statistical Analysis**

The results pertaining to the optimum bubbling frequencies of *A. obliquus* (80 bubbles of  $CO_2$  per 2 hour) and *C. globosa* (60 bubbles of  $CO_2$  per 2 hour) were subjected to statistical analysis using One Way Anova method. Throughout the analysis, significant variation among different parameters, that are likely to influence the growth and survival of these microalgal members under 3 treatment sets were identified by considering the F value. The CD values were worked out for significant parameters to draw conclusions regarding the variations among 3 treatment sets and the results are reported in Tables 1.7 and Table 1.8.

Sl. No.	TREATMENT SETS	Average	F- value	CD
рН	1		I	
1	Control set	7.34	(115 (70*	
2	Aerated set	7.407	6415.678*	0.0582
3	Carbon dioxide treated set	5.323		
Conductiv	vity			
1	Control set	834.2	66726*	
2	Aerated set	832.8	00./30*	4.0696
3	Carbon dioxide treated set	818.9		
Resistivity	7			
1	Control set	1.173	17 4*	
2	Aerated set	1.174	1/.4*	3.4991
3	Carbon dioxide treated set	1.194		
Turbidity		·		
1	Control set	13.23333	11.02*	0.6182
2	Aerated set	13	11.03*	
3	Carbon dioxide treated set	14		
Cell Coun	t	<b>I</b>		
1	Control set	32	1.75	NS
2	Aerated set	28	1.75	
3	Carbon dioxide treated set	34		
Micromet	ry			
1	Control set	45	2 (01	NS
2	Aerated set	48.75	3.081	
3	Carbon dioxide treated set	52.5		
Dissolved	oxygen			
1	Control set	9.666	2 00	
2	Aerated set	9.6	3.88	NS
3	Carbon dioxide treated set	14		
Free carbo	on dioxide			
1	Control set	44	615*	
2	Aerated set	44	04.3**	9.9427
3	Carbon dioxide treated set	79.226		
Alklainity				
1	Control set	130	40*	
2	Aerated set	110	42*	9.2532
3	Carbon dioxide treated set	140		
Biomass				
1	Control set	0.043	171 2727*	
2	Aerated set	0.045	1/1.2/2/*	0.0043
3	Carbon dioxide treated set	0.069		

Table 1.7. Comparative study of various parameters of *Acutodesmus* obliquus with respect to 80 bubbles of CO<sub>2</sub> per 2 hour.

\* Significant at 5% level; NS: not significant (critical value of F at 5% level for 3 treatments and 3 replications =5.143)

Table 1.8. Comparative study of various parameters of *Chlamydomonas* globosa with respect to 60 bubbles of  $CO_2$  per 2 hour.

SI No	TREATMENT SETS	Average	F- value	CD
	рН			
1	Control set	7.77	5500 410*	
2	Aerated set	7.776	5598.412*	0.0621
3	Carbon dioxide treated set	5.716		
Conduct	ivity	·		
1	Control set	946.233		
2	Aerated set	946.2	554.656*	1.4898
3	Carbon dioxide treated set	930.733		
Resistivi	ty			
1	Control set	1.033	15.072*	
2	Aerated set	1.033	15.073*	3.1732
3	Carbon dioxide treated set	1.050		
Turbidit	ty .			
1	Control set	10.4	202.040*	
2	Aerated set	11.1	202.048*	0.4238
3	Carbon dioxide treated set	13.95		
Cell Cou	int	·		
1	Control set	146	127 120*	
2	Aerated set	155	137.139	13.1516
3	Carbon dioxide treated set	218		
Microm	etry			
1	Control set	37.5	2	NS
2	Aerated set	37.5	3	
3	Carbon dioxide treated set	33.75		
Dissolve	d oxygen			
1	Control set	10.4	1000*	
2	Aerated set	10.4	1089*	0.4524
3	Carbon dioxide treated set	17		
Free car	bon dioxide			
1	Control set	11	2772 60*	
2	Aerated set	11	2773.39*	4.0740
3	Carbon dioxide treated set	105.666		
Alklaini	ty			
1	Control set	105	12*	
2	Aerated set	100	13.	11.3327
3	Carbon dioxide treated set	120		
Biomass				
1	Control set	0.052	17 210*	
2	Aerated set	0.049	17.310	0.0064
3	Carbon dioxide treated set	0.062		

\* Significant at 5% level; NS: not significant (critical value of F at 5% level for 3 treatments =5.143)

In the case of *A. obliquus*, while comparing the results at 5% level, the mean pH values of the three sets showed significant variations when compared with the CD value = 0.0582 (F value = 6415.68, P value = 0.000). The mean pH values of carbon dioxide treated set of *C. globosa* exhibited significant variations between control set and aerated set and there were no significant variations between control set and aerated set while comparing the results with CD value 0.0621 (F value = 5598.412, P value = 0.000).

Upon comparing the results of the mean conductivity values of *A. obliquus* and *C. globosa*, the carbon dioxide treated set showed significant variations with both control set and aerated set with the CD value 4.0696 and 1.4898 (F value = 66.736,554.6555; P value = 0.000, 0.000), respectively. There were no significant variations between the control and aerated set in the case both microalgal species.

While analyzing the results of *A. obliquus* and *C. globosa* pertaining to mean resistivity values, it was noticed in that there were no significant variations between the three treatment sets, when compared to the CD values 3.4991 and 3.1732 (F value = 17.4, 15.07386; P value = 0.003, 0.004), respectively.

In the case of *A. obliquus*, the mean turbidity values of the carbon dioxide treated set showed significant variations with both control and aerated set with the CD value 0.6182 (F value = 11.03,P value = 0.009). However, there were no significant variations between control set and aerated set. Upon comparing the with CD value (0.4238) it was noticed that the *C. globosa* exhibited significant variations between three treatment sets (F value = 202.0476, P value = 0.000633).

Upon analyzing the F value (1.75) of the cell count pertaining to *A. obliquus*, it was observed that there were no significant variations among the three treatment sets and considered as non-influencing / non-significant parameter.

However the mean cell count results of *C. globosa* revealed the significant variations of carbon dioxide treated set with both control set and aerated set with the CD value 13.1516 (F value = 137.1386, P value = 0.000) and there were no significant variations between the control and aerated set.

Upon comparing the F value of the treatment sets of *A. obliquus* (3.681) and *C. globosa* (3), micrometry was noted to be anon-significant parameter.

Comparison of the mean values with F value (3.88) of *A. obliquus*, dissolved Oxygen content was considered as non-significant parameter. The mean values of Dissolved Oxygen content of carbon dioxide treated set of *C. globosa* showed significant variations when compared to both control and aerated set with CD value 0.4524 (F value = 1089 and P value = 0.000).

The mean free carbon dioxide values of carbon dioxide treated set in both treatment sets exhibited highly significant variations with control and aerated set for *both A. obliquus* (CD value = 9.9427, F value = 64.5 and P value = 0.00) and *C. globosa* (CD value = 4.0740, F value = 2773.59 and P value = 0.000). However there were no significant variations between control and aerated set.

While comparing the mean alkalinity values of *A. obliquus* (CD value=9.253, F value= 42 and P value=0.000), it was noticed that there were significant variations between control set, aerated set and carbon dioxide treated set. In the case of *C. globosa*, the carbon dioxide treated set exhibited significant variations with control set and aerated set (CD value = 9.2532,F value =13 and P value = 0.006) and there were no significant variations between control set.

Upon comparing the F value of *A. obliquus* (171.27) and *C. globosa* (17.31), the biomass content was considered as a significant parameter. Here the carbon dioxide treated set of both *A. obliquus* (CD value = 0.0043, F value =

171.2727 and P value = 0.000) and *C. globosa* (CD value = 0.0064, F value = 17.31081 and P value = 0.003) exhibited significant variations with both control and aerated set.

Upon consolidation of the statistical analysis pertaining to the effect of optimum bubbling ratios on the physico-chemical and biological parameters of both microalgal species, the parameters like pH, conductivity, resistivity, turbidity, free carbon dioxide, alkalinity and biomass were noticed to be significant. However apart from these parameters, the cell count and dissolved oxygen content were also noticed to be significant for *C. globosa*. From the above observations it can be also concluded that among the growth parameters of both species, turbidity and biomass was noted to be significant.

#### **Summary and Conclusion**

Among various strategies for  $CO_2$  sequestration, biological sequestration using photosynthetic microalgae have received considerable attention in recent times. Microalgae, one of the most important living resources of both fresh and marine systems can be employed for  $CO_2$  sequestration, as they have higher photosynthetic efficiency, higher biomass production and faster growth rate, compared to other energy crops. Also they can readily be incorporated into engineered systems.

In light of the above, the present study has been carried out with the objective of assessing the  $CO_2$  assimilation capabilities of five indigenous micro algal species belonging to Chlorophyceae, namely *Chlamydomonas grovei*, *Chlamydomonas globosa*, *Desmodesmus opliensis*, *Monoraphidium contortum* and *Acutodesmus obliquus*. These micro algal samples were collected from heterogeneous environments and their pure cultures were maintained in the laboratory using Bolds Basal medium. They were then individually subjected to  $CO_2$  assimilation studies.

For assimilation studies, 12 litres of Bolds Basal medium was prepared and to this, four litres of culture medium containing pure cultures of the respective microalgal species was added. After incubation, the microalgal culture was transferred to 16 one litre conical flasks and then separated into three sets of five conical flasks. The first set of five flasks were treated as control and was maintained as such. To the culture containing conical flasks of the second set, ambient air has been bubbled at an interval of two hours and treated as aerated set. To the third set, carbon dioxide from a cylinder has been bubbled at an interval of two hours and was considered as  $CO_2$  treated set. The culture contained in the last conical flask (16<sup>th</sup>) was treated as the initial control set and has been used to analyze all parameters meant for the initial day of treatment.

The experimentation was initiated at 6 am on the initial day and all the three sets were kept at illumination during the day time (6 am to 6 pm). Sampling and analyzing of culture has been carried out at 6 am of each day. pH, conductivity, resistivity, dissolved oxygen, free carbon dioxide and alkalinity content of the algal culture together with cell size (micrometry), turbidity, cell count and biomass content of the micro algal members were worked out. Monitoring of the cultures was carried out for a period of 120 hours.

For monitoring the carbon dioxide sequestration potentials of each microalgal member, the results pertaining to major growth parameters like turbidity of the medium owing to micro algal growth, together with cell count and biomass content of micro algal members were considered. While comparing the carbon dioxide assimilation efficiencies of five microalgal members under study, in the case of *C. grovei*, higher turbidity, cell count and biomass content was noticed in the CO<sub>2</sub> treatment set of 10 bubbles/2 hours. With *C. globosa*, increased turbidity, cell count and biomass content was observed in  $CO_2$  treatment set of 60 bubbles/2 hours. In the case of *D. opoliensis*, higher

turbidity, cell count and biomass was noticed in the  $CO_2$  treatment set of 20 bubbles/2 hours. Concerning *M. contortum*, maximum turbidity, cell count and biomass was noticed in the  $CO_2$  treatment set of 15 bubbles /2 hours, while with *A. obliquus*, increased turbidity, cell count and biomass content was observed in  $CO_2$  treatment set of 80 bubbles/2 hours.

Results of the present study indicated that all the microalgal members under study exhibited varied ranges of tolerance to  $CO_2$  supply. The results of *A*. *obliquus* (80 bubbles/2 hours) and *C. globosa* (60 bubbles/2 hours) are highly promising which exhibited, higher turbidity, cell count and biomass production in  $CO_2$  treatment sets. The species *C. gorvei* (10 bubbles/2 hours), *M. contortum* (15 bubbles/2 hour) and *D. opoliensis* (20 bubbles/2 hours) also exhibited moderate efficiencies in carbon dioxide assimilation.

There are several reports regarding the feasibility of utilizing microalgal species in  $CO_2$  assimilation like *Chlorella, Scenedesmus, Botryococcus, Euglena gracilis, Nannochloropsis oculata* etc. The present study also testifies the efficiency of five microalgal members (*C. grovei, C. globosa, D. opliensis, M. contortum* and *A. obliquus*) in  $CO_2$  assimilation. The response of microalgal members to various bubbling ratios were monitored through individual treatment sets. The comparison and consolidation of results revealed the possibilities of utilizing two highly promising microalgal species, *A. obliquus* and *C. globosa* in  $CO_2$  assimilation / sequestration studies, evidenced by higher turbidity, high cell count and high biomass in  $CO_2$  treatment sets.

# Chapter II

# pH specific modification of culture medium for growth maximization of *Chlamydomonas globosa* and *Acutodesmus obliquus*

# Introduction

Microalgae are extensively been used for varying purposes. Their basic features like diversity, abundance, minimal growth requirements, survivability in extreme environments, potential for the synthesis of biomolecules, biofuels, bioadsorption of heavy metals and pollutants, bio sequestration of greenhouse gases, generation of valuable byproducts, better ability to handle in engineered systems etc. are unique. Due to these captivating properties, researchers consider them as a promising tool for multi-orientated research purposes.

One among the major challenges in utilizing microalgae for specific purposes is their biomass production in required quantities in specific time intervals. Most of the microalgal members, which are abundant and virulent under natural habitats, seem to be slow growing under culture conditions, both in an out of laboratories. This can be due to lack of optimum climatic conditions, inadequate nutrient strength, lack of proper aeration and gaseous influx, insufficient light intensities and duration and other culture conditions.

Apart from this, the risk of developing pure cultures without contamination is another major constraint in their utilization for multidimensional purposes. Many mass multiplication techniques/protocols fail to maintain pure cultures of target species due to invading organisms. The major contaminants, apart from unwanted algal members include fungi, bacteria, yeast, molds and a wide category of zooplanktons, of which most of them are predators (Goldman and Ryther, 1976; Goldman, 1979, 1980).

Thus, one of the major problems faced in the utilization of microalgae as a raw material for various industrial purposes is the lack of cost effective and efficient mass multiplication techniques/protocols, which ensures contamination free population within a short span of time. It is quite disturbing to note that among 30,000 species of microalgae, only a limited number of them are maintained in laboratories and studied for their utilization for various purposes (Sharmila et al., 2014). Therefore, standardization of conditions offering mass multiplication of microalgal species in limited time interval is a challenge to the scientific community.

In recent times, several media have been formulated for microalgal culture, which includes BG-11 Medium, Bolds Basal medium, C medium, Chu-10medium, Combo medium, Forsberg medium 11, URO Medium etc. Later, most of them were modified for meeting required targets and few of them were developed by analyzing the water samples collected from their respective habitats, together with a detailed estimation on the nutrient requirements of respective microalgae. Still the problems associated with culture media preparation are persisting, which include faulty formulations, erroneous ionic concentration represented by pH, contamination, precipitate formation etc.

Considering these constraints, many research activities are undertaken to develop protocols for rapid multiplication of microalgal members. As the growth preferences of microalgal members vary in accordance with their species characteristics, a unified protocol for the mass multiplication of microalgae is inappropriate. Several studies proved that species-specific modification of culture conditions could contribute to the mass multiplication of microalgal members. Modification of the physical and chemical conditions of the microalgal cultures results in differences in cell composition, production of higher biomass and synthesis of bioactive metabolites. An experimental approach has been initiated in the present study with an objective to evaluate the responses and growth pattern of *Chlamydomonas globosa* and *Acutodesmus obliquus* in Bolds Basal medium at varying ranges of pH. This has been carried out as part of maximizing their biomass production by altering culture conditions and enabling them to be utilized for various commercial/industrial purposes. The specific objective outlined in this study includes:

• Determining the optimum pH favouring maximized growth of selected micro algal members like *Chlamydomonas globosa* and *Acutodesmus obliquus*, cultured in Bolds Basal medium.

#### **Review of Literature**

Research and developmental activities based on algal mass culturing initiated in Germany during 1940's (Harder and Witsch, 1942). Later on, many other countries including Netherlands (Wassink et al., 1953), Japan (Tamiya, 1957), France (Clement, 1975), India (Becker and Venkataraman, 1980), South Africa (Toerien and Grobbelaar, 1980) have initiated studies on micro algal culturing and their mass multiplication.

In large-scale cultivation and laboratory level culturing of microalgae, several environmental factors need to be considered. Some of them include pH, light and temperature (Khalil et al., 2010; Gong et al., 2014; Khatoon et al., 2014; Sukenik et al., 1993; Rocha et al., 2003; VanWagenen et al., 2012; Doan et al., 2011; Moazami et al., 2012; Bartley et al., 2013; Rai and Rajasekhar, 2014). These factors not only influence the growth, multiplication and photosynthetic activities, but also alter the carbon fixation and distribution of carbon into different types of macromolecules (Juneja et al., 2013). Through

optimization of these environmental factors, biomass maximization and subsequent usage can be achieved. However, the optimum and tolerance range of parameters tend to be species specific and may vary.

Among various factors, hydrogen ion concentration is an important factor for the optimal growth of microalgae (Khalil et al., 2010). It has an important role in influencing the microalgal growth, functioning of proteins and enzymes, nutrient absorption, carbon and trace metal utilization, lipid and antioxidant production etc. (Borowitzka and Borowitzka, 1988; Chen and Durbin 1994; Pelizer et al., 2002; Sakarika and Kornaros, 2016; Razzak et al., 2015). The pH of the culture medium also influences the photosynthetic activity of microalgae. In high and low pH, the photosynthetic activity decreases (Bakuei et al., 2015) and the high pH ranges alters the nutrient absorption. The low pH also performs as the enzyme inhibitor during photosynthesis (Bitong et al., 2011). Hence, it is essential to maintain the pH of the culture medium at optimum ranges, as extreme pH conditions may lead to the disruption of cellular activity and collapsing of cultured organism.

As an important factor in carbon sequestration studies, the pH determines the solubility and availability of  $CO_2$  and other nutrients (Juneja et al., 2013; Razzak et al., 2015; Varshney et al., 2016). While initiating carbon dioxide sequestration experimentations, the carbon dioxide discharge to culture medium results in shifting of medium pH to acidic ranges, which negatively influences the growth (Lee and Lee, 2003) and only a few microalgal members can thrive in this extreme pH ranges (Singh and Ahluwalia, 2013; Eloka-Eboka and Inambao, 2017). As a result of growth and multiplication of the microalgal cells in acidic pH, the depletion of inorganic carbon content of culture medium occurs, which in turn leads to increase in pH values (Razzak et al., 2015). The higher pH ranges restricts the availability of carbon from  $CO_2$  and reduces the affinity of microalgae to free  $CO_2$  (Azov, 1982; Rotatore

and Colman, 1991; Chen and Durbin, 1994; Juneja, et al., 2013), which leads to algal growth inhibition (Azov, 1982; Chen and Durbin, 1994).

Certain factors like buffering capacity and composition of the culture medium, metabolic functions associated with microalgae etc. also influences the pH of the culture medium (Dineshkumar et al., 2017). Previous research works proved that the microalgal members like *Scenedesmus* (Nalewajko et al., 1997) and *Chlorella* (Khalil et al., 2010) have pH-buffering capacity, which differs among the species or genera.

Most of the previous studies were based on the manipulations of the initial pH (Watanabe et al., 1992; Yue and Chen, 2005). Several studies have been carried out to assess the effect of changing pH on the growth of microalgae (Moss, 1973; Dubinsky and Rotem, 1974; Lane and Burris, 1981; Tubea et al., 1981; Van der Westhuizen and Eloff, 1983; Gehl et al., 1987; Chen and Durbin, 1994; Mayo, 1997). On the other hand, most of the recent experimentations are attempted to increase the lipid content (Gardner et al., 2011; Sharma et al., 2012; Shah, 2014; Bartley et al., 2014; Khatoon et al., 2014; Mandotra et al., 2016) and enhancing antioxidant potentialities (Ismaeil et al., 2016; Guedes et al., 2011) of microalgal members.

Modification of the pH of the culture medium normally results in metabolic inhibition (Goldman, 1982) and disruption of cellular processes (Razzak et al., 2015), which leads to limited growth and ultimately death and collapse of culture. The acidic pH of the culture medium can alter the nutrient uptake (Gensemer et al., 1993) and induce metal toxicity. The higher alkaline pH range also limits the availability of carbon from carbon dioxide and results in suppression of growth (Chen and Durbin, 1994; Azov, 1982).

The decline in growth at pH greater than 9 was also reported by many (Taraldsvik and Myklestad 2000; Munir et al., 2015; Mandotra et al., 2016;

Al- Safaar et al., 2016). On the other hand, Garcia et al. (2000) observed reduced growth rate above pH 8 and Chen and Durbin (1994) noticed reduced growth above pH 8.8. Aggregation of *Chlorella vulgaris* in pH 9.5, and little growth on extreme acidic and alkaline conditionswas noticed by Sakarika and Kornaros (2016). Varshney et al. (2016) observed growth inhibition in *Acutodesmus* sp. above and below pH ranges 4 and 11 were the color of species changed from green to white within 48 hours ultimately resulted in cell death.

Many report variations in the pH of culture medium due to external stimulus. The common trend of shifting the medium pH to alkaline ranges was noticed during microalgal growth (Armstrong and Calder, 1978; Zang et al., 2014). Moheimani (2013) reported the increasing pH levels during day time. The photosynthesis seems to be the established / general conclusion regarding the increasing pH in culture medium (Verduin, 1964; Zhang et al., 2014). During daytime, utilization of carbon dioxide for photosynthesis increases the pH levels and the respiration process during nighttime decreases the pH levels (Bartley et al., 2014). The photosynthetic activity was also influenced by the pH of the culture medium that the photosynthetic rate reduces at extreme higher and lower pH ranges (Bakuei et al., 2015). Dubinsky and Rotem (1974) also noticed similar variations in the pH values of culture medium. The pH variation observed in the culture medium may be due to the nature of the metabolites secreted by microalgae with response to the nutrients of the culture medium (Al-Shatri et al., 2014).

Several researchers reported on the cellular level adaptations and mechanisms of microalgae to resist and prevail over the extreme pH conditions (Lane and Burris, 1981; Malis-Arad and MC Gowan, 1982 a, b; Guckert and Cooksey, 1990; Visviki and Santikul, 2000). The ultra structural monitoring of the

microalgal cells provides information about the structural changes and adaptation techniques to cope with extreme pH ranges.

The acid tolerant species and acidophilic species can maintain neutral cystosolic pH for broad ranges of external pH (Lane and Burris, 1981; Gerloff-Eliaset al., 2005). However, the maintenance of neutral cystosolic pH was an energy consuming mechanism that the energy required to eliminate and neutralize the hydrogen ions, otherwise used for cell growth results in decreased photosynthetic rates (Nixdorf et al., 2003; Kamjunke et al., 2005). In spite of the external pH, growth of microalgae canbe achieved through optimizing the intracellular pH to 7.5 (Razzak et al., 2015).

Raven (1981) stated that the pH difference in the culture medium also affect the metabolic activities and membrane transport. The acidophilic algae possess an inner positive electric potential difference across plasmalemma, which reduces the proton influx. To survive in low pH, it is essential to decrease these proton influx and increase the efficiency of proton pump (Gross, 2000).

As an adaptation technique, increased membrane fluidity by altering the fatty acid composition was observed in *Euglena mutabilis* while exposed to acidic environment (Halter et al., 2012). Jiang et al. (2012) proposed the method of acid adaptation to enhance the acid resistance of microalgae exposed to the extreme acidic conditions. However the acid tolerance of microalgae mainly depends on the extent of adaptive period and the pH to which microalgae subjected for adaption (Davis et al., 1996).

There were only a few reports regarding the morphological changes of microalgae subjected to extreme environments like highly acidic and alkaline conditions. Visviki and Santikul (2000) conducted ultra-structure monitoring to reveal the adaptation techniques of *Chlamydomonas applanata* to extreme

acidic environments. On exposure of *Chalmydomonas* cells to acidic pH (4.4), cells with larger pyrenoids, decreased volume of starch granules and palmelloid colonies were observed. In pH 3.4, authors observed clumping and sticky palemelloid colonies with excessive mucilage production, abnormal cell division, loss of mobility and death of both single cells and colonies. Moreover, the existing single cells were 30 % larger than control cells with thicker cell walls. Authors also reported that the cell wall thickening act as an appropriate barrier for restricting the passive hydrogen ion uptake and thereby ensure the survival of microalgae treated in acidic stress.

In the alkaline pH ranges, the flexibility of the mother cell wall increases consecutively, which prevents the rupturing and autospore release and thereby extends the time for cell cycle completion (Malis-Arad and MC Gowan, 1982 a,b) Moreover, the cell division reduces due to the stress attributed by the culture medium and one stage of the cell cycle will be more sensitive than other stages (Guckert and Cooksey, 1990). Guckert and Cooksey (1990) while conducting experiments using *Chlorella* CHORI observed autosporangial stage as more sensitive stage. They also stated that the longer cell cycle results in the lower cell densities in extreme pH conditions. Gardner et al. (2011) also reported the presence of larger cells with irregular conformations in higher alkaline cultures especially in pH 10.3.

The adaptation of a microalgal species to specific environment plays a supreme role in its yield (Sakarika and Kornaros, 2016). Any differences or modifications in the pH values can cause changes in the quantity and composition of pigments, protein, carbohydrate, glycerol and secondary carotenoids or may cause cell death (Liu and Lee, 2000; Khalil et al., 2010; Khatoon et al., 2014).

Huang and Su (2014) reported pH as a significant factor influencing the chlorophyll content. Abe et al. (1999) observed the inhibition of chlorophyll synthesis at extreme pH (nearly 3). Beklioglu and Moss (1995) stated that the increasing pH causes decreasing of chlorophyll-*a*. However, Poza-carrion et al. (2001) and Sharma et al. (2014) declared the significant increase of phycobiliproteins in cyanophycean members with respect to increasing pH. There were also reports regarding the enhanced chlorophyll production in alkaline pH ranges (Ismaiel et al., 2016; Richmond and Grobbelaar 1986).

Several researchers (Jimenez and Niell 1991; Celekli and Donmez, 2006; Liu and Lee, 2000) reported the significant improvement in accumulation and production of  $\beta$ -Carotene and secondary carotenoids in microalgae due to effect of pH. Abalde et al. (1991) reported that the cells subjected to the extreme pH conditions exhibit hindered cell division, which consecutively leads to carotenoid accumulation. Conversely, Del Campo et al. (2000) observed limited accumulation of carotenoids at higher and lower ranges of pH.

The pH of the microalgal culture medium has a supreme role in the accumulation of lipid that the key enzyme for the lipid synthesis (ACCase) is pH dependent (Thampy and Wakil, 1985; Mandotra et al., 2016). During the adverse conditions, the growth reduces and the deviated energy was stored in the form of lipids (Abu-Rezq et al., 1999). Several authors (Guckert and Cooksey, 1990; Moheimani, 2013; Khatoon et al., 2014; Rai et al., 2015; Mandotra et al., 2016) discussed the influence of pH on the enhanced lipid accumulation. However Bartley et al. (2014) disagree with the role of pH on the enhanced lipid accumulation and Rai et al. (2015) commented that the mechanism behind this accumulation of lipid was unknown.

The morphological observations performed by Guckert and Cooksey (1990) on *Chlorella* CHORI proved that the alkaline pH inhibits the autospore

release and results in reduced growth and the redirected energy was utilized in the biosynthesis of TAG. Similar observations made by Gardner et al. (2011) stated that the actual mechanism of TAG accumulation due to pH was unknown. According to Bartley et al. (2014), pH have no effect on the fattyacyl profile. On the other hand Mandotra et al. (2016) claimed that pH have marked effect on the FA profile of algae.

During acidic pH stress, as an adaptation technique to control the osmotic imbalance, *Dunaleilla acidophila* accumulated glycerols (Fuggi et al., 1988). While the *Pinnilaria braunii* and *Chlamydomonas* sp. accumulated triacylglycerides (Tatsuzawa et al., 1996). Tatsuzawa et al. (1996) reported the ability of microalgae to increase saturated fatty acid content under acidic conditions to inhibit elevated proton concentrations and to decrease membrane fluidity. Poerschmann et al. (2004) observed such a mechanism in *Chlamydomonas* sp.

pH also influences the biochemical configuration of micro algae. Guedes et al. (2011) conducted experiments to evaluate the individual and combined effects of pH and temperature on the growth rate and antioxidant production of *Scenedesmus obliquus* through a full factorial design. The results revealed that the *Scenedesmus obliquus* exhibited augmented antioxidant production under higher pH ranges. Ismaiel et al. (2016) studied the effect of pH on the activity and production of various antioxidants developed by *Spirulina platensis* and observed the production of several antioxidants in the extreme conditions to alleviate the harmful effects of reactive oxygen species.

The formation of precipitate in microalgal cultures maintained at extreme pH ranges were discussed by several authors (Becker, 1994; Visviki and Santikul, 2000; Sirisansaneeyakul et al., 2011; Dineshkumar et al., 2017). These precipitations also influence the microalagal cell cycle (Malis-Arad and MC Gowan, 1982 a,b).

As stated in previous session, the contamination of micro algal cultures can happen due to various reasons. While initiating microalgal mass culturing attempts, the lack of appropriate controlling mechanisms of contaminations due to other microorganisms appears to be a major problem (Das et al., 2011). The major contaminants includes fungus, virus, protozoa, golden algae, bacteria and zooplanktons (Moreno-Garrido and Can<sup>°</sup>avate, 2001; Shi et al., 2006; Zhou et al., 2011; Peng et al., 2015; Touloupakis et al., 2015; Ma et al., 2017). The cross contaminations due to other microalgae (Piazzi and Ceccherelli, 2002) was also a major setback.

Among various environmental factors influencing biomass production, adjusting the pH of the microalgal medium can restrict the contaminations due to invading organisms (Wang et al., 2013). Liu and Lu (1990) recommend shifting of pH values to 3 for reducing flagellate populations in algal culture. Maintaining the culture medium pH in acidic range (pH-3) for nearly 1-2 hours can reduce the contaminations occurred due to rotifers (Becker, 1994). Ma et al. (2017) also confirmed this approach of lowering pH values to reduce contamination.

In order to prevent the proliferation of invading microbes, the pH of cyanobacterial cultures can be increased to alkaline ranges (Mc Ginn et al., 2011). In spite of such extreme pH conditions, cyanobacteria can grow without altering the biochemical composition (Touloupakis et al., 2015). The ability of microalgae to develop in alkaline pH seems to enhance the interest in outdoor cultivation by reducing the risk of biological contaminations (Muthuraj et al., 2014). Considering the chance of growth inhibition of microalgae at extreme pH ranges, it is important to seek the optimum pH range, at which the proliferation of contaminants and maximization of specific microalgal species takes place.

Generally, algae may have different growth responses with respect to varying ranges of pH. Certain factors like metabolic effects of the microalgal cells or the chemical influences on the medium determine the pH tolerance limit (Azov, 1982). Even though many microalgal members grow near neutral pH ranges (7.0-7.4), the optimum pH at which maximum growth occurs may vary (Sakarika and Kornaros, 2016). Ho et al. (2011) and Dineshkumar et al. (2017) considered pH ranges 7-9 as optimum. There were also reports regarding the optimum pH as the initial culture pH at which microalgae adapted to grow (Visviki and Santikul, 2000 and Hansen, 2002). Celekli and Donmez (2006) and Gong et al. (2014) stated that initial pH has little impact on the growth of microalgae.

Most of the available reports regarding an augmentation in the growth of micro algae based on pH specific medium modification confines to their exploitation in the perspective of lipid production/biodiesel production through stress induction. While compiling the available literature, it was noticed that the responses of microalgal members to varying ranges of pH confines only to a limited microalgal species having industrial significance. Preliminary studies in our laboratory proved the efficiencies of Chlamydomonas globosa and Acutodesmus obliquus as excellent candidates for carbon sequestration. However, the production of biomass in sufficient quantities seems to be a significant factor in determining the selectivity of such species for specific purposes. There were no reports regarding the pH specific medium modification of C. globosa. Literature related to A. obliquus is scanty. In this light, the present work has been attempted with an objective of determining the optimum pH favouring the maximized growth of selected micro algal species like C. globosa and A. obliquus cultured in Bolds Basal medium.
#### **Materials and Methods**

As pH was noted to be an important factor influencing the growth of microalgal members, an attempt has been carried out to assess the optimum pH favouring the maximum growth and multiplication of *Acutodesmus obliquus* and *Chlamydomonas globosa*, under culture conditions. Such a standardization is appropriate as their biomass production in sufficient quantities is a pre-requisite for various studies, including carbon dioxide sequestration.

#### **Experimental design**

For the present study, pure cultures of *C. globosa* and *A. obliquus* maintained in BB medium was used. The preparation of BB medium was carried out as per standard methods mentioned in Chapter I.

Each algal member has been attempted separately. For experimentation, treatment sets were maintained with conical flasks of 100 ml capacity (21 nos), each containing 50 ml Bolds Basal medium. The pH of culture medium was adjusted from 3-12, with a gradation of 0.5 using 0.5 N NaOH and 0.05 N HCl. After adjusting to the required pH, 5 ml of pure cultures of the respective micro algal members were added to the respective conical flasks and the changes in pH, if any, was adjusted. The control set was maintained at pH 6.6, which represents the original pH of the culture medium. Detailed work plan is depicted in Plate 3.

Monitoring of the treatment sets were carried out for a period of 7 days. The culture conditions like pH, temperature, conductivity and resistivity and micro algal growth parameters like turbidity, cell count, cell size, and biomass were monitored throughout the treatment period. The methodology, as depicted in Chapter I was followed for monitoring the culture conditions and growth performances of micro algal members. The biomass estimation was carried

out on the initial and final days of treatment. All the sets were kept at illumination during day time (6 am – 6 pm) at a light intensity of 40  $\mu$ mol m<sup>2</sup> s<sup>1</sup>. Throughout experimentation, the temperature and humidity associated with culture medium of treatment sets were monitored. The temperature of the medium during experimentation ranged from 26 – 29.8°C while the humidity ranged from 48 - 54 %. Every day after observation, the altered pH was readjusted to their experimental condition. The entire study has been carried out during March to September, 2016. The results are reported.

#### **Results and Discussion**

In the present study, an attempt has been carried out to assess the growth performances of selected microalgal members (*C. globosa* and *A. obliquus*) in Bolds Basal medium under varying culture conditions altered by pH. Results of conductivity, resistivity, cell size, turbidity, cell count and biomass content of the micro algal members in response to different ranges of pH are given in tables 2.1(a) - 2.2(g). The results are represented as median values of three trials. Responses of microalgal members to varying ranges of pH on the first day (initial day) and seventh day (final day) are represented in plate 4. Similarly, changes in turbidity, cell count and biomass associated with the microalgal members under different ranges of pH are represented as Figures 1a - 3b in plate 5.

During the treatment period, at higher alkaline pH ranges (pH 9 and above), a precipitation was noticed. Several researchers (Becker, 1994; Visviki and Santikul, 2000; Sirisansaneeyakul et al., 2011 and Dineshkumar et al., 2017) reported precipitate formation in cultures on account of pH exceeding 9. Hence, in the present study, median values of major growth parameters above pH 9 were neglected.

The natural changes in pH accomplished by the treatment sets in response to the pre adjusted pH are worked out. In the case of *C. globosa* all the treatment sets with altered pH showed a tendency to move towards neutral range. The pH values increased in acidic ranges and decreased in alkaline ranges to attain a neutral range. Here in the control set, the pH values ranged from 6.82 on the first day to 7.78 on the seventh day. A similar trend was noticed with *A. obliquus* and the pH values associated with control ranged from 6.63 on the

Similar changes were also noticed by Zhang et al. (2014). There are various reasons attributed to this process. Verduin (1964) reported that algae have the ability to increase the pH of the culture media through photosynthesis. Similar reasons were also attributed by Dubinsky and Rotem (1974), while monitoring the changes in pH due to algae in relation to initial pH and temperature. The mechanisms like photosynthesis and respiration leads to day and night pH fluctuations (Bartley et al., 2014). Furthermore, Al-Shatri et al. (2014) stated that the variation in the pH of the culture medium was due to secretion of metabolites by microalgae in response to the nutrients present in the medium. Al-Qasmi et al. (2012) discussed the capabilities of algae to manipulate the growth and biochemical composition with respect to altered physiochemical conditions.

In general, algal species exhibit different ranges of tolerance to pH (Ying et al., 2014). However, the changes in the pH values indicate the adaptabilities of the micro algal members to survive in modified environments by altering the pH to optimum levels. The capacity to survive in extreme environments by acclimatizing the metabolism corresponding to changing environmental conditions makes microalgae more interesting candidates for research (Chiranjeevi and Mohan, 2016).

first day to 7.19 on the seventh day.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	6.82	6.97	7.01	7.51	7.56	7.61	7.78	7.51
рН-3	4.86	3.47	3.35	3.38	3.4	3.38	3.41	3.4
рН-3.5	5.19	4.23	4.04	3.97	3.88	4.04	3.88	4.04
pH-4	5.25	4.55	4.44	4.4	4.4	4.41	4.44	4.44
pH-4.5	5.4	4.83	4.72	4.7	4.73	4.68	4.63	4.72
рН-5	5.57	5.16	5.16	5.05	5.06	5.12	5.07	5.12
рН-5.5	5.87	5.52	5.55	5.54	5.54	5.53	5.54	5.54
рН-6	6.32	6.04	6.03	6.05	6.05	6.04	6.03	6.04
рН-6.5	6.76	6.56	6.57	6.58	6.57	6.58	6.55	6.57
pH-7	7.14	7.03	7.11	7.08	7.13	7.11	7.2	7.11
рН-7.5	7.37	7.37	7.41	7.48	7.54	7.54	7.52	7.48
рН-8	7.51	7.59	7.67	7.72	7.81	7.78	7.79	7.72
рН-8.5	7.62	7.71	7.84	7.91	8.04	8.16	7.99	7.91
рН-9	8.06	8.04	8.16	8.32	8.41	8.49	8.35	8.32
рН-9.5	8.3	8.51	8.66	8.87	9.01	9.05	8.85	8.85
pH-10	8.59	9.11	9.3	9.42	9.59	9.59	9.46	9.42
рН10.5	8.83	9.02	9.66	9.76	9.87	9.91	9.94	9.76
pH-11	9.1	10.23	10.34	10.3	10.43	10.35	10.41	10.34
рН-11.5	11.03	11.17	11.2	11.19	11.14	11.13	11.2	11.17
pH-12	11.87	11.85	11.94	11.92	11.9	11.93	11.93	11.92

 Table 2.1(a). Variation in pH noticed in cultures of Chlamydomonas globosa.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	6.63	6.64	6.65	6.68	6.79	6.95	7.19	6.68
рН-3	3.21	3.04	2.96	3.06	2.99	3.03	3.03	3.03
рН-3.5	4.35	3.6	3.49	3.54	3.47	3.58	3.48	3.54
pH-4	5.41	4.29	4.01	4.13	3.99	4.08	4.05	4.08
рН-4.5	5.68	4.99	4.75	4.71	4.62	4.66	4.66	4.71
рН-5	5.87	5.24	5.17	5.2	5.13	5.17	5.1	5.17
рН-5.5	6	5.67	5.63	5.54	5.58	5.6	5.54	5.6
рН-6	6.25	6.08	6.06	6.08	6.05	6.06	6.06	6.06
рН-6.5	6.54	6.58	6.56	6.58	6.5	6.53	6.55	6.55
pH-7	7.09	7.07	7.07	7.09	6.96	7	7.03	7.07
pH-7.5	7.49	7.5	7.55	7.58	7.49	7.55	7.5	7.5
рН-8	7.75	7.86	7.95	7.93	7.89	8.01	8	7.93
рН-8.5	7.89	8.01	8.14	8.23	8.09	8.25	8.28	8.14
рН-9	7.97	8.2	8.3	8.43	8.38	8.5	8.54	8.38
рН-9.5	8.2	8.48	8.61	8.71	8.72	8.96	8.97	8.71
pH-10	8.53	8.91	9.15	9.26	9.31	9.6	9.58	9.26
pH10.5	8.5	9.19	9.5	9.73	9.68	10.11	9.99	9.68
pH-11	9.11	9.87	10.18	10.28	10.19	10.5	10.45	10.19
pH-11.5	10.12	10.65	10.82	10.9	10.7	11.1	10.89	10.82
pH-12	11.59	11.7	11.81	11.69	11.54	11.79	11.81	11.7

 Table 2.2(a).Variation in pH noticed in cultures of Acutodesmus obliquus.

## Conductivity (µS)

In the case of *C. globosa* maximum conductivity was observed in pH 12 (8842  $\mu$ S) on the sixth day and minimum in pH 6 (745.1  $\mu$ S) on the first day. In control set, the conductivity ranged from 789.2  $\mu$ S on the first day to 851.6  $\mu$ S on the seventh day. Upon comparing the final median values the maximum conductivity was recorded in pH 12 (8373  $\mu$ S) and minimum in pH 6 (753.2  $\mu$ S).

Throughout the experimental period, the treatment set containing *A. obliquus* showed maximum conductivity in pH 12 (5552  $\mu$ S) on the sixth day and minimum in pH 5.5 (746.4  $\mu$ S) on the first day. The conductivity values of the control set ranged from 762.4  $\mu$ S on the first day to 831.9  $\mu$ S on the seventh day. While analyzing the final median values, the maximum conductivity was noticed in pH 12 (3880  $\mu$ S) and minimum in pH 6 (754.9  $\mu$ S).

The conductivity of a solution depends on the concentration of ions present in it. Generally, conductivity value of solution increases with increasing concentration of ions. Here in the present study, pH was maintained in varied ranges from 3-12 with a gradation of 0.5 using 0.5 N NaOH and 0.05 N HCl. Moreover, every day after observation, the altered pH was readjusted to their initial pH range. Therefore, to monitor the changes in the ionic concentration of culture medium pertaining to treatment sets, the conductivity values were recorded.

In all the treatment sets, conductivity values were noted to be higher towards higher alkaline ranges (pH 12) and lower towards near neutral ranges (pH 6 and 6.5). During experimentation, a gradual increase in conductivity was observed in all treatment sets, including control. From the initial day to final day, the conductivity increased more rapidly in the treatment sets than control, which may be due to the increased ionic influx owing to the addition of NaOH / HCl to adjust pH.

			[					
Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	789.2	791.7	824.3	830.5	853.2	855.5	851.6	830.5
рН-3	838.1	856.7	864.4	880.7	884.2	898.5	903.6	880.7
рН-3.5	781.2	791.2	802.5	811.5	815.1	822.3	828.3	811.5
pH-4	758.8	779.7	788.2	798.9	806.8	812.3	818.3	798.9
рН-4.5	754.5	763.8	767.4	771.6	778.1	779.8	788.3	771.6
рН-5	758.7	771.7	774.7	776.8	791.2	794.5	801.4	776.8
рН-5.5	748.8	753.7	755.5	758.1	763.6	765.3	767.5	758.1
рН-6	745.1	748.1	749.3	753.2	755.1	758.1	761.7	753.2
рН-6.5	806.1	809.3	808.3	812.9	815	817	820.3	812.9
pH-7	826.2	827.5	830.4	834.1	836	838.3	835.9	834.1
рН-7.5	851.8	865.7	867.3	873.8	873	879.8	870.9	870.9
рН-8	861	884.6	892	902.3	909.5	917.9	925.6	902.3
рН-8.5	851.8	875.9	892.2	918.7	937.6	972	985.6	918.7
рН-9	875.1	927.7	962.7	1005	1043	1085	1132	1005
рН-9.5	894.1	965.1	1051	1150	1240	1349	1435	1150
рН-10	954	1148	1304	1484	1709	1953	2119	1484
рН10.5	1016	1193	1546	1828	2125	2386	2683	1828
pH-11	1176	1593	1986	2293	2634	2975	3350	2293
рН-11.5	2133	2894	3409	3827	4160	4484	4962	3827
рН-12	6496	6948	8153	8493	8373	8842	8762	8373

Table 2.1(b). Variation in conductivity (µS) noticed in the cultures of *Chlamydomonas globosa*.

r	1	1	1	1	1	1		
<b>Experimental set</b>	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	762.4	764.4	764	769.7	804.9	826.2	831.9	769.7
рН-3	852.2	913.9	917.6	914.2	929.7	937.6	964.4	917.6
рН-3.5	764.9	818.1	828.4	823.1	832.8	829.6	834.1	828.4
рН-4	778.7	802.7	809.8	809.2	840.4	842.5	840.1	809.8
рН-4.5	753.3	760.3	778.8	784.7	829.3	835	837.1	784.7
рН-5	751.3	760.6	763.1	765.4	786.6	792.2	798	765.4
рН-5.5	746.4	751.7	752.3	811.5	814.1	819.5	821.8	811.5
рН-6	748.5	751.6	752.4	755	754.9	772.5	775.9	754.9
рН-6.5	753.2	760.5	765.1	770.4	773.4	776.1	775.9	770.4
рН-7	793.3	796.7	799.3	813.9	814.9	818	816.9	813.9
рН-7.5	831.3	832.3	833.9	835.5	835.4	840.8	842.2	835.4
рН-8	859	873.4	883.4	886	899.6	909.1	910.8	886
рН-8.5	872.4	905.5	923.9	959	972.5	981.6	991.1	959
рН-9	878.8	909.8	931.2	950.6	1033	1049	1066	950.6
рН-9.5	894.3	947.3	978.1	1016	1048	1078	1131	1016
pH-10	921.1	988.3	1064	1145	1247	1393	1495	1145
рН10.5	920	1062	1238	1398	1568	1853	2010	1398
pH-11	956.8	1220	1556	1791	2060	2560	2725	1791
рН-11.5	1163	1599	2023	2447	2721	3243	3410	2447
pH-12	2349	2931	3663	3880	4115	5552	5069	3880

Table 2.2(b). Variation in conductivity ( $\mu$ S) noticed in the cultures of *Acutodesmus obliquus*.

#### Resistivity ( $k\Omega$ )

In the treatment set containing *C. globosa*, the maximum resistivity was observed in pH 6 (1.312 k $\Omega$ ) on the first day and the minimum in pH 12 (0.104 k $\Omega$ ) on the seventh day. In the control set, resistivity values ranged from 1.239 k $\Omega$  on the first day to 1.188 k $\Omega$  from the seventh day. While comparing the final median values of the treatment set, maximum resistivity was noticed in pH 6 (1.298 k $\Omega$ ) and minimum in pH 12 (0.117 k $\Omega$ ).

In *A. obliquus*, maximum resistivity was observed in pH 5.5 and pH 6 (1.31 k $\Omega$ ) on the first day and the minimum in pH 12 (0.18 k $\Omega$ ) on the sixth day. In control set, the resistivity values ranged from 1.29 k $\Omega$  on the first day to 1.17 k $\Omega$  on the seventh day. Upon comparing the final median values, maximum resistivity was recorded in pH 6 (1.3 k $\Omega$ ) and minimum in pH 12 (0.25 k $\Omega$ ).

Resistivity is directly related to the amount of additives, mainly dissolved salts, alkali, chlorides, sulphates and carbonate compounds present in the medium. If the sample solution contains high concentration of ions, it exhibits low resistivity. In the present study, the respective pH ranges were maintained and readjusted daily during experimentation using 0.5 N NaOH and 0.05 N HCl. Hence, for monitoring the changes associated with ionic concentrations due to the addition of NaOH and HCl to the culture medium, the resistivity values of the treatment sets were registered.

Upon comparing the final median values, the resistivity pertaining to cultures under treatment was found to be high in neutral ranges and low at high alkaline ranges. The resistivity values of cultures were noted to be in agreement of conductivity in all treatments of both microalgae retained at varying pHs.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	1.239	1.235	1.186	1.162	1.146	1.144	1.188	1.186
рН-3	1.166	1.141	1.131	1.109	1.103	1.086	1.085	1.109
рН-3.5	1.251	1.236	1.218	1.205	1.199	1.189	1.18	1.205
рН-4	1.288	1.254	1.24	1.22	1.212	1.203	1.195	1.22
pH-4.5	1.296	1.28	1.274	1.267	1.259	1.253	1.24	1.267
рН-5	1.288	1.267	1.262	1.258	1.236	1.23	1.22	1.258
рН-5.5	1.305	1.297	1.294	1.289	1.283	1.277	1.274	1.289
рН-6	1.312	1.307	1.305	1.298	1.295	1.289	1.283	1.298
рН-6.5	1.212	1.208	1.209	1.2	1.199	1.197	1.194	1.2
рН-7	1.183	1.181	1.177	1.172	1.169	1.166	1.169	1.172
рН-7.5	1.148	1.129	1.127	1.119	1.12	1.111	1.122	1.122
рН-8	1.135	1.107	1.096	1.083	1.073	1.065	1.056	1.083
рН-8.5	1.148	1.116	1.096	1.064	1.043	1.006	0.99	1.064
рН-9	1.117	1.054	1.015	0.973	0.942	0.901	0.864	0.973
рН-9.5	1.096	1.013	0.93	0.848	0.788	0.726	0.681	0.848
рН-10	1.025	0.852	0.75	0.659	0.572	0.507	0.462	0.659
рН10.5	0.999	0.819	0.632	0.535	0.49	0.41	0.363	0.535
pH-11	0.831	0.614	0.493	0.427	0.371	0.329	0.29	0.427
рН-11.5	0.458	0.338	0.287	0.256	0.235	0.218	0.186	0.256
рН-12	0.151	0.141	0.12	0.115	0.117	0.111	0.104	0.117

Table 2.1(c). Variation in resistivity (k $\Omega$ ) noticed in the cultures of *Chlamydomonas globosa*.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	1.29	1.28	1.28	1.27	1.21	1.18	1.17	1.27
рН-3	1.14	1.07	1.06	1.07	1.05	1.05	1.02	1.06
рН-3.5	1.28	1.2	1.18	1.19	1.17	1.18	1.17	1.18
рН-4	1.26	1.22	1.21	1.21	1.16	1.16	1.16	1.21
рН-4.5	1.3	1.29	1.26	1.25	1.18	1.17	1.17	1.25
рН-5	1.3	1.29	1.28	1.28	1.24	1.24	1.23	1.28
рН-5.5	1.31	1.3	1.3	1.21	1.2	1.19	1.19	1.21
рН-6	1.31	1.3	1.3	1.3	1.29	1.27	1.26	1.3
рН-6.5	1.3	1.28	1.28	1.27	1.26	1.26	1.26	1.27
pH-7	1.23	1.23	1.23	1.2	1.2	1.19	1.2	1.2
рН-7.5	1.18	1.17	1.17	1.17	1.17	1.17	1.16	1.17
рН-8	1.14	1.12	1.11	1.1	1.08	1.07	1.07	1.1
рН-8.5	1.12	1.08	1.06	1.02	1.01	0.99	0.99	1.02
рН-9	1.11	1.07	1.05	1.03	0.95	0.93	0.91	1.03
рН-9.5	1.09	1.03	1	0.96	0.93	0.91	0.87	0.96
pH-10	1.06	0.99	0.92	0.86	0.78	0.7	0.65	0.86
pH10.5	1.07	0.92	0.79	0.7	0.62	0.53	0.49	0.7
pH-11	1.02	0.8	0.63	0.55	0.47	0.38	0.36	0.55
pH-11.5	0.84	0.61	0.48	0.4	0.36	0.31	0.29	0.4
pH-12	0.42	0.33	0.27	0.25	0.24	0.18	0.19	0.25

Table 2.2(c). Variation in resistivity (k $\Omega$ ) noticed in the cultures of *Acutodesmus obliquus*.

## **Turbidity (NTU)**

In the treatment set containing *C. globosa*, the maximum turbidity was noticed in pH 8 (21.2 NTU) on the seventh day and minimum was recorded in pH 3 (4.7 NTU) on the first day. In the control set, turbidity ranged from 5.5 NTU on the first day to 11 NTU on the seventh day. In the present experimentation, minimum turbidity was noticed in acidic pH ranges (3-4.5). While comparing the final median values, maximum turbidity was noticed in pH 9 (10.4 NTU) and the minimum in pH 4 and 5 (5.6 NTU).

In the case of *A. obliquus*, maximum turbidity was monitored in pH 9 (5.9 NTU) on the seventh day and minimum in pH 3.5, 4.5 and 5 (1.4 NTU) on the second day. The turbidity values of the control set ranged from 1.7 NTU on the first day to 3.9 NTU on the seventh day. Upon comparing the final median values of the treatment set, maximum turbidity was noticed in pH 9 (4.6 NTU) and minimum in pH 3 (2.1 NTU).

Assessment of turbidity helps in estimating the microalgal cell densities in culture medium. Here, while comparing the final median values, both species exhibited maximum turbidity in alkaline ranges and minimum in acidic ranges, which is indicative of their growth enhancement in alkaline culture conditions. However, during experimentation, in higher alkaline ranges, formation of precipitate was noticed. Hence, for drawing conclusions regarding the optimum pH range in which maximum turbidity occurred, pH ranges above 9 were neglected.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	5.5	5.2	7.1	7.3	9.5	11.9	11	7.3
рН-3	4.7	4.9	5.7	5.3	6	6.9	6.2	5.7
рН-3.5	4.9	4.9	5.8	5.5	6	6.9	6.3	5.8
pH-4	4.9	4.8	5.6	5.1	5.8	6.9	6	5.6
рН-4.5	4.9	4.9	5.8	5.5	5.9	6.8	6.8	5.8
рН-5	5	5	5.6	5.2	5.9	6.9	6.9	5.6
рН-5.5	5.1	5.3	6.1	5.4	6.3	7.5	7	6.1
рН-6	5.5	5.6	7	4.4	7.2	9.2	9.6	7
рН-6.5	5.2	6.9	7.3	7.9	9	10.3	11.1	7.9
pH-7	5.3	6.9	7.6	8.2	10.1	12.4	14.3	8.2
рН-7.5	5.3	7.2	8.4	8.4	10.4	13.4	20.1	8.4
pH-8	5.8	8.1	8.9	10	12	16.5	21.2	10
pH-8.5	6.2	7.9	9.1	10.2	11.7	16.3	20.6	10.2
рН-9	6.1	8.8	9.9	10.4	11.6	15.1	18.8	10.4
рН-9.5	6.6	8.9	10.2	10.6	11.8	14.8	19.2	10.6
pH-10	7.2	10	11.2	11.2	11.9	13.7	17.2	11.2
pH10.5	6.2	8.3	9.1	10.4	11.2	12.2	14.2	10.4
pH-11	7.2	9	8.4	9.3	10.4	12.4	13.2	9.3
рН-11.5	7	8.6	8.4	8.2	8.3	8.8	9	8.4
pH-12	6.1	7.3	6.5	6.5	6.6	7.1	7.1	6.6

Table 2.1(d). Variation in turbidity (NTU) noticed in the cultures of *Chlamydomonas globosa*.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	1.7	1.8	3.2	3.6	3.3	3.3	3.9	3.3
рН-3	1.7	1.9	2.2	2.9	2.1	2.2	2	2.1
рН-3.5	1.8	1.4	2.3	3	2.3	2.2	2	2.2
рН-4	1.8	1.8	2.6	3.2	2.7	3	2.2	2.6
рН-4.5	1.9	1.4	2.7	3.3	2.9	3.6	2.5	2.7
рН-5	1.9	1.4	2.8	3.5	3.1	3.5	2.8	2.8
рН-5.5	2	1.6	2.7	3.6	2.8	3.7	2.6	2.7
рН-6	2	1.7	2.5	3.6	3	3.9	3	3
рН-6.5	1.6	1.9	2.1	3.4	2.9	3.5	2.8	2.8
pH-7	1.9	1.6	2.5	2.6	3.4	3.5	2.5	2.5
рН-7.5	2	1.8	2.9	3	2.7	4.1	2.9	2.9
pH-8	2.2	1.7	3.1	4.5	3.6	4.1	3.5	3.5
рН-8.5	2.2	2.1	3.5	4.3	4.7	5.2	5.2	4.3
рН-9	2.2	2.1	3.8	5.3	4.9	5.9	4.6	4.6
рН-9.5	2.6	3	5	6.3	5	6	5	5
pH-10	3	3.3	5.4	5.6	5.6	6	5.3	5.4
pH10.5	2.8	2.8	4.3	6	5.1	5.9	5.4	5.1
pH-11	3	3.3	4.7	6.3	5.8	5.4	4.7	4.7
pH-11.5	4	3.5	5	5.4	4.3	5.5	3.7	4.3
pH-12	4.2	3.9	4.6	4.1	4.3	4.2	3.1	4.2

Table 2.2(d). Variation in turbidity (NTU) noticed in the cultures of Acutodesmus obliquus.

#### **Cell count**

In the treatment set containing *C. globosa,* the maximum cell count was observed in pH 8 (170.25 cells x  $10^4$  per ml) on seventh day and the minimum in pH 12 (19 cells x  $10^4$  per ml) on the second and seventh day of the treatment. During the treatment period, cell count in control set ranged from 49.25 cells x  $10^4$  per ml on the first day to 158.26 cells x  $10^4$  per ml on the seventh day. While comparing the final median values, the maximum cell count was noticed in pH 8 (83.5 cells x  $10^4$  per ml) and minimum in pH 3 (32 cells x  $10^4$  per ml). The optimum range at which the maximum cell count was noted is pH 8 (83.5).

In the case of *A. obliquus*, the maximum cell count was noticed in pH 5 (24.75 cells x 10<sup>4</sup> per ml) on the fifth day and minimum in pH 12 (2 cells x 10<sup>4</sup> per ml) on the seventh day. In control set, the cell count ranged from 11 cells x 10<sup>4</sup> per ml on the first day to 9 cells x 10<sup>4</sup> per ml on the seventh day. While comparing the final median values, maximum cell count was observed in pH 5 (20 cells x 10<sup>4</sup> per ml) and minimum in pH 3 (4 cells x 10<sup>4</sup> per ml). Here the cell count was significant in acidic ranges (pH 5 and pH 5.5) and the optimum pH in which the maximum cell count occurred was five.

The estimation of cell count at regular intervals provides information regarding the growth of microalgae under varying treatment conditions. In the present study, *C. globosa* exhibited maximum cell count in alkaline range (pH 8), whereas that of *A. obliquus* was in acidic range (pH 5). The cell count obtained by both the microalgal members under specific pH was higher than the control. Varshney et al. (2016) conducted studies on the effects of pH on the growth of *Acutodesmus* species and reported that the algae exhibited high growth rates over a wide range pH (5–10). There are no reports regarding the growth rate of *C. globosa* with respect to varying pH ranges.

Several researchers also noticed decline in the growth of microalgal species under higher alkaline ranges (Garcia et al., 2000; Taraldsvik and Myklestad, 2000; Munir et al., 2015; Al-Safaar et al., 2016; Mandotra et al., 2016). In the higher alkaline pH , the increased flexibility of the mother cells inhibits the rupturing of cells and cell release from autospore and thereby prevents or lags the cell cycle which results in declined growth (Malis-Arad and MC Gowan, 1982 a,b; Guckert and Cooksey 1990; Rai et al., 2015).

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	49.25	50.25	67.5	73.5	81.75	138.25	158.26	73.5
рН-3	31	32	25.75	38.25	32.61	33.25	28.2	32
рН-3.5	34.25	35.5	44.5	41.26	33	39.28	38.2	38.2
pH-4	35	39.46	37.25	37.25	36.45	36.45	42.56	37.25
рН-4.5	40	36.95	38.75	49.82	46.25	56.38	66.2	46.25
рН-5	34.75	39.5	38.25	38.5	49.25	55.23	68.2	39.5
рН-5.5	37.5	39.25	53.5	48.9	45	59	72.45	48.9
рН-6	36.5	40.25	60.75	65.6	63.25	79.25	82.56	63.25
рН-6.5	45	48	67	72	80.25	118.32	122.68	72
pH-7	49	54.82	67.5	80.75	97.38	124.62	146.3	80.75
рН-7.5	48.5	59.65	72	79.5	87	140.23	160.23	79.5
рН-8	49.75	60.28	67	83.5	92.64	142.03	170.25	83.5
рН-8.5	49.25	56.25	66.5	68.5	92	139.6	160	68.5
рН-9	39.75	51	67.75	62.75	95.41	139.25	150.23	67.75
рН-9.5	50.66	50.26	64	75.25	94.5	140	146.25	75.25
pH-10	45	46.5	56.5	49.75	68	135.69	135.62	56.5
pH10.5	48.75	49.25	49.5	66.39	60.25	118.25	120.36	60.25
pH-11	39	38	50.2	55	55	114.02	110.25	55
рН-11.5	37.25	40	31.25	34	35	88.72	99.56	37.25
pH-12	32	19	22.5	22	22	22	19	22

Table 2.1(e). Variation in cell count (cells x 10<sup>4</sup> per ml) noticed in the cultures of *Chlamydomonas globosa*.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	11	8.5	10	7	13.25	14.05	9	10
рН-3	5.75	4.25	3.25	4.75	3	2.75	4	4
рН-3.5	5.25	4	7.5	5.25	8.5	4	8.75	5.25
pH-4	10	10.25	8.25	7.5	7.5	10.75	9.75	9.75
рН-4.5	9.5	12	11.75	10.75	17	11.25	17.75	11.75
рН-5	10	15.25	20	18.25	24.75	24	23	20
рН-5.5	10	13	12.25	18.5	19.75	21.5	19.25	18.5
рН-6	9	10.25	10	9	11.25	14.75	13.25	10.25
рН-6.5	11.25	11.75	8.75	7.75	8.8	9	13.5	9
pH-7	8.75	10.25	8.5	6.75	17.5	9	15.5	9
рН-7.5	9	10.75	7.75	8	8	11.75	14	9
рН-8	8.75	10.25	11.25	10.75	8.5	13.5	15.25	10.75
рН-8.5	10	9.25	9.5	10.75	14.25	4	19.5	10
рН-9	9.5	11	9.75	10.75	11.75	12	19.25	11
рН-9.5	8.5	11.25	11.5	10.5	10.75	12.25	18.75	11.25
pH-10	7.25	9.75	10.5	9	10.75	8.5	10.75	9.75
рН10.5	8.5	6.5	7.5	6.25	11.75	9.25	7.5	7.5
pH-11	4.5	6.5	7.75	6	5.25	4.5	8.8	6
рН-11.5	6.75	5.8	6	4.5	5.75	3	3.75	5.75
pH-12	4.25	4.2	3.5	3.25	6.5	4.25	2	4.2

Table 2.2(e).Variation in cell count (cells x 10<sup>4</sup> per ml) noticed in the cultures of *Acutodesmus obliquus*.

#### Cell size

In the experiment using *C. globosa*, the maximum cell size was noticed in pH 5 (38.5  $\mu$ m) on the first day and minimum in pH 5, 6, 7.5, 9.5, 10.5, 11 and 12. In the control set, the cell size ranged from 24.75 on the first day to 27.45 on the seventh day. While comparing the final median values, the maximum cell size was noticed in acidic pH 4.5 (29.5  $\mu$ m) and minimum in higher alkaline pH 12 (22  $\mu$ m).

In the treatment sets of *A. obliquus* the maximum cell size was noticed in pH 5.5 (55  $\mu$ m) on the first day and minimum in pH 5 (22  $\mu$ m) on the seventh day. In the control set, the cell size remained without change during experimentation. Here in the treatment sets, the maximum and minimum cell size was observed in acidic range of pH. In the case of control set, there has a gradual decrease in cell size until fifth day and then an increase in the sixth day. While evaluating the final median values of treatment set, maximum cell size was monitored in pH 9 (46.5  $\mu$ m) and minimum in pH 12 (33  $\mu$ m).

Structural changes like shrinkage or enlargement in cell size owing to the stress imparted by the addition of 0.5 N NaOH and 0.05 N HCl has been monitored through micrometry. While comparing the final median values, *C. globosa* showed increased cell size in acidic range, whereas *A. obliquus* showed an increase in cell size in near neutral to alkaline range. Here both species exhibited decreased cell size in higher alkaline range (pH 12). Moreover, in the case of *A. obliquus*, increased cell size was noticed in the control set when compared to that of treatment sets.

The presence of microalgal cells with increased cell size in acidic ranges were already reported (Visviki and Santikul, 2000; Hargreaves and Whitton, 1976). During morphometric analysis of *Chlamydomonas applanata* in pH 3.4, Visviki and Santikul (2000) noticed single cells with thicker cell walls, which

were 30 % larger than control cells. Authors reported that during extreme acidic conditions, the cell volume increases due to the proliferation of the cell wall by the enhanced deposition of cell wall polysaccharides. This cell wall thickening can ensure the survival of microalgae in acidic conditions by acting as an appropriate barrier in restricting the passive hydrogen ion uptake. On the other hand, while conducting experimentations using *Scenedesmus* species, Gardener et al. (2011) observed the presence of large cells in pH 9.3 with majority of single cells and two-cell grouping. The authors also noticed larger cells with irregular conformations in pH 10.3.

There were no reports regarding decrease in cell size owing to higher alkaline ranges of pH (pH 12). Thus, cell size cannot be considered as a parameter to assess the growth of a micro algal species as undivided cells are likely to appear in large size and newly formed cells in small size under culture conditions.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	24.75	24.75	22	24.75	24.75	29.5	27.45	24.75
рН-3	27	24.75	27.5	29.24	24.75	29.5	27.5	27.5
рН-3.5	30.25	30.25	30.25	27.6	24.75	27.5	24.75	27.6
рН-4	33	33	28.4	24.5	24.75	29	29.5	29
рН-4.5	33	30.25	30.25	24.75	25.42	24.75	29.5	29.5
рН-5	38.5	33	27.5	22	27.5	24.5	22.45	27.5
рН-5.5	24.75	33	27.5	24.75	22.35	30.5	27.5	27.5
рН-6	27.5	27.5	27.5	29.5	22	30	27.5	27.5
рН-6.5	24.75	27.5	27.5	24.75	24.75	24.75	29.5	24.75
pH-7	24.75	24.75	30.25	27.25	27.5	24.75	24.5	24.75
рН-7.5	22	29.5	30.25	24.75	27.5	22	29.5	27.5
рН-8	27.5	33	27.5	27.25	27.5	24.75	27.75	27.5
рН-8.5	27.5	30.25	35.8	29.25	27.5	24.75	29.95	27.5
рН-9	27.5	29.25	33	27.5	26.5	27	30.25	27.5
рН-9.5	27.5	27.5	24.75	27.5	30.25	22	26.75	27.5
рН-10	27.5	27.5	24.75	24.5	24.75	24.75	28	24.75
рН10.5	27.5	27.5	24.75	22	22	22	32.25	24.75
pH-11	27.5	27.5	22.75	24.25	27.5	22	22.45	24.25
рН-11.5	27.5	27	22.5	24.5	24.75	22.45	22.75	24.5
рН-12	27.5	27	22	22	22	22	22	22

Table 2.1(f). Variation in cell size (µm) noticed in the cultures of *Chlamydomonas globosa*.

Experimental set	First day	Second day	Third day	Fourth day	Fifth day	Sixth day	Seventh day	Median
Control	46.75	46.75	41.25	38.5	44	49.5	46.75	46.75
рН-3	49.5	44	35.75	38.5	38.5	37.5	46.75	38.5
рН-3.5	46.75	41.25	44	27.5	38.5	33	44	41.25
рН-4	41.25	47	38.5	38.5	38.5	37.5	44	38.5
рН-4.5	41.25	47	44	35.75	30.25	27.5	35.75	35.75
рН-5	52.25	38.5	35.75	41.25	24.75	27.5	22	35.75
рН-5.5	55	44	41.25	38.5	27.5	27.5	27.5	38.5
рН-6	49.5	46.75	41.25	38.5	35.75	38.5	44	41.25
рН-6.5	46.75	44	44	38.5	35.75	38.5	46.75	44
pH-7	52.25	44	38.5	41.25	38.5	38.5	38.5	38.5
рН-7.5	46.75	44	41.25	38.5	38.5	46.75	44	44
рН-8	41.25	46.5	41.25	41.25	38.5	46.75	38.5	41.25
рН-8.5	46.75	46.5	41.25	44	38.5	44	41.25	44
рН-9	46.75	46.5	44	49.25	38.5	38.5	46.75	46.5
рН-9.5	41.25	46.5	46.75	44	41.25	38.5	44	44
pH-10	41.25	46.5	44	41.25	35.75	38.5	46.75	41.25
рН10.5	38.5	46.5	41.25	41.25	33	44	35.75	41.25
pH-11	38.5	41.225	41.25	41.25	33	33	35.75	38.5
рН-11.5	41.25	38.5	38.5	41.25	33	30.25	35.75	38.5
pH-12	41.25	30.25	30.25	41.25	33	41.25	27.5	33

Table 2.2(f).Variation in cell size (µm) noticed in the cultures of *Acutodesmus obliquus*.

## **Biomass**

In trials with *C. globosa*, maximum biomass was noticed in pH 9 (0.062g) and minimum in pH 4.5 (0.013g). In the control set, the biomass obtained was 0.031gm.

With *A. obliquus*, maximum biomass production was noticed in pH 9 (0.034 g) and minimum in acidic pH, ranging from 5.5 - 6.5 (0.014 g). In control set, the biomass obtained was noted to be 0.018 g.

Table 2.1(g). Variation in biomass (g) noticed in the cultures of *Chlamydomonas globosa*.

Experimental condition	Trial 1	Trial 2	Trial 3	Median
Control	0.026	0.03	0.031	0.031
рН-3	0.013	0.02	0.017	0.015
рН-3.5	0.016	0.02	0.015	0.016
pH-4	0.021	0.02	0.018	0.018
pH-4.5	0.019	0.01	0.013	0.013
рН-5	0.015	0.01	0.018	0.015
рН-5.5	0.028	0.02	0.019	0.019
рН-6	0.019	0.02	0.021	0.019
рН-6.5	0.019	0.03	0.027	0.027
pH-7	0.019	0.03	0.035	0.027
рН-7.5	0.028	0.04	0.036	0.036
рН-8	0.03	0.04	0.038	0.038
рН-8.5	0.038	0.06	0.056	0.056
рН-9	0.055	0.07	0.062	0.062
рН-9.5	0.052	0.08	0.053	0.053
pH-10	0.054	0.05	0.075	0.054
рН10.5	0.057	0.05	0.07	0.057
pH-11	0.06	0.07	0.072	0.07
рН-11.5	0.091	0.08	0.082	0.082
pH-12	0.099	0.08	0.08	0.08

Experimental condition	Trial 1	Trial 2	Trial 3	Median
Control	0.03	0.02	0.02	0.018
рН-3	0.017	0.02	0.01	0.017
рН-3.5	0.012	0.02	0.02	0.016
рН-4	0.014	0.02	0.02	0.017
рН-4.5	0.018	0.03	0.01	0.018
рН-5	0.024	0.03	0.02	0.024
рН-5.5	0.014	0.02	0.01	0.014
рН-6	0.013	0.01	0.02	0.014
рН-6.5	0.013	0.01	0.03	0.014
рН-7	0.013	0.02	0.02	0.024
рН-7.5	0.018	0.02	0.02	0.018
pH-8	0.016	0.03	0.03	0.032
рН-8.5	0.026	0.03	0.03	0.028
рН-9	0.03	0.03	0.04	0.034
рН-9.5	0.044	0.04	0.04	0.04
рН-10	0.05	0.05	0.05	0.05
рН10.5	0.051	0.06	0.07	0.06
рН-11	0.06	0.06	0.05	0.06
рН-11.5	0.06	0.05	0.08	0.06
рН-12	0.122	0.07	0.06	0.07

Table 2.2(g).Variation in biomass (g) noticed in the cultures of *Acutodesmus obliquus*.

Interestingly biomass content of respective microalgae in optimum ranges seems to be double that of the biomass content obtained in their respective control sets. Throughout experimentation, the biomass obtained from treatment sets above pH 9 was high, compared to other pH ranges, which was mainly due to precipitate formation. This was evidenced by low cell count and the white coloured precipitate formation within treatment sets.

Precipitation in culture medium owing to higher pH (pH 9 and above) was also confirmed in several reports (Becker, 1994; Visviki and Santikul, 2000; Sirisansaneeyakul et al., 2011; Dineshkumar et al., 2017). Such a situation is reported to influence the microalgal cell cycle adversely (Malis-Arad and MC Gowan, 1982 a,b). Lane and Burris (1981) and Gensemer et al. (1993) reported that extreme range of pH influences the growth, photosynthetic activity and nutrient assimilation of algae. Sakarika and Kornaros (2016) observed cell lysis in extreme acidic (pH 3 and 4) and alkaline ranges (pH 11) during the initial days of cultivation. Studies by Garcia et al. (2000), Taraldsvik and Myklestad (2000), Munir et al. (2015), Mandotra et al. (2016); Al-Safaar et al. (2016) noticed decline in growth at higher alkaline pH ranges. The reduction in growth at extreme range of pH can be due to the decrease in the rate of certain significant biochemical reactions as well as changes in cell membrane properties (Taraldsvik and Myklestad, 2000).

Throughout experimentation, the temperature pertaining to the culture medium of control and treatment sets were monitored. This has been carried out in a general view that microalgal growth fluctuates with an increase or decrease in temperature from the optimum range.

In the case of *C. globosa*, higher temperature was recorded on the fifth day  $(29.8^{\circ}C)$  and the lower temperature on the first and seventh day  $(29^{\circ}C)$ . Higher temperature of 29 °C was observed on the sixth day and lower temperature of 26 °C on the fourth day of the treatment set pertaining to *A. obliquus*. It was also noticed that there were no variations in the temperature among acidic, neutral and alkaline conditions. Moreover analysis of data on temperature envisaged that the temperature prevalent in the treatment sets were almost within the range to support the growth of micro algal members under study. Acién et al. (2013) reported 25-35°C as the optimal temperatures

range for freshwater microalgae and stated that for a short period, freshwater microalgae can tolerate temperature up to 40°C.

Each species of microalgae has a specific pH range in which appropriate growth and multiplication takes place (Sakarika and Kornaros, 2016). However, the optimal and tolerance ranges may vary between species. Various literatures suggest that the maximum algal growth occurs around neutral pH (pH 7 to 7.6) and the optimum pH is the initial culture pH at which microalgae adapted to grow (Visviki and Santikul, 2000 and Hansen, 2002). On the other hand, Gong et al. (2014) reported that the initial pH ranges have little impact on cell density. However, the optimum pH ranges pertaining to microalgal species may differ.

There were several reports regarding the optimum pH ranges of *Chlorella* species. The results of the research works pertaining to Yeh et al. (2010) suggested pH 8.5 as optimum range for *Chlorella vulgaris* ESP-3. On the other hand, Gong et al. (2014) observed maximum growth of *Chlorella vulgaris* at pH ranges 10- 10.5. At the same time, Sakarika and Kornaros (2016) recorded maximum biomass production of *Chlorella vulgaris* in pH 7.5. The experimentations of Khalil et al. (2010) confirmed enhanced growth of *Chlorella ellipsoidea* under alkaline conditions, especially at pH 10. Ma et al. (2017) observed sustainable biomass production of *Chlorella sorokiniana* GT-1 at pH 6.5. The investigations of Moheimani (2013) reported augmented biomass production of *Chlorella* sp. in pH 7. While Ponnuswamy et al. (2014) recommended the pH ranges of 4-7 as optimum. On the other hand, Rai et al. (2015) monitored higher biomass production in pH 7.

Attempts were also conducted to identify the optimum pH of *Scenedesmus* species. Xiao et al. (2011) recommended pH range of 6.5 - 8.5 as ideal for maximized growth of *Scenedesmus quadricauda*. Similarly Mandotra et al.

(2016) observed increased biomass production of *Scenedesmus abundans* in pH 8.

In 1997, Zhang et al. observed increased growth of *Chlorococcum* sp. in pH 8. For the maximized growth of *Chlamydomonas applanata*, Visviki and Santikul (2000) reported pH 7.4 as optimum. During experimentations, regarding the influence of pH on the growth of *Spirulina platensis*, Panday et al. (2010) noticed maximum biomass production at pH 9. The effect of pH on the growth of *Tetraselmis suecica* was analyzed by Moheimani (2013) and suggested pH 7.5 as most favorable range for enhanced biomass production. Bartley et al. (2014) observed pH 8–9 as ideal for enhanced growth of *Nanochloropsis salina*.

In the present study, while comparing the final median values of major growth parameters pertaining to the treatment sets containing *C. globosa,* maximum cell count was noticed at pH 8. In addition, increased turbidity and biomass content were obtained in pH 9. This indicates that the optimum pH requirements of *C. globosa* were ranging from 8-9.

Upon comparing the final median values of major growth parameters associated with *A. obliquus*, highest turbidity was noticed in pH 9, cell count in pH 5 and biomass in pH 9. Here, the algal member is showing a wide range of pH (5-9) ideal for growth and multiplication. Varshney et al. (2016) reported pH 8 as the optimum range for *Acutodesmus* sp. The results are indicative of the adaptability of *A.obliquus* even to acidic ranges and thereby its applicability for carbon sequestration experiments. This property of *A. obliquus* is significant as most micro algal members fails to develop in culture conditions, where there will be a drastic shift in pH to acidic conditions due to extraneous supply of carbon dioxide/ flue gas.

## **Correlation analysis**

Correlation analysis is a prominent tool, which helps in identifying the degree of relationship existing between the variables. In the present study, the relationship between pH (control set and optimum pH ranges) and major growth parameters like turbidity, cell count and biomass of two microalgal species have been analyzed statistically using Microsoft Excel. The significance of the each parameter was calculated using the t-value and the results are represented in Tables 2.3 and 2.4.

Table 2.3. Correlation of pH with respect to major growth parameters inChlamydomonas globosa.

Experimental set	Turbidity	Cell count	Biomass
Control set	4.10*	3.42*	1.42
pH 8	2.84*	2.69	4.91*
pH 8.5	2.68	2.87*	0.79
рН 9	2.42	3.02*	3.12*

(\* Significant at 5% level; NS: not significant (t-value at 5% level for 3 treatments and 3 replications =2.776)

In the present study, the relationship (t-value) between the optimum pH range and their respective growth parameters of *Chlamydomonas globosa* were worked out. Similarly, the t-values of control set with respect to growth parameters were also analyzed. Here in the case of treatment sets, the turbidity content of the pH 8, cell count values of both pH 8.5 and pH 9 and biomass content pertaining to pH 8 and pH 9 exhibited significant positive relation. In the case of control set, the turbidity content and cell count values exhibited significant positive correlation with control pH.

<b>Experimental set</b>	Turbidity	Cell count	Biomass
Control set	1.87	0.46	4.41*
рН 5	-2.57	-3.54*	-2.34
рН 9	3.89*	1.81	7.5*

 Table 2.4. Correlation of pH with respect to major growth parameters in

 Acutodesmus obliquus.

(\* Significant at 5% level; NS: not significant ( t value at 5% level for 3 treatments and 3 replications =2.776)

Here, with *Acutodesmus obliquus*, the t-values between the optimum pH ranges and their respective growth parameters were worked out. Likewise, the t-values of control set were also analyzed. While analyzing the relationship between pH 9and growth parameters the turbidity content and biomass content exhibited significant positive relation. Here in the case of pH 5, the cell count showed a significant negative relationship. In the case of control set, the biomass content exhibited significant positive relationship. In the case of control pH. The present work thus confirms the capabilities of *Acutodesmus obliquus* to grow in acidic ranges and thereby its utilization in carbon sequestration experimentations. For mass multiplication and subsequent utilization of *Chlamydomonas globosa* and *Acutodesmus obliquus* in Bolds Basal medium, the optimum pH range of respectively 8-9 and 5-9 can be followed.

#### **Summary and conclusion**

In recent times, microalgae have been employed for a wide range of beneficial purposes. For exploiting their potentials for such purposes, their biomass is required in sufficient quantities.

One among the major constraints in their utilization for such purposes is that most of the members, which are most abundant and virulent in natural conditions, seem to be slow growing in culture conditions. Moreover, the risk of maintaining cultures without contamination is another constraint in their utilization for multidimensional purposes. Hence, standardization of species specific culture conditions is a prerequisite for meeting the required targets.

pH is noted to be an important factor influencing the growth and multiplication of micro algal species in culture conditions. pH of the medium is also significant in carbon sequestration studies, as it determines the availability and solubility of  $CO_2$  and other nutrients. In the present study, an attempt has been carried out to assess the growth performances of selected microalgal members (*Chlamydomonas globosa* and *Acutodesmus obliquus*) in Bolds Basal medium under varying culture conditions, altered by pH.

For experimentation, treatment sets were maintained with conical flasks of 100 ml capacity (21 nos), each containing 50 ml Bolds Basal medium. The pH of culture medium was adjusted from 3-12, with a gradation of 0.5 using 0.5 N NaOH and 0.05 N HCl. After adjusting to the required pH, 5 ml of pure cultures of the respective micro algal members were added to the respective conical flasks and the changes in pH, if any, was readjusted. The control set was maintained at pH 6.6 (original pH of the culture medium).

The culture conditions like pH, temperature, conductivity and resistivity and micro algal growth parameters like turbidity, cell count, cell size, and biomass were monitored throughout the treatment period. The biomass estimation was carried out on the initial and final days of treatment. All the sets were kept at illumination during daytime (6 AM - 6 PM) and retained under light intensity of 40 µmol m<sup>2</sup> s<sup>-1</sup>. The temperature of the medium during experimentation ranged from 26 – 29.8°C while the humidity ranged from 48 - 54 %. Every day after observation, the altered pH was readjusted to their experimental condition. Monitoring of the treatment sets were carried out for a period of 7 days.

For the confirmation of appropriate pH level at which maximum growth of both microalgal members under the study had occurred, the median values of the 7 days triplicate data were worked out and compared. However, during experimentation, in higher alkaline ranges, formation of precipitate was noticed. Hence, for drawing conclusions regarding the optimum pH range in which maximum turbidity occurred, the final median values of pH ranges above 9 were neglected.

Upon comparing the final median values of major growth parameters pertaining to the treatment sets containing *Chlamydomonas globosa*, the maximum cell count was recorded at pH 8, whereas increased turbidity and biomass content were noticed in pH 9. While analyzing the final median values of major growth parameters pertaining to *Acutodesmus obliquus*, increased turbidity and biomass content were noticed in pH 9, while maximum cell count was observed in pH 5.

The present findings thus confirms that the optimum range of pH favouring the growth of *Chlamydomonas globosa* in BB medium was noted to be 8-9 and that of *Acutodesmus obliquus* at a wide range of 5-9. Hence for ensuring better biomass production of these micro algal members in BB medium, the above mentioned pH ranges can be followed.

## Chapter III

# PBR based feasibility studies on the carbon sequestration efficiency of selected micro algal members

#### Introduction

Screening studies carried out in Chapter I proved the efficiencies of micro algal species like *Chlamydomonas globosa* and *Acutodesmus obliquus* in assimilating extraneous supply of carbon dioxide. Attempts were also carried out to assess the optimum pH favouring the enhanced growth and development of *C. globosa* and *A.obliquus* in chapter II. As these species are sustaining in higher dosages of carbon dioxide, an attempt has been carried out to assess their better efficiencies in carbon dioxide accumulation using a proto type Photo Bio Reactor (PBR), which can ensure optimum conditions of algal growth and carbon assimilation.

PBRs can be defined as closed (or nearly closed) vessels for phototrophic production, comprising of a solid phase (microalgal cells), liquid phase (growth medium), gaseous phase ( $CO_2$  and  $O_2$ ) and super imposed light-radiation field (Posten, 2009). Based on the illuminated surface, PBRs are categorized as flat plate (Sierra et al., 2008; Slegers et al., 2011), tubular (Molina et al., 2001) and column type (Eriksen, 2008). Based on their mode of liquid flow, PBRs can be grouped as stirred type, bubble column and airlift reactor type (Gupta et al., 2015). Also there are wide ranges of hybrid type PBRs (Kumar et al., 2011).

Screening studies on carbon dioxide assimilation efficiencies of micro algae revealed the potentialities of *C. globosa* and *A. obliquus* under laboratory conditions. Their efficiencies were further assessed in a laboratory scale closed vertical bubble column PBR for optimization of their growth conditions and system efficiency for ensuring their industrial/commercial applicability. Though there are approaches on system designs for various micro algal members, including *A. obliquus*, no such efforts have ever been carried out on *C. globosa*, which are found to be effective in  $CO_2$  sequestration under present laboratory conditions.

## **Review of literature**

The first laboratory 'PBR', capable of controlling critical parameters like light, nutrients and temperature was developed in early 1940's (Myers and Clark, 1944; Ketchum et al., 1949). Since then, so much of R & D activities were carried out for their effective utilization in various sectors. The use of photo bioreactors for microalgal  $CO_2$  sequestration has gained significant attention in recent times, as it offers increased microalgal productivity due to controlled environmental conditions, optimized space utilization and exhibits higher photosynthetic efficiency compared to open systems.

Several factors need to be considered while designing a photobioreactor for carbon sequestration. For better performance of the system, suitable candidate species that can manage extremes of pH, temperature, shear stress (Kumar et al., 2011) and having high growth rate and low risk of contamination are preferred. Moreover, proper mixing should be ensured for efficient light distribution to cells (Kumar et al., 2011). The continuous removal of oxygen is also essential, as excessive concentration of dissolved oxygen inhibits photosynthesis (Molina et al., 2001). The selection of optimum cell concentration, not all the supplied light energy is captured by the cells (Zhang et al., 2001) though, highly dense culture makes cells more tolerant to high  $CO_2$  concentration (Chiu et al., 2008). The identification of suitable light

source may avoid the problem of availability of light to the culture and also the formation of biofilms.

Until now, several researchers worked on developing novel photobioreactors for improving the  $CO_2$  fixation efficiencies of microalgae by overcoming the existing drawbacks and offering significant yield with high production efficiency and reduction of biological contamination (Chisti, 2007; Huntley and Redalje, 2007; Vasudevan and Briggs, 2008). Some of the attempts in this direction are detailed below.

A gas recycling photobioreactor was developed by Yun and Park (1997) for *Chlorella vulgaris* UTEX -259 with increasing gas retention time and high  $CO_2$  fixation rate. With increased gas retention time, the  $CO_2$  concentrations can be maintained in optimal ranges for algal photosynthesis. It has also been reported that, using the same photobioreactor, microalgal species without  $CO_2$  tolerance can also be cultivated without growth inhibition.

A modified flat-plate photobioreactor has been developed by Hu et al. (1998) for obtaining high cell density under elevated  $CO_2$  concentrations. The reactor was made up of an acrylic plastic plate, with an outer and inner chamber. The outer chamber functions as a temperature regulator and the bubbling tube was placed at the bottom of the inner chamber. Studies using this photobioreactor revealed that, daily replacement of the culture medium was effective in maintaining ultrahigh-cell-density culture of  $CO_2$  tolerant unicellular *Chlorococcum littorale*.

The carbon dioxide fixation efficiencies of *Euglena gracilis* were explored by Chae et al. (2006) by using a laboratory-scale photo-bioreactor (100 L working volume) with baffles stimulated plug-flow of the culture medium. The effect of lightwas minimized photo-bioreactor (100 L working volume) to 20 cm. A pilot scale novel photo-bioreactor (1000 L working volume) using sunlight as energy source and flue gas as carbon source was also fabricated for the experimentation. It was designed in such a way that, the dark and light regions were separated using a cover and a scraper for the internal circulation of culture medium between light and dark regions. The experimentations showed an enhanced cell yield of ten times in pilot-scale reactor than the laboratory-scale reactor.

A membrane-sparged helical tubular photobioreactor (MSTR) with a working volume of 800 ml was designed for monitoring the carbon dioxide fixation efficiencies of *Chlorella vulgaris* (Fan et al., 2008). The performance of MSTR, including the light intensity, gas flow rate and characteristics of membrane module on  $CO_2$  fixation were compared with draft tube airlift photobioreactor, a bubble column and a membrane contactor. The results revealed that the limitation of  $CO_2$  removal was improved in MSTR and registered higher  $CO_2$  fixation rate (0.95–5.40 times) when compared to the conventional reactors. Jana et al. (2017) investigated the carbon dioxide bio fixation potential of *Arthrospira* sp. cultured in membrane photobioreactor (MPBR) of cylindrical glass vessel. The authors reported effective  $CO_2$  dissolution and significant removal of dissolved oxygen by bentonite clay derived membrane.

Studies were conducted to improve the  $CO_2$  utilization efficiency of *Chlorella* sp. AG10002 in vertical tubular photobioreactor (Ryu et al., 2009). The work as a whole optimized the bubble size for maximum cell concentration and noticed that, while culturing using large bubbles, the mass transfer may decreases due to reduction in the interfacial area between gas and liquid. In addition, they recommended the utilization of slanted cross-sectional type baffle for increased cell growth. To determine the economic and technical feasibility of micro algae-based carbon sequestration, Wilson et al. (2014)

cultured *Scenedesmus acutus* using flue gas through house designed closed loop, vertical tube photobioreactor.

Fransico et al. (2010) evaluated the carbon dioxide sequestration efficiencies, lipid productivity and biofuel quality of six microalgal strains (*Aphanothece, Dunaliella, Chlorella, Scenedesmus, Phormidium and Phaeodactylum*) in bubble column photobioreactor with working volume of 3.0 L.A dispersion system with 1.5 cm diameter was positioned at the center of the column and the reactor was continuously illuminated. The comparison of the results reported *Chlorella vulgaris* as better strain.

A novel assembled culture system, fermentor-helical combined photobioreactor was designed by Jia et al. (2011). In this reactor, the temperature and pH can be controlled in the fermentor and the  $CO_2$  fixation can be accomplished in the helical tube.  $CO_2$  removal ratio of 95% was achieved in this novel culture system.

A novel graphical integration technique for quantifying the moles of  $CO_2$  sequestered was developed by Kargupta et al. (2015) for evaluating the  $CO_2$ sequestration potentialities of *Chlorella pyrenoidosa* and *Scenedesmus abundans* using tubular batch photobioreactor. For experimentation, three completely mixed tubular batch reactors (CMTBRs) were connected in parallel and the gas was continuously supplied through the base of the reactor. Using mirror arrangement, the illumination was provided by a fluorescent lamp. The results showed that continuous bubbling enhanced the biomass productivity and  $CO_2$  sequestration efficiencies of microalgae.

Naderi et al. (2015) evaluated the carbon biofixation efficiencies of *Chlorella vulgaris* under different light intensities using a 3-L bench top bioreactor (New Brunswick's BioFlo/CelliGen 115) of 2-L working volume. The system was enclosed by a water jacket to achieve precise temperature control. The
DO and pH meters were fixed at the top of the bioreactor. Using a Resistance Temperature Detector (RTD), the culture temperature was measured.

A promising approach of integrated use of wastewater and flue gas for  $CO_2$  fixation and biomass production was established by Kuo et al. (2016) through a column-type glass-fabricated photobioreactor with 1 L working volume. The effect of high gas superficial velocity on  $CO_2$  capture from air by *Chlorella vulgaris* using airlift bioreactor was investigated by Sadeghizadeh et al. (2017). The authors commented that *Chlorella vulgaris* exhibited resistance to shear stress and enhanced potentialities in growth and  $CO_2$  capture under high input gas superficial velocities.

A few attempts were also conducted using flask type photobioreactors. A flask culture photobioreactor was developed by Fluke et al. (2015) to screen the efficient microalgal species for high carbon dioxide fixation. The photobioreactor consists of an upper 250 ml Erlenmeyer flask for microalgal growth and a lower flask with buffer mixture for  $CO_2$  generation. The LED based light panels were used as light source and the  $CO_2$  partial pressure was maintained constantly by reloading the buffer at an interval of 48 hours. The  $CO_2$  partial pressure in headspace of the flask was accurately maintained in different concentrations using buffer mixture. The authors recommended culture flask photobioreactor for screening of microalgal strain for  $CO_2$  sequestration efficiency.

Duarte et al. (2017) conducted batch cultivation studies in closed Erlenmeyer flask-type photobioreactors with a working volume of 1.8 L. The gaseous mixture was supplied to the cultures at an interval of 2 hours of the light period for 10 min. through a porous curtain spurger, placed on the base of photobioreactor. The authors confirmed the approach of intermittent flue gas supply to the culture medium, which enhances the growth and  $CO_2$  biofixation by *Chlorella fusca* and *Spirulina* sp.

For improving the  $CO_2$  fixation efficiencies, Ho et al. (2013) cultivated *Scenedesmus obliquus* CNW-N in 1-liter glass vessel, illuminated through external light source. The experimentation showed that the highest  $CO_2$ fixation rate was achieved by 50% replacement of culture medium. The lipid productivity and carbon dioxide fixation rate of *Chlorella protothecoides* were monitored and reported by Binnal and Babu (2017) in a 5L lab scale photobioreactor made up of borosilicate vessel. The authors adopted Response Surface Methodology (RSM) to optimize the environmental condition.

A series of experimentations on the effect of flue gas on the growth of *Scenedesmus dimorphus* were conducted by Jiang et al. (2013) by employing glass columns photobioreactors. The toxicity of flue gas was reduced by adopting strategies like neutralization of culture medium by CaCO<sub>3</sub> addition and intermittent spurging by pH feedback control. The authors reported that the obstacles were well controlled and resulted in increased algal cell growth.

Optimization studies on the operational parameters like enhancement of height, intermittent supply of  $CO_2$  and reduction of flow rates to lengthen the  $CO_2$  residence time has been conducted by Basu et al. (2015). For maximizing the  $CO_2$  fixation efficiency of *Scenedesmus obliquus* SA1 (KC733762) the open cylindrical glass tube was employed. The authors fixed 15%  $CO_2$  supply at 0.43 LPH for 12 h per day as an optimum condition.

Literatures are also available on the comparison of the efficiencies of photobioreactors. The  $CO_2$  utilization efficiencies of three types of gasspurged photobioreactor designs, including a flat plate (working volume of 7.0 dm<sup>3</sup>), a polyethene column (working volume of 20 dm<sup>3</sup>) and a bubble column (working volume of 1.4 dm<sup>3</sup>) were attempted by Lakaniemi et al. (2012). For experimentation, *Chlorella vulgaris* was cultivated. The authors observed long time  $CO_2$  retention and higher carbon dioxide utilization efficiencies in bubble column.

The carbon dioxide fixation efficiency of *Haematococcus pluvialis* under both indoor (working volume 5 L photobioreactor of 1 column) and outdoor conditions with working volume of 20 L (4 sequential columns) was monitored by Lee et al. (2015). The carbon distribution pattern in the photobioreactor was also investigated. Based on the results, authors stated that, enhanced carbon fixation efficiency of 4 fold under indoor and 3.63 fold under outdoor conditions can be attained through sequential operation system.

The efficiencies of a single reactor and six-parallel photobioreactors were compared for  $CO_2$  mitigation using *Chlorella* species (Chiu et al., 2008). A cylindrical glass reactor (30 cm length, 7 cm diameter) with a working volume of 800 ml was designed and used. The results revealed that the  $CO_2$  reduction, biomass production and lipid production were six times superior in the six-parallel photobioreactors than that of single photobioreactor. However, the  $CO_2$  removal efficiencies were noted to be similar in both cases. The strategy of employing high initial cell density with photobioreactor operation, in series, was adopted by Yadav et al. (2015) towards reducing the hindering effect of pure flue gas and to maximize the  $CO_2$  fixation. A bubble column photobioreactor with a working volume of 500 mL for *Chlorella* species was designed.

Cheng et al. (2013) conducted studies to improve the carbon fixation efficiency of *Chlorella* PY-ZU1 in sequential column bioreactor. By operating in a multi-stage sequential bioreactor,  $CO_2$  was categorized and utilized. The authors reported that the residence time of  $CO_2$  has enhanced exponentially and the multi-capture by microalgae ultimately resulted in increased  $CO_2$  fixation efficiency.

The strategy of internal illumination of the cultures was also carried out. An internally illuminated photobioreactor (IIPBR) of 18-1 prototype was designed by Pegallapati and Nirmalakhandan (2011) for cultivating *Nannochloropsis salina and Scenedesmus* sp. The PBR was developed according to the principles of airlift/bubble column. For maximizing biomass productivity, spurging Carbondioxide- Enriched Air (CEA) was attempted. The authors claimed that the light utilization can be improved through internal illumination and the energy for mixing can be reduced through the airlift operations. When compared to normal bubble columns, the IIPBR has the potential to minimize the energy input to 50%.

The idea of internal illumination was also attempted by Kurano et al. (1995). The  $CO_2$  fixation capability of *Chlorococcum littorale* was monitored using PBR of three types of culture vessels. A large vessel illuminated via fluorescent lamps was placed inside the culture medium and other two vessels were illuminated from outside. The authors noticed maximum  $CO_2$  fixation rate in the smallest vessel due to increased irradiation and daily medium exchange by cell harvest.

Research works were also conducted to develop photobioreactors to utilize sunlight as energy source. Hirata et al. (1996) described a photobioreactor equipped with specially designed illumination plate made of Pyrex glass for collecting sunlight. The efficiency of sun light collection and transmission to the algal cells was streamlined. Results obtained from photobioreactor were compared with Roux flasks illuminated with fluorescent or xenon lamp. The results revealed that the cell growth and  $CO_2$  fixation rate achieved in the photobioreactor were lower than Roux flask cultures. The photobioreactors with solar collectors to monitor its feasibility in  $CO_2$  biofixation was also developed by Ono and Cuello (2006). The authors stated that, high initial cost constraints associated with solar collector and photobioreactor can equalize through the reduction in land-related cost.

Studies pertaining to the outdoor cultivation of microalgal species using various photobioreactors were carried out and reported. An outdoor air lift photobioreactor with working volume  $0.1 \text{ m}^3$  was designed for the flue gas reduction using *Scenedesmus obliquus*. The photobioreactor consists of an external pump for circulating flow between the down-comer and riser tube to enhance the CO<sub>2</sub> transfer. The results of experimentation revealed CO<sub>2</sub> removal ratio of 67% in the pilot scale system (Li et al., 2011).

The feasibility of outdoor cultivation of *Chlorella vulgaris* in 80 L (8 L x 10 sets) bubble column PBR using various  $CO_2$  input concentrations was monitored by Guo et al. (2015). The possibility of culturing *Acutodesmus obliquus* in large-scale outdoor tubular photobioreactor (500-L) using gaseous emissions from a methanol plant was reported by Chen et al. (2016). In the OTP, the gas exchange and circulation of the algal suspension was accomplished with the aid of a pneumatic diaphragm pump. The maximum biomass obtained in the OTP was consistent with that of the indoor experimentations. Eloka-Eboka and Inambao (2017) cultured *Chlorella vulgaris, Dunaliella, Scenedesmus quadricauda* and *Synechococcus* sp. in both open pond and closed photobioreactor of model BF-115 Bioflo/celliGen. The open pond consists of working volume 850 L exposed to 12 h continuous sun light. It was reported that, species with lower biomass productivity exhibited increased lipid content. The authors recommended *Dunaliella* as an efficient strain in  $CO_2$  sequestration.

Upon analyzing the literature, it has been noticed that though PBRs were widely used and have several advantages over open systems, still there are major drawbacks that make them uneconomical. These are in the areas of species selection, system design and culture conditions. Lack of appropriate candidate species for withstanding fluctuations in pH and temperature, inadequate temperature controlling facility, biofilm formation, improper mixing of microalgal culture with devices generating heat, damages due to aeration, inadequate arrangements/positioning of suitable illumination source etc. were noticed as major obstacles in the design and scaling up of PBR technology.

In the present study, a proto type PBR (closed, vertical bubble column type) was designed to assess the carbon sequestration efficiencies of micro algal species like *C. globosa* and *A. obliquus*. A review of the literature revealed that most of the studies regarding engineered design structures for carbondioxide sequestration are confined to limited number of microalgal species like *Chlorella* and *Scenedesmus*. Though there are fragmentary information pertaining to *A. obliquus*, reports regarding photobioreactor design for *C. globosa* is scanty. Thus the present objective of the study is to:

• Assess the CO<sub>2</sub> assimilation efficiency of *Chlamydomonas globosa* and *Acutodesmus obliquus*, cultured in a PBR under controlled conditions.

### Materials and methods

The growth performances of *Acutodesmus obliquus* and *Chlamydomonas globosa*, which survived under higher levels of carbon dioxides supply, were assessed for their carbon dioxide sequestration potentialities using a proto type Photo bio reactor maintained at controlled conditions. Description of the PBR used in the study is as follows:

#### PBR design

A closed vertical bubble column structure having a size of  $10 \times 10 \times 45$  cm and a volume of 4.5 L was designed using Acrylic material. The technical details concerning the experimental system is given in Plate 6 a and the

photograph in Plate 6 b. For the supply of both culture medium and carbon dioxide gas, an inlet, with control facility was maintained at the bottom. Similarly an outlet was maintained at the top of the column to permit the gases within the column, if any, to come out. The outlet of the PBR is connected to the inlet of a reservoir of culture medium kept in a closed / sealed bottle (2 L). The outlet of the PBR is retained in the bottom of the reservoir to permit the gases, if any, to bubble out through the medium. A controlled outlet is also provided to the reservoir for the release of gases, if any.

#### **Experimental layout**

For experimentation, micro algal cultures, which were maintained in respective pH, as outlined in Chapter II were used. At a time a single experiment was carried out with a particular species and a particular carbon dioxide dosage.

For experimentation, pure cultures of candidate species (3 litres) with an initial concentration of 0.5 OD, maintained in bolds basal medium was transferred to the vertical column of the PBR through its inlet. Care was taken to maintain adequate space for the accumulation of gases in the upper side of the PBR. The other end of the inlet was connected to a cylinder of carbon dioxide, having control facilities. The outlet of the PBR was connected to the inlet of the reservoir containing 2 liters of Bolds Basal medium. The inlet of the reservoir was maintained in such a way to permit bubbling of gases, if any, which passes through it.

The carbon dioxide from the cylinder, having control facilities, was allowed to bubble at regular intervals through the culture medium contained in the column of the PBR. The frequency of carbon dioxide supply was set for 60 or 80 bubbles /2 hours from 6 am to 6 pm. The PBR was maintained at a light

intensity of 100 $\mu$ mol m<sup>-2</sup> s<sup>-1</sup> using LED lights and fluorescent tubes. The room temperature was maintained at 28±0.5°C and humidity at 48%. For monitoring the nature and magnitude of the gas generated at the top chamber of the vertical closed column structure, the reservoir containing BB medium (devoid of microalgae) was monitored for gaseous influxes, specifically with respect to CO<sub>2</sub> and O<sub>2</sub>. The carbon sequestration efficiencies of both microalgal members were monitored for 3 and 6 days in separate trials.

During experimentation,  $CO_2$ , as per its pre-determined magnitude and frequency was supplied through the inlet of the PBR. The physico-chemical characteristics of the culture medium within PBR (before and after experimentation), together with growth performances and biochemical characteristics of microalgae (before and after experimentation) were carried out. Along with studies on the culture medium contained in PBR, the medium contained in the reservoir was also subjected to the analysis of pH, free carbon dioxide and dissolved oxygen for determining the changes, if any, associated with them due to gaseous influx.

The characteristics of the culture medium were worked out on the first day, before experimentation and after experimentation for the stipulated period of 3/6 days. The physicochemical parameters analyzed include pH, temperature, conductivity, resistivity, free carbon dioxide, dissolved oxygen and alkalinity following Trivedi and Goel (1987). Growth parameters associated with microalgal members, which were attempted before and after experimentation include turbidity, specific growth, biomass productivity, cell count and cell size. The biochemical parameters associated with micro algae like pigments (Shof and Lium, 1976), total lipid (Bligh and Dyer, 1959), protein (Lowry et al., 1951), total carbohydrate (Dubois et al., 1956) and mineral composition (Pancha et al., 2015)were worked out on the initial day and final day of experimentation.

The methods followed for the estimation of pH, conductivity, resistivity, free carbon dioxide, dissolved oxygen and alkalinity of the culture medium and turbidity, cell count and cell size associated with micro algal growth are already depicted in Chapter 1. Other parameters attempted, like biomass productivity and specific growth, pigments, total lipids, total protein, total carbohydrate and mineral components are as per the details listed below:

# Biomass productivity and specific growth (Chen et al., 2016; Guillard and Ryther, 1962)

The biomass productivity during treatment period was determined by recording the optical density of microalgal culture at 684 nm using spectrophotometer (Systronics, 2201).Specific growth rate is the measure of number of generations (the number of doublings) that occur per unit of time. The specific growth rate was obtained using following equation (Guillard and Ryther, 1962):

$$\mu = \frac{\ln (Nt/No)}{Tt - To}$$

Nt = OD value on final day of treatment

- No = OD value on initial day treatment
- Tt = Final day of treatment
- To = Starting day of treatment

# Pigments - chlorophyll (Shof and Lium, 1976)

**Reagents:** 

DMSO

#### **Procedure:**

Due to the critical "light harvesting" role in photosynthesis, the measurements of photosynthetic pigments have received considerable attention. The pigments like chlorophyll *a*, chlorophyll *b*, total chlorophyll and carotenoids were estimated using Shof and Lium (1976) method. To the 0.025 g of fresh microalgal sample, 7 ml of Dimethyl sulphoxide was added and kept in an oven at 60°C for one hour. The extract was then transferred to a graduated tube and made up to a total volume of 10 ml with DMSO (can be assayed immediately or transferred to vials and stored between 0-4°C until required for analysis). The optical density was recorded at 480nm, 645nm and 663nm using a spectrophotometer (Systronics 2201). Quantitative estimation of the pigments was carried out using the following equations:

Chlorophyll a = 
$$\frac{12.7 \times \text{OD}663 - 2.69 \times \text{OD}645}{1000 \times \text{W} \times 1} \times \text{V} \times \text{df}$$

Chlorophyll b= 
$$\frac{22.9 \times \text{OD}645 - 4.68 \times \text{OD}6663}{1000 \times \text{W} \times 1} \times \text{V} \times \text{df}$$
  
Total Chlorophyll= 
$$\frac{20.2 \times \text{OD}645 + 8.02 \times \text{OD}6663}{1000 \times \text{W} \times 1} \times \text{V} \times \text{df}$$

Carotenoid = OD480+[(0.114× OD663)-(0.638× OD645)]× $\frac{V \times df}{1000 \times W}$ 

V = Volume

df = Dilution factor

- W = Weight of the microalgae taken
- OD = Optical density at a particular wave length

# Total Lipids (Bligh and Dyer, 1959)

## **Reagents used:**

Extraction solvent- Chloroform: Methanol: Distilled water (5:10:4)

## **Procedure:**

For estimating the total lipid content, 20 mg fresh microalgal sample was homogenized using extraction solvent (Chloroform: Methanol: Distilled water-5:10:4) and centrifuged at 4000 rpm for 10 minutes. After keeping in room temperature for overnight the solution was again homogenized using 5 ml Chloroform and 5 ml Distilled water and centrifuged. Carefully transferred the lower layer of the centrifuge tube to a pre weighed beaker using syringe or micropipette and kept in room temperature for evaporation. The dried beakers along with extracts were weighed. The total lipids were calculated from the difference in final weight of beaker and the weight of the beaker alone and expressed in percentage.

## Total Protein (Lowry et al., 1951)

## **Reagents used:**

- 1) 10% TCA Reagent A
- 2) 1N Sodium Hydroxide- Reagent B
- Alkaline Sodium Carbonate Solution (Dissolving 2 g of Sodium Carbonate in 100 ml of 0.1N Sodium hydroxide) -Reagent C
- 4) Copper sulphate-Sodium potassium tartarate solution (Mixing 0.5% Copper sulphate and 1% Sodium potassium tartarate solution) -Reagent D
- 5) Alkaline copper reagent (Mixing 50 ml reagent C and 1 ml of reagent D) Reagent E
- 6) Folin-ciocalteu regent

#### **Procedure:**

Total protein content was determined by using the method of Lowry et al. (1951). The microalgal cultures were centrifuged and washed with distilled water. For extraction, 20 mg of fresh microalgal sample was homogenized using reagent A and centrifuged at 5000 rpm for 10 minutes. The pellets were then treated with reagent B and boiled for 30 minutes, cooled and centrifuged. The resulting supernatant was then made up to a known volume. For estimation, to the 1 ml of sample (0.1 ml of the supernatant mixed with 0.9 ml distilled water) 5ml of reagent E were added and incubated for 10 minutes. Finally, 0.5 ml of folin-ciocalteu regent was also added. After 30 minutes of incubation the absorbance at 750 nm was measured. The amount of total protein was calculated from the standard graph prepared using Bovin Serum Albumin (BSA) and the reagents were taken as blank.

### Total Carbohydrates (Dubois et al., 1956)

## **Reagents used:**

- 1) Phosphate buffer 0.1M (pH 6.8)
- 2) 5% phenol
- 3) Sulphuric acid

## **Procedure:**

Total carbohydrate contents of microalgae were determined by phenol -  $H_2SO_4$  method (Dubois et al., 1956). 20 mg of fresh microalgal biomass was homogenized in 5ml of sodium phosphate buffer (pH 6.8) and centrifuged at 5000 rpm for 10 minutes. To the 1 ml of the supernatent 5% phenol solution and 5ml of sulphuric acid were added and mixed thoroughly. The sample was then kept for incubation at room temperature for 30 minutes and the

absorbance was measured at 490 nm against a reagent blank using spectrophotometer. Glucose was used as a standard for the preparation of calibration graph.

#### Mineral composition (Pancha et al., 2015)

## **Reagents used:**

- 1) Conc.  $HNO_3$
- 2) HClO<sub>4</sub>
- 3)  $H_2SO_4$
- 4) 2% HCl solution

#### **Procedure:**

For estimation of the mineral content, 100 mg of dried microalgal biomass was taken in a beaker and digested using 10 ml of Conc. HNO<sub>3</sub> for overnight. To the digested sample, 2.5 ml HClO<sub>4</sub> and 250  $\mu$ l H<sub>2</sub>SO<sub>4</sub> was added on the next day and heated until the elimination of white smoke. The resulting content was then dissolved in 100 ml of 2% HCl solution and filtered using Whatsmann filter paper. Using the flame photometer (Systronics, 128) the analysis of sodium and potassium was carried out.

The results of the above analyses were depicted and interpretations were made in results and discussion. The results of major influencing parameters pertaining to both treatment sets of *C. globosa* and *A. obliquus* were also subjected to statistical analysis using t-test.

## **Results and discussion**

Although microalgal culturing using photobioreactors have been in practice since 1940s, their multi-oriented application and scale up activities are still a

challenge. Efforts are going on, both in field and laboratory conditions, to improve the productivities of various micro algal members at low cost. Here in the present study a prototype of a closed vertical bubble column type PBR was developed under laboratory conditions for maximizing the growth of micro algal species like *C. globosa* and *A. obliquus* under elevated levels of carbon dioxide.

The study has been undertaken during August to November, 2017. Separate studies were carried out for both microalgal members. However optimum conditions of temperature ( $28\pm0.5^{\circ}$ C), humidity (48%) and light intensity ( $100\mu$ mol m<sup>-2</sup> s<sup>-1</sup>) were followed throughout the study. The parameters selected (physico chemical characteristics of culture medium and growth and biochemical characteristics of micro algal members) were worked out before (control) and after experimentation ( $3^{rd}$  or  $6^{th}$  day) and the results are reported.

An attempt has also been carried out to standardize the bubbling frequencies (60 and 80 bubbles/2 hours) in accordance with the extent of  $CO_2$  dissolved in a closed system of Bolds Basal medium. This has been carried out by bubbling  $CO_2$  gas from the source at specific frequencies through the medium (3.0 litres) taken in the closed chamber of the PBR. The resultant free  $CO_2$  content associated with the medium was then analyzed titrimetrically using phenolphthalein and 0.05N NaOH and the results are represented in mg/L.

Accordingly, it has been noticed that a bubbling frequency of 60 and 80 bubbles/2 hours in 3.0 litres of Bolds Basal medium retained 127.6 and 145.2mg/L of free CO2 content, respectively, within BB medium.

#### Studies on Chlamydomonas globosa

The responses of *C. globosa* with respect to optimum flow rate of carbon dioxide (60 bubbles/2 hours) from 6 am to 6 pm has been monitored for 3 and 6 days under optimum conditions. The median values of the results

concerning culture media and morphological / biochemical characteristics of the *C. globosa* are depicted in table 3.1. For drawing conclusions regarding the performances of micro algae in closed vertical bubble column structure under optimum carbon dioxide flow rate, the percentage difference of the major influencing parameters were worked out.

Upon comparing the results of *C. globosa*, a marked decrease in the pH values of both 3 days (4.78%) and 6 days treatment was noticed (9.84%). With respect to free carbon dioxide content, an increase of 249.1% was noticed in 3 days and 181.82 % in 6 days of treatment. In the case of dissolved oxygen content, an increase was noticed in the  $3^{rd}$ day (400%) and the  $6^{th}$  day (386.11%).

In the case of microalgal growth parameters, maximum turbidity was observed after 6 days of treatment (167.27%) than 3 days (156.96%). Upon comparing the results pertaining to biomass productivity a marked increase of 230.5% was recorded in 6 days of treatment than 3 days (156.4%). Also the cell count of 6 days treatment showed significant increase with 237.73% than 3 days of treatment (81.92%). While comparing the results of cell size, higher value of 25% was noticed in 3 days treatment than 6 days of treatment (6.66%), compared to their initial values.

On comparison of the pigment content of microalgae concerning both treatment sets, chlorophyll *a* content of the 3 days treatment showed lesser values (32.675%) than 6 days treatment (36.39%). Similarly a lower value of 39.45% was noted for chlorophyll *b* content in 3 days treatment, whereas a higher value of 41.337% was noticed with 6 days of treatment. Upon comparing the percentage differences of total chlorophyll content, the 3 days treatment exhibited a decrease of 38.72%, while 6 days treatment showed an increase of 38.08%. Here the carotenoid content exhibited a marked decrease of 12.844% in 3 days of treatment and 31.648% in 6 days.

Upon comparison of the biochemical parameters, lipid content showed a significant increase in 3 days of treatment (185.7%) than 6 days (14.32%). An increase of 21.186% of protein content was observed in 3 days of treatment, while it was considerably less (7.2%) in 6 days of treatment. In the case of carbohydrate content, a marked increase was noticed in 6 days of treatment (128.6%), whereas a lower value of 12.5 % was recorded in 3 days of treatment.

 Table 3.1. Changes in culture media characteristics and morphological / biochemical responses of Chlamydomonas globosa to varying durations of carbon dioxide supply.

SI.	Parameters Analysed	Units	CO <sub>2</sub> supply hours of int	in 60 bubbles / 2 erval for 3 days	CO <sub>2</sub> supply in 60 bubbles / 2 hours of interval for 6 days	
INO			Initial day	Final day	Initial day	Final day
1.	рН	-	7.12	6.78	7.01	6.32
2.	Conductivity	μs	832.7	784.6	834.5	777.8
3.	Resistivity	kΩ	1.319	1.400	1.316	1.411
4.	Dissolved Oxygen	mg/L	6.8	34	7.2	35
5.	Free Carbon dioxide	mg/L	48.4	168.96	48.4	136.4
6.	Alkalinity	mg/L	120	160	140	145
7.	Turbidity	NTU	46	118.2	50.1	133.9
8.	Specific growth	μd <sup>-1</sup>	0.136		0.071	
9.	<b>Biomass production</b>	g/L	1.545	3.962	1.495	4.941
10.	Cell Count	cells/ml	$0.448 \ge 10^6$	$0.815 \text{ x}10^6$	$0.387 \times 10^{6}$	$1.307 \text{ x} 10^6$
11.	Cell size	μm	30	37.5	37.5	40
12.	Chlorophyll a	mg/ g <sup>-1</sup>	8.808	5.93	6.859	9.355
13.	Chlorophyll b	$mg/g^{-1}$	4.771	2.889	3.544	5.009
14.	Total Chlorophyll	$mg/g^{-1}$	14.36	8.8	10.4	14.36
15.	Carotenoid	$mg/g^{-1}$	0.436	0.38	0.534	0.365
16.	Lipid	%	10	12	10	8
17.	Protein	mg/mg	0.118	0.143	0.152	0.141
18.	Carbohydrate	mg/mg	0.008	0.009	0.014	0.032
19.	Sodium	ppm	17.1	10.7	4.3	16
20.	Potassium	ppm	20.5	25.3	26.8	24.9

During experimentation with *C. globosa*, the medium contained in the reservoir was assessed for parameters like pH, free carbon dioxide and dissolved oxygen for estimating the magnitude and nature of release of gases from the PBR. The results are depicted in table 3.2.

While comparing the pH of the bolds basal medium contained in the reservoir for 3 treatment days, an increase was noticed in the final day (6.88), compared to initial day (6.6). On the other hand, in 6 days treatment, the pH of the final day (6.66) exhibited a small decrease with respect to initial value (6.7). Similarly, in the case of 3 days treatment, a decrease in the free carbon dioxide content was observed in the final day (61.6 mg/L), when compared to the initial value (74.8 mg/L). Conversely in 6 days treatment, an increase in the final free carbon dioxide content (74.8 mg/L) was observed with respect to initial value (61.6 mg/L). However, there were no marked differences in the final values of dissolved oxygen content of the 3 days (8 mg/L) and 6 days treatment sets (7.4 mg/L) compared to their initial values of 8 mg/L and 7.4 mg/L, respectively.

Table 3. 2. Results on the analysis of Bolds Basal medium contained in the outlet reservoir of PBR on treatment studies with *Chlamydomonas globosa*.

Sl.No	Parameters Analysed	CO <sub>2</sub> supply of 2 hours for an in	60 bubbles / nterval of 3 days	CO <sub>2</sub> supply in 60 bubbles / 2 hours for an interval of 6 days	
		Initial day	Final day	Initial day	Final day
1.	рН	6.6	6.88	6.7	6.66
2.	Free Carbon dioxide (mg/L)	74.8	61.6	61.6	74.8
3.	Dissolved Oxygen(mg/L)	8	8	7.4	7.4

#### Studies with Acutodesmus obliquus

Similar to *C. globosa*, the responses of *A. obliquus* with respect to an optimum flow rate of carbon dioxide (80 bubbles/2 hours) has been monitored under ideal conditions. The median values of the results concerning the culture media and morphological / biochemical results concerning *A. obliquus* are depicted in table 3.3.

Analysis of the results regarding pH values showed a significant reduction in both 3 days (23.7%) and 6 days (20.65%) of treatment. In the case of free carbon dioxide content, an increase of 980 % was noticed with 3 days and 900% with 6 days of treatment. In the case of dissolved oxygen content, a significant increase of 200% was noticed in 6 days of treatment and an increase of only 79.51% with 3 days of treatment.

Upon comparing the major growth parameters, in the case of turbidity, marked increase was noticed in 6 days (110.5%), while an average increase of 31.175% was noticed in 3 days of treatment. An increase in biomass productivity was noticed in 6 days (89.686%) and 3 days of treatment (59.6%).Similarly the cell count of micro algae at 6 days of treatment exhibited a significant increase of 200% while an increase of only 79.51% was noticed in 3 days of treatment. Comparing the results, decrease in cell size was noticed in both 3 days (13.33%) and 6 days of treatment (6.66%).

In the case of pigment content, both treatments exhibited percentage decrease with respect to their initial values. The chlorophyll a content of the 3 days treatment showed a decrease of 11.189% and 6 days treatment showed a decrease of 11.6%. Likewise the percentage decrease of chlorophyll b content of the 3 days and 6 days treatment was noted to be 12.2% and 1.772% respectively. The total chlorophyll content of the 3 days treatment exhibited 7.5% reduction and in the case of 6 days treatment, a decrease of 18.73% was

noticed. In the case of carotenoid content, the 3 days treatment showed 17.32% and 6 days treatment exhibited 6.31 % of decrease.

While analyzing the biochemical parameters like lipid content, an increase of 60% was registered in 6 days and 50 % in 3 days of treatment. An increase of 14.86% of protein content was noticed in the 3 days treatment, while 6.58% of increase was noticed in the 6 days treatment. In the case of carbohydrate content, an increase of 28.57% was observed in 6 days of treatment, while 3 days of treatment exhibited a reduction of 66.67%.

			CO <sub>2</sub> supply in 80 bubbles / 2		CO <sub>2</sub> supply in 80 bubbles / 2	
Sl.No	Parameters Analysed	Units	hours of interval for 3 days		hours of interval for 6 days	
			Initial day	Final day	Initial day	Final day
1.	pH	-	7.75	5.91	7.94	6.30
2.	Conductivity	μs	859.3	822.2	854.7	800.4
3.	Resistivity	kΩ	1.278	1.336	1.285	1.351
4.	Dissolved Oxygen	mg/L	9.8	21.8	6.4	24.8
5.	Free Carbon dioxide	mg/L	44	475.2	35.2	352
6.	Alkalinity	mg/L	150	140	100	150
8.	Turbidity	NTU	58.7	77	49.3	103.8
9.	Specific growth	μd <sup>-1</sup>	0.1069		0.0346	
10.	<b>Biomass production</b>	g/L	1.847	2.948	1.655	3.043
11.	Cell Count	cells/ml	$0.122 \text{ x} 10^6$	$0.189 \text{ x} 10^6$	$0.117 \text{ x} 10^6$	$0.351 \text{ x} 10^6$
12.	Cell size	μm	56.25	48.75	56.25	52.5
13.	Chlorophyll <i>a</i>	mg/ g <sup>-1</sup>	7.436	6.604	6.31	5.579
14.	Chlorophyll <i>b</i>	mg/ g <sup>-1</sup>	2.851	2.503	2.653	2.606
15.	Total Chlorophyll	mg/ g <sup>-1</sup>	10.28	9.505	10.07	8.183
16.	Carotenoid	mg/ g <sup>-1</sup>	0.485	0.401	0.428	0.401
17.	Lipid	%	20	30	10	16
18.	Protein	mg/mg	0.148	0.17	0.076	0.081
19.	Carbohydrate	mg/mg	0.009	0.003	0.014	0.018
20.	Sodium	ppm	19.5	19	7	23
21.	Potassium	ppm	18.8	15.7	21	22.5

Table 3.3. Changes in media characteristics and morphological / biochemical responses of *Acutodesmus obliquus* to varying durations of carbon dioxide supply.

In the case of *A. obliquus* also, the gaseous release from the PBR was collected in the reservoir containing pure bolds basal medium. The culture medium contained in the reservoir was then subjected to the analysis of parameters like pH, free carbon dioxide and dissolved oxygen for assessing the magnitude and nature of gases released from the PBR. The results are depicted in table 3.4.

Upon comparing the pH values of the 3 days treatment, a decrease in pH was noticed in the final day (6.71) when compared to that of initial day (7.19). Similar trend was also observed in 6 days of treatment, where the pH of the final day (5.30) decreased with respect to initial day (7.07). In the case of free carbon dioxide content of the 3 days treatment, an increase in the final free carbon dioxide content (88 mg/L) was observed with respect to initial value (79.2 mg/L). Likewise in the 6 days treatment, a significant increase in the free carbon dioxide content was noticed in the final day (352 mg/L) when compared to that of initial day (48.4 mg/L). While comparing the dissolved oxygen content of 3 days treatment, a decrease was noticed in the final day (6.6 mg/L) with respect to the initial day (7.8 mg/L). In the case of 6 days treatment also a reduction in the dissolved oxygen content was observed in the final day (8.2 mg/L), when compared to the initial day (8.2 mg/L).

Table 3.4. Results on the analysis of Bolds basal medium contained in the outlet reservoir of PBR on treatment studies with *Acutodesmus obliquus*.

Sl.No	Parameters Analysed	CO <sub>2</sub> supply in 2 hours for an in	80 bubbles / terval of 3 days	CO <sub>2</sub> supply in 80 bubbles / 2 hours for an interval of 6 days	
		Initial day	Final day	Initial day	Final day
1.	pH	7.19	6.71	7.07	5.30
2.	Free Carbon dioxide (mg/L)	79.2	88	48.4	352
3.	Dissolved Oxygen(mg/L)	7.8	6.6	8.2	6.8

In the present study, carbon dioxide, at an optimum flow rate as elucidated in chapter 1, has been supplied at regular intervals of 2 hours to the PBR containing candidate species for facilitating increased gas retention within the system. There are several reports regarding the significance of intermittent sparging of carbon dioxide to overcome the growth inhibition on microalgae (Jiang et al., 2013; Radmann et al., 2011; Duarte et al., 2017). This strategic approach of intermittent sparging can not only minimize the acidic inhibition on microalgae, but also maximizes  $CO_2$  consumption efficiency (Jiang et al., 2013). Guo et al. (2015) also recognized that a constant supply of  $CO_2$  was not essential for the increased conversion of  $CO_2$  to biomass.

Upon consolidation of the results, it has been noticed that with both microalgal species, the pH of the medium after PBR studies were in acidic range, which normally happens due to the influx of  $CO_2$  (Chiu et al., 2008; Kumar et al., 2011; Naderi et al., 2015). It was also noticed in the present study that both microalgal species survived in acidic range of pH. This is indicative of their adaptabilities and better growth performances with respect to changing environmental conditions, thereby becoming ideal candidates for  $CO_2$  assimilation.

On an assessment of the DO values of treatments of both microalgal species, 3 days of treatment of *C. globosa* and 6 days of treatment of *A. obliquus* maintained higher levels of DO. With respect to  $CO_2$  consumption and biomass production, the DO content increases (Naderi et al., 2015). Also it is stated that an increase in DO leads to toxic effects like photo-bleaching and reduction in biomass production in controlled systems (Naderi et al., 2015). However in the present study no reduction in biomass production was noticed in the 6 days of treatment of *A. obliquus* with respect to increased DO content. Here the presence of increased DO content was an indication of their active photosynthetic process (Kumar et al., 2010) in presence of available  $CO_2$  and subsequent release of oxygen.

Estimation of biomass production and growth rate measurements was considered as vital tools in evaluating the bio-sequestration efficiencies of microalgae (Cheah et al., 2015). Upon comparing the results of major growth parameters of 3 days and 6 days of treatment, it was noticed that both microalgal species exhibited maximum turbidity, biomass productivity and cell count in 6 days of treatment. However, the results of 3 days treatment of both microalgae were higher than their respective control/initial values. On the other hand, while comparing the results of specific growth of both microalgal species, increased growth was noticed in the 3 days treatments. Increased growth rate of both microalgal species under intermittent supply of carbon dioxide than their control indicates them as promising candidates for  $CO_2$  biofixation.

Here, both 3 days and 6 days of treatment of *A. obliquus* and 3 days of treatment of *C. globosa* exhibited a decreasing trend in the pigment content with respect to their initial values. At the same time the 6 days of treatment set of *C. globosa* exhibited increased pigment production, compared to their initial values. However the results pertaining to the caroteinoid content of all treatment sets exhibited a decreasing trend. The variation in pigments can be attributed by variation in the metabolic responses of the microalgal members due to varying inputs in culture conditions. Similarly the higher CO<sub>2</sub> concentration in culture medium inhibited the growth of microalgae by disturbing the pigment contents (Gordillo et al., 1998). Jana et al. (2017) also observed a decline in chlorophyll *a* content under elevated CO<sub>2</sub> levels and commented that the presence of lower chlorophyll content may be due to the acclimatization of microalgae to higher CO<sub>2</sub> rich environments.

Upon comparison of the results of the biochemical components of *C*. *globosa*, an increase in lipid content and protein content was noticed in 3 days of treatment, while maximum carbohydrate content was noticed in 6 days of treatment. In the case of *A. obliquus*, the increased lipid content and carbohydrate, content was recorded in the 6 days of treatment and the maximum protein content was observed in 3 days of treatment.

While comparing the lipid content of both microalgal species, increased production was noticed in 3 days and 6 days of treatment of *A. obliquus*. Tang et al. (2011) also reported the efficiency of *Acutodesmus obliquus* SJTU-3 to accumulate increased lipid content under various  $CO_2$  concentrations, ranging from 0.03 to 50%. The ability of *A. obliquus* to accumulate high lipid content was considered as a significant factor in  $CO_2$  assimilation (Solovchenko and Khozin-Goldberg, 2013). Moreover, Fransico et al. (2010) commented that the lipid-rich species exhibit lower biomass productivity and established that the increased biomass productivity and lipid content were mutually exclusive. Here in the present study also lower biomass productivity was noticed with treatments of *A. obliquus* when compared to the treatments of *C. globosa*.

Higher carbohydrate content in both *C. globosa* and *A. obliquus* was recorded in 6 days of treatment. On the other hand, increased protein content was recorded in 3 days of treatment in both *C. globosa* and *A. obliquus*. Many researchers consider that elevated concentrations of  $CO_2$  stimulate the production of relevant proteins, which can influence cell physiology (Chen et al., 2013). Here in the treatments also increased protein content was recorded in 3 days in both microalgal members, in which increased free carbon dioxide content was noticed.

Throughout the treatment tenure of both microalgal species, the outlet gas from the PBR was trapped in a reservoir containing Bolds Basal medium (devoid of microalgae) and were subjected to analysis to have an idea of the gas trapped within it. The analytical results concerning media in the reservoir were compared with the results of the media contained in the PBR.

While comparing the results of the reservoir of *C. globosa*, it was noticed that during 3 days of treatment, the final pH increased (4.24%) and free carbon dioxide decreased (17.64%) with respect to their initial values. However there were no differences between the initial and final DO content (0%). It can be assumed that the supplied CO<sub>2</sub> concentration to the PBR (60 bubbles / 2 hours) might have been in an optimum range, leading to proper assimilation and a resultant influx of oxygen into the reservoir, contributing to an increase in pH to the alkalinity range. To support this, an increased DO accumulation (400%)in the culture medium contained in the PBR was also noticed. It can be assumed that excess of oxygen transfer to the reservoir might have contributed to an increase in pH. However, a small decrease in pH (4.78%) was noticed in the PBR, compared to their initial values.

In the case of 6 days treatment with *C. globosa*, the final pH decreased (0.59%) and free CO<sub>2</sub> content increased (21.42%) in the reservoir, compared to their initial values. However there were no differences between the initial and final DO content (0%). Here, the prolonged supply of carbon dioxide (60 bubbles / 2 hours) above the saturation limit of the culture medium contained in the PBR might have resulted in decrease in pH and increase in accumulation of carbon dioxide in the reservoir. To support this, a decrease in pH (9.84\%) to the acidic range was noticed in the in the culture medium contained in PBR during 6 days of treatment. Likewise, while comparing the DO content of PBR, a decrease was noticed in 6 days of treatment (386.11%) than 3 days.

Upon comparing the results of *A. obliquus* confining to the reservoir, in 3 days treatment, the final pH values decreased (6.68%), free carbon dioxide content increased (11%) and the DO content reduced (15.38%) with respect to

their initial values. The prolonged supply of carbon dioxide (80 bubbles / 2 hours) above the optimum levels of assimilation by the culture medium contained in PBR might have resulted in an influx of excess of carbon dioxide to the reservoir. A marked decrease in pH (23.7%), significant increase in free carbon dioxide content (980%) and reduced DO content (122.4%) was also noticed in the PBR containing *A. obliquus* pertaining to 3 days of treatment.

In the case of 6 days of treatment with *A. obliquus*, there observed a decrease in the final pH value (25.03%), significant increase in the free carbon dioxide content (627.27%) and reduction in DO content (17.07%) in the reservoir, with respect to the initial values. As stated above, the continuous supply of carbon dioxide (80 bubbles / 2 hours) to the PBR above the limit of assimilation might have attributed to such a super saturation level. To support this, a reduction in pH (20.65%), decrease in free carbon dioxide content (900%) and increased DO content (287.5%) was noticed in the PBR containing *A. obliquus* pertaining to 6 days of treatment.

#### **Statistical analysis**

The results pertaining to both treatment sets of *C. globosa* and *A. obliquus* were subjected to statistical analysis using t-test. Making use of this analysis, the variations between the t values of each influencing parameter concerning *C. globosa* and *A. obliquus* were worked out and are depicted in Tables 3.5 and 3.6, respectively.

Sl.No	Parameters Analysed	3 days treatment (t value)	6 days treatment (t value)
1.	Dissolved Oxygen	211.01	68.57
2.	Turbidity	58.5	114.4
3.	Biomass productivity	21.01	123.07
4.	Cell Count	33.95	14.38
5.	Chlorophyll a	48.12	75.63
6.	Chlorophyll b	159.49	149.49
7.	Total Chlorophyll	21.00	25.98
8.	Lipid	2.27	2.06
9.	Protein	0.82	1.009
10.	Carbohydrate	0.030	0.54

Table 3.5. Results of t-test on the responses of *Chlamydomonas globosa* to optimum dosage of carbon dioxide.

t for n-1(=2) degree of freedom at 5% level=9.93

While comparing the results of *C. globosa* at 5% level, t-values of the DO content of both treatments exhibited highly significant variation, while in the case of turbidity, significant variation was with 6 days of treatment (114.4). In the case of biomass productivity the t-values pertaining to 6 days of treatment (123.07) exhibited highly significant variation, whereas the cell count showed variations in 3 days of treatment (33.95).

As far as the t-values of chlorophyll *a* concerned, the significant variations was observed in 6 days treatment set (75.63). With chlorophyll *b*, both 3 days treatment (159.49) and 6 days treatment (149.49) exhibited highly significant variation. Similarly in the case of total chlorophyll, significant variation was observed in both 3 days (21.00) and 6 days of treatment (25.98).

Sl.No	Parameters Analysed	3 days treatment (t value)	6 days treatment (t value)
1.	Dissolved Oxygen	45.36	64.79
2.	Turbidity	40.36	99.50
3.	Biomass production	14.99	56.90
4.	Cell Count	19.70	26.96
5.	Chlorophyll a	4.89	18.89
6.	Chlorophyll b	2.99	2.86
7.	Total Chlorophyll	5.21	225.13
8.	Lipid	6.002	4.02
9.	Protein	2	1.80
10.	Carbohydrate	11.11	0.83

 Table 3.6. Results of t-test on the responses of Acutodesmus obliquus to optimum dosage of carbon dioxide.

t for n-1(=2) degree of freedom at 5% level=9.93

Upon comparing the results of *A. obliquus* at 5% level, t values of the DO content of both treatment sets showed variations in both 3 days (45.36) and 6 days (64.79) of treatment. Here in the case of turbidity (99.50) and biomass productivity (56.90), the 6 days treatment set exhibited highly significant variation. As far as the cell count concerned, the 6 days treatment set (26.96) also exhibited variation. With chlorophyll *a*, variation was observed in 6 days of treatment (18.89), while no variation was observed in both treatments set for chlorophyll *b*. In the case of total chlorophyll, highly significant variation was noticed in 6 days of treatment (225.134).

There were no significant variations in the biochemical parameters of both microalgal members in both the treatment sets that the t-values of the treatments of both microalgal species was noticed to be below 9.93, the degree of freedom at 5% level.

Upon comparing the results of major influencing parameters of *C. globosa*, increased biomass productivity, turbidity, cell count, pigment production and carbohydrate production were experienced in the 6 days of treatment. The enhanced production of dissolved oxygen, lipid content and protein content were recorded in 3 days of treatment. While analyzing the results pertaining to *A. obliquus* the enhanced dissolved oxygen content, biomass productivity, turbidity, cell count, lipid content and carbohydrate production were noticed in 6 days of treatment and the increased production of protein was registered in 3 days of treatment.

On an overall assessment of above stated results, it can be concluded that both microalgal species were found to be effective in sequestering carbon dioxide. Inspite of acidic pH and higher free carbon dioxide content, both microalgal species exhibited increased biomass productivity, turbidity, cell count in 6 days of treatments. This characteristically indicates their efficiencies in mitigating gaseous carbon dioxide in newly designed PBR, under controlled laboratory conditions. The 6 days of treatment of *C. globosa* are also highly promising with increased pigment and carbohydrate production. Moreover maximized lipid content and carbohydrate production were noticed in 6 days of treatment of *A. obliquus*. The capability of *A. obliquus* to accumulate lipid is considered as an important factor of  $CO_2$  tolerance (Solovchenko and Khozin-Goldberg, 2013), in which lipid biosynthesis represents a sink for the excess products of carbon fixation.

Various literatures suggest that *Chlorella* and *Scenedesmus* can be considered as promising candidates to fix  $CO_2$  and also for handling in adequate engineering systems. Several photobioreactor designs have been developed for the optimization of enhanced  $CO_2$  fixation under varying conditions. The results of research works of Chen et al. (2016) on *Acutodesmus obliquus*  claimed that the species can tolerate higher  $CO_2$  concentrations of 50% and the maximum biomass production was obtained at 20 %  $CO_2$ .

The efficiency of *A. obliquus* in assimilating external supply of  $CO_2$  in photobioreactor was already reported. However, there were no reports on sequestration efficiencies and photobioreactor designs for *C. globosa*. The present study highlights the excellent efficiencies of both *C. globosa* and *A. obliquus* in sequestering carbon dioxide. The  $CO_2$  fixation efficiencies of both microalgal species was maximized in the present laboratory scale closed vertical bubble column PBR and cultured using optimum  $CO_2$  concentration. The present venture constitutes an important step in the development of an environmentally sustainable solution/strategy to mitigate  $CO_2$  from small to medium scale point sources through a biological approach, using *C. globosa* and *A. obliquus*.

The presence of excess of  $CO_2$  and their resultant influxes in the reservoir concerning PBR of *C. globosa* and *A. obliquus* (except 3 days treatment with *C. globosa*) are indicative of an alteration in the design or possibility of coupling Photo Bio Reactors to accommodate / assimilate the excess quantities/unused share of  $CO_2$  generated within them.

## Summary and conclusion

In the present study, an attempt has been carried out to assess the efficiencies of *Acutodesmus obliquus* and *Chlamydomonas globosa* in carbon dioxide sequestration using a proto type Photo Bio Reactor (PBR), which can ensure optimum conditions of algal growth and carbon assimilation. Such type of standardization will help in the development of small to medium scale commercial installations which can effectively confiscate the released carbon dioxide using effective micro algal members. For fulfilling the target, a closed vertical bubble column structure having a size of 10 x 10 x 45 cm and a volume of 4.5 L was designed using acrylic material. An inlet, with control facility was maintained at the bottom for the supply of both culture medium and carbon dioxide gas. Likewise an outlet was maintained at the top of the structure to allow the gases within the column, if any, to come out. The outlet of the PBR was connected to the inlet of a reservoir of Bolds Basal medium kept in sealed bottle. The outlet of the PBR was kept at the bottom of the reservoir to permit the gases, if any, to bubble out through the medium. A controlled outlet was also provided to the reservoir for the release of gases, if any.

For experimentation, 3 litres of pure cultures of candidate species with an initial concentration of 0.5 OD, was transferred to the vertical column of the PBR. The carbon dioxide from the cylinder, having control facilities, was allowed to bubble at regular intervals through the culture medium contained in the column. The frequency of carbon dioxide supply was set for 60 or 80 bubbles/2 hours from 6 am to 6 pm. The PBR was retained at a light intensity of 100  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup> using LED light and fluorescent tubes. The room temperature was maintained at 28±0.5°C and humidity at 48%. The carbon sequestration efficiencies of both microalgal members were monitored for 3 and 6 days in separate trials. Moreover for analyzing the nature and magnitude of the gas generated at the top of the vertical closed column structure, the reservoir containing pure BB medium was monitored for gaseous influxes, specifically with respect to CO<sub>2</sub> and O<sub>2</sub>.

The physico-chemical attributes like pH, conductivity, resistivity, free carbon dioxide, dissolved oxygen, alkalinity and growth performances including turbidity, specific growth, biomass productivity, cell count, cell size were worked out on initial day and final day of the experimentation. The biochemical characteristics of micro algae including pigment contents, total lipid content, protein content, total carbohydrate and mineral composition were also analyzed before and after experimentation. The Bolds Basal medium contained in the reservoir was also subjected to the analysis of pH, free carbon dioxide and dissolved oxygen for determining the changes associated with them due to gaseous influx during experimentations in PBR.

While analyzing the results of major influencing parameters of *C. globosa*, increased biomass productivity, turbidity, cell count, pigment production and carbohydrate production were monitored in the 6 days treatment while enhanced dissolved oxygen content, lipid content and protein content were noticed in the 3 days treatment. Upon comparing the results pertaining to *A. obliquus*, the increased dissolved oxygen content, biomass productivity, turbidity, cell count, lipid content and carbohydrate production were observed in the 6 days treatment and the enhanced protein content was recorded in the 3 days treatment.

Here, inspite of acidic pH and higher free carbon dioxide content, both microalgal species exhibited increased biomass productivity, turbidity, cell count in 6 days of treatments which characteristically indicates their capabilities in mitigating gaseous carbon dioxide in newly designed PBR, under controlled laboratory conditions. The 6 days of treatment of *C. globosa* were also promising with increased pigment and carbohydrate production. Moreover maximized lipid content and carbohydrate production were noticed in 6 days of treatment of *A. obliquus*.

The present study highlights the capabilities of both *C. globosa* and *A. obliquus* in sequestering carbon dioxide. The  $CO_2$  fixation efficiencies of both microalgal species has maximized in the present laboratory scale, closed vertical bubble column PBR, while cultured using optimum  $CO_2$  concentration. The present venture constitutes an eco-friendly approach in developing a sustainable solution for the mitigation of  $CO_2$  emission from small to medium scale point sources by utilizing the carbon fixation efficiencies of *C. globosa* and *A. obliquus*.

Uncontrolled greenhouse gas emissions due to various anthropogenic activities have contributed substantially to global warming and climate change. The greenhouse gases mainly include Carbon dioxide ( $CO_2$ ), Methane ( $CH_4$ ) and oxides of Nitrogen (NOx). Of these,  $CO_2$  is of major concern today, owing to its higher concentration in the atmosphere.

Among various strategies for  $CO_2$  sequestration, biological sequestration using photosynthetic microalgae have received considerable attention in recent times. Microalgae, one of the most important living resources of both fresh and marine systems can be employed for  $CO_2$  sequestration, as they have higher photosynthetic efficiency, higher biomass production and faster growth rate, compared to other energy crops. They can easily be incorporated into engineered systems.

The present study is an attempt to assess the potentialities of indigenous freshwater microalgal species in carbon dioxide sequestration. The specific objectives outlined in the present study include:

- Maintenance of pure cultures of microalgal species using standard methods and selection of microalgal species which are active under culture conditions.
- Monitoring the responses of selected microalgal species under varying dosages of carbon dioxide supply.
- Determining the optimum pH favouring maximized growth of selected microalgal members.
• Assessment of the CO<sub>2</sub>assimilation efficiency of selected microalgal members in PBR under controlled conditions.

For a meaningful elucidation of the objectives, the study has been undertaken in three stages and their outcomes are depicted in three chapters. They include:

Chapter I. Screening of microalgae for CO<sub>2</sub> assimilation efficiency

Chapter II. pH specific modification of culture medium for growth maximization of selected micro algal species

Chapter III. Photo Bio Reactor based feasibility studies on the carbon sequestration efficiency of selected micro algal members

## Chapter I

The study has been carried out with the objective of assessing the  $CO_2$  assimilation capabilities of five indigenous microalgal species belonging to Chlorophyceae, namely *Chlamydomonas grovei*, *Chlamydomonas globosa*, *Desmodesmus opliensis*, *Monoraphidium contortum* and *Acutodesmus obliquus*. These micro algal samples were collected from heterogeneous environments and their pure cultures were maintained in the laboratory using Bolds Basal medium. They were then individually subjected to  $CO_2$  assimilation studies.

For assimilation studies, 12 litres of Bolds Basal medium was prepared and to this, four litres of culture medium containing pure cultures of the respective microalgal species was added. After incubation, the microalgal culture was transferred to 16 one litre conical flasks and then separated into three sets of five conical flasks. The first set of five flasks was treated as control and was maintained as such. To the culture containing conical flasks of the second set, ambient air has been bubbled at an interval of two hours and treated as aerated set. To the third set, carbon dioxide from a cylinder has been bubbled at an interval of two hours and was considered as  $CO_2$  treated set. The culture contained in the last conical flask (16<sup>th</sup>) was treated as the initial control set and has been used to analyze all parameters meant for the initial day of treatment.

The experimentation was initiated at 6 am on the initial day and all the three sets were kept at illumination during the day time (6 am to 6 pm). Sampling and analyzing of culture has been carried out at 6 am of each day. pH, conductivity, resistivity, dissolved oxygen, free carbon dioxide and alkalinity content of the algal culture together with cell size (micrometry), turbidity, cell count and biomass content of the micro algal members were worked out. Monitoring of the cultures was carried out for a period of 120 hours.

For monitoring the carbon dioxide sequestration potentials of each microalgal member, the results pertaining to major growth parameters like turbidity of the medium owing to micro algal growth, together with cell count and biomass content of micro algal members were considered. While comparing the carbon dioxide assimilation efficiencies of five microalgal members under study, in the case of *C. grovei*, higher turbidity, cell count and biomass content was noticed in the CO<sub>2</sub> treatment set of 10 bubbles / 2 hours. With *C. globosa*, increased turbidity, cell count and biomass content was observed in CO<sub>2</sub> treatment set of 60 bubbles / 2 hours. In the case of *D. opoliensis*, higher turbidity, cell count and biomass was noticed in the CO<sub>2</sub> treatment set of 20 bubbles / 2 hours. Concerning *M. contortum*, maximum turbidity, cell count and biomass was noticed in the CO<sub>2</sub> treatment set of 15 bubbles / 2 hours, while with *A. obliquus*, increased turbidity, cell count and biomass content was observed in CO<sub>2</sub> treatment set of 80 bubbles / 2 hours.

Results of the present study indicated that all the micro algal members under study exhibited varied ranges of tolerance to CO<sub>2</sub> supply. The results of *A. obliquus* (80 bubbles/2 hours) and *C. globosa* (60 bubbles/2 hours) are highly promising which exhibited, higher turbidity, cell count and biomass production in CO<sub>2</sub> treatment sets. The species *C. gorvei* (10bubbles/2 hours), *M. contortum* (15 bubbles/2 hours) and *D. opoliensis* (20bubbles/2 hours) also exhibited moderate efficiencies in carbon dioxide assimilation.

## **Chapter II**

In recent times, microalgae have been employed for a wide range of beneficial purposes. For exploiting their potentials for such purposes, their biomass is required in sufficient quantities.

One among the major constraints in their utilization for such purposes is that most of the members, which are most abundant and virulent in natural conditions, seem to be slow growing in culture conditions. Moreover, the risk of maintaining cultures without contamination is another constraint in their utilization for multidimensional purposes. Hence, standardization of species specific culture conditions is a prerequisite for meeting the required targets.

pH is noted to be an important factor influencing the growth and multiplication of micro algal species in culture conditions. pH of the medium is also significant in carbon sequestration studies, as it determines the availability and solubility of  $CO_2$  and other nutrients. In the present study, an attempt has been carried out to assess the growth performances of selected microalgal members (*C. globosa* and *A. obliquus*) in Bolds Basal medium under varying culture conditions, altered by pH.

For experimentation, treatment sets were maintained with conical flasks of 100 ml capacity (21 nos), each containing 50 ml Bolds Basal medium. The pH of culture medium was adjusted from 3-12, with a gradation of 0.5 using

0.5 N NaOH and 0.05 N HCl. After adjusting to the required pH, 5 ml of pure cultures of the respective microalgal members were added to the respective conical flasks and the changes in pH, if any, was readjusted. The control set was maintained at pH 6.6 (original pH of the culture medium).

The culture conditions like pH, temperature, conductivity and resistivity and microalgal growth parameters like turbidity, cell count, cell size, and biomass were monitored throughout the treatment period. The biomass estimation was carried out on the initial and final days of treatment. All the sets were kept at illumination during daytime (6 am – 6 pm) and retained under light intensity of 40  $\mu$ mol m<sup>2</sup> s<sup>-1</sup>. The temperature of the medium during experimentation ranged from 26-29.8°C while the humidity ranged from 48 - 54%. Every day after observation, the altered pH was readjusted to their experimental condition. Monitoring of the treatment sets were carried out for a period of 7 days.

For the confirmation of appropriate pH level at which maximum growth of both microalgal members under the study had occurred, the median values of the 7 days triplicate data were worked out and compared. However, during experimentation, in higher alkaline ranges, formation of precipitate was noticed. Hence, for drawing conclusions regarding the optimum pH range in which maximum turbidity occurred, the final median values of pH ranges above 9 were neglected.

Upon comparing the final median values of major growth parameters pertaining to the treatment sets containing *C. globosa*, the maximum cell count was recorded at pH 8, whereas increased turbidity and biomass content were noticed in pH 9. While analyzing the final median values of major growth parameters pertaining to *A. obliquus*, increased turbidity and biomass content were noticed in pH 9, while maximum cell count was observed in pH 5.

The present findings thus confirms that the optimum range of pH favouring the growth of *C. globosa* in BB medium is 8-9 and that of *A. obliquus* is 5-9. Hence for ensuring better biomass production of these microalgal members in BB medium, the above mentioned pH ranges can be followed.

## Chapter III

In this chapter, an attempt has been carried out to assess the efficiencies of *A*. *obliquus* and *C. globosa* in carbon dioxide sequestration using a proto type Photo Bio Reactor (PBR), which can ensure optimum conditions of algal growth and carbon assimilation. Such type of standardization will help in the development of small to medium scale commercial installations which can effectively confiscate the released carbon dioxide using effective micro algal members.

For fulfilling the target, a closed vertical bubble column structure having a size of 10 x 10 x 45 cm and a volume of 4.5 L was designed using acrylic material. An inlet, with control facility was maintained at the bottom for the supply of both culture medium and carbon dioxide gas. Likewise an outlet was maintained at the top of the structure to allow the gases within the column, if any, to come out. The outlet of the PBR was connected to the inlet of a reservoir of Bolds Basal medium kept in sealed bottle. The outlet of the PBR was kept at the bottom of the reservoir to permit the gases, if any, to bubble out through the medium. A controlled outlet was also provided to the reservoir for the release of gases, if any.

For experimentation, 3 litres of pure cultures of candidate species with an initial concentration of 0.5 OD, was transferred to the vertical column of the PBR. The carbon dioxide from the cylinder, having control facilities, was allowed to bubble at regular intervals through the culture medium contained in the column. The frequency of carbon dioxide supply was set for 60 or 80

bubbles /2 hours from 6 am to 6 pm. The PBR was retained at a light intensity of 100  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup> using LED light and fluorescent tubes. The room temperature was maintained at 28±0.5°C and humidity at 48%. The carbon sequestration efficiencies of both microalgal members were monitored for 3 and 6 days in separate trials. Moreover for analyzing the nature and magnitude of the gas generated at the top of the vertical closed column structure, the reservoir containing pure BB medium was monitored for gaseous influxes, specifically with respect to CO<sub>2</sub> and O<sub>2</sub>.

The physico-chemical attributes like pH, conductivity, resistivity, free carbon dioxide, dissolved oxygen, alkalinity and growth performances including turbidity, specific growth, biomass productivity, cell count, cell size were worked out on initial day and final day of the experimentation. The biochemical characteristics of microalgae including pigment contents, total lipid content, protein content, total carbohydrate and mineral composition were also analyzed before and after experimentation. The bolds basal medium contained in the reservoir was also subjected to the analysis of pH, free carbon dioxide and dissolved oxygen for determining the changes associated with them due to gaseous influx during experimentations in PBR.

While analyzing the results of major influencing parameters of *C. globosa*, increased biomass productivity, turbidity, cell count, pigment production and carbohydrate production were monitored in the 6 days treatment while enhanced dissolved oxygen content, lipid content and protein content were noticed in the 3 days treatment. Upon comparing the results pertaining to *A. obliquus*, the increased dissolved oxygen content, biomass productivity, turbidity, cell count, lipid content and carbohydrate production were observed in the 6 days treatment and the enhanced protein content was recorded in the 3 days treatment.

Here, inspite of acidic pH and higher free carbon dioxide content, both microalgal species exhibited increased biomass productivity, turbidity, cell count in 6 days of treatments which characteristically indicates their capabilities in mitigating gaseous carbon dioxide in newly designed PBR, under controlled laboratory conditions. The 6 days of treatment of *C. globosa* were also promising with increased pigment and carbohydrate production. Moreover maximized lipid content and carbohydrate production were noticed in 6 days of treatment of *A. obliquus*.

The present study highlights the capabilities of both *C. globosa* and *A. obliquus* in sequestering carbon dioxide. The  $CO_2$  fixation efficiencies of both microalgal species has maximized in the present laboratory scale, closed vertical bubble column PBR, while cultured using optimum  $CO_2$  concentration. The present venture constitutes an eco-friendly approach in developing a sustainable solution for the mitigation of  $CO_2$  emission from small to medium scale point sources by utilizing the carbon fixation efficiencies of *C. globosa* and *A. obliquus*.

- Abalde, J., Fabregas, J. and Herrero, C. 1991. β-Carotene, vitamin C and vitamin E content of the marine microalga *Dunaliella tertiolecta* cultured with different nitrogen sources. *Bioresource Technology*, 38(2-3), pp.121-125.
- Abe, K., Nishimura, N. and Hirano, M. 1999. Simultaneous production of βcarotene, vitamin E and vitamin C by the aerial microalga *Trentepohlia aurea*. *Journal of Applied Phycology*, 11(4), pp.331-336.
- Abu-Khader, M.M. 2006. Recent progress in CO<sub>2</sub> capture/sequestration: A review. *Energy Sources, Part A*, 28(14), pp.1261-1279.
- Abu-Rezq, T.S., Al-Musallam, L., Al-Shimmari, J. and Dias, P. 1999. Optimum production conditions for different high-quality marine algae. *Hydrobiologia*, 403, pp.97-107.
- Acién, F.G., Sevilla, J.M.F and Grima, E M. 2013. Photobioreactors for the production of microalgae, *Reviews in Environmental Science and Biotechnology*, 12, pp. 131–151.
- Agoramoorthy, G., Chandrasekaran, M., Venkatesalu, V. and Hsu, M.J. 2007. Antibacterial and antifungal activities of fatty acid methyl esters of the blind-your-eye mangrove from India. *Brazilian Journal of Microbiology*, 38(4), pp.739-742.
- Al-Qasmi, M., Raut, N., Talebi, S., Al-Rajhi, S. and Al-Barwani, T. 2012. A review of effect of light on microalgae growth. *Proceedings of the World Congress on Engineering*, 1, pp.4-6.

- Al-Safaar, A.T., Al-Rubiaee, G.H. and Salman, S.K. 2016. Effect of pH condition on the growth and lipid content of microalgae *Chlorella vulgaris* and *Chroococcus minor. Journal of Scientific and Engineering Research*, 7(11), pp.1139.
- Al-Shatri, A.H.A., Ali, E., Al-Shorgani, N.K.N. and Kalil, M.S. 2014. Growth of *Scenedesmus dimorphus* in different algal media and pH profile due to secreted metabolites. *African Journal of Biotechnology*, 13(16), pp.1714-1720.
- Anjos, M., Fernandes, B.D., Vicente, A.A., Teixeira, J.A. and Dragone, G.
   2013. Optimization of CO<sub>2</sub> bio-mitigation by *Chlorella* vulgaris. Bioresource Technology, 139, pp.149-154.
- Armstrong, J.E. and Calder, J.A. 1978. Inhibition of light-induced pH increase and O<sub>2</sub> evolution of marine microalgae by water-soluble components of crude and refined oils. *Applied and Environmental Microbiology*, 35(5), pp.858-862.
- Azov, Y. 1982. Effect of pH on inorganic carbon uptake in algal cultures. Applied and Environmental Microbiology, 43(6), pp.1300-1306.
- Baba, M., Suzuki, I. and Shiraiwa, Y. 2011. Proteomic analysis of high-CO<sub>2</sub>inducible extracellular proteins in the unicellular green alga, *Chlamydomonas reinhardtii. Plant and Cell Physiology*, 52(8), pp.1302-1314.
- Bakuei, N., Amini, G., Najafpour, G.D., Jahanshahi, M. and Mohammadi, M. 2015. Optimal cultivation of *Scenedesmus* sp. microalgae in a bubble column photobioreactor. *Indian Journal of Chemical Technology*, 22, pp.20-25.

- Bartley, M.L., Boeing, W.J., Dungan, B.N., Holguin, F.O. and Schaub, T. 2014. pH effects on growth and lipid accumulation of the biofuel microalgae *Nannochloropsis salina* and invading organisms. *Journal* of Applied Phycology, 26(3), pp.1431-1437.
- Basu, S., Roy, A.S., Mohanty, K. and Ghoshal, A.K. 2013. Enhanced CO<sub>2</sub> sequestration by a novel microalga: *Scenedesmus obliquus* SA1 isolated from bio-diversity hotspot region of Assam, India. *Bioresource Technology*, 143, pp. 369-377.
- Bayless, D.J., Kremer, G.G., Prudich, M.E., Stuart, B.J., Vis-Chiasson, M.L., Cooksey, K. and Muhs, J. 2001. Enhanced practical photosynthetic CO<sub>2</sub> mitigation. Proceedings of the First National conference on Carbon sequestration.5A4: 1-14.
- Becker, E.W. and Venkataraman, L.V. 1980. Production and processing of algae in pilot plant scale: Experiences of the Indo-German project. In:Shelef, G. and Soeder, C.J. (Eds.), *Algae Biomass, Production and Use*, Elsevier, pp.35-50.
- Becker, E.W. 1994. Microalgae Biotechnology and Microbiology. Cambridge University Press. Cambridge, p.293.
- Beklioglu, M. and Moss, B. 1995. The impact of pH on interactions among phytoplankton algae, zooplankton and perch (*Perca fluviatilis*) in a shallow, fertile lake. *Freshwater Biology*, 33(3), pp.497-509.
- Benemann, J.R., 1993. Utilization of carbon dioxide from fossil fuel-burning power plants with biological systems. *Energy conversion and management*, 34(9-11), pp.999-1004.
- Beyhan, O., Aktas, M., Yilmaz, N., Simsek, N. and Gercekcioglu, R. 2011. Determination of fatty acid compositions of some important almond

(*Prunus amygdalus* L.) varieties selected from Tokat province and Eagean region of Turkey. *Journal of Medicinal Plants Research*, 5(19), pp.4907-4911.

- Bilanovic, D., Andargatchew, A., Kroeger, T. and Shelef, G. 2009. Freshwater and marine microalgae sequestering of CO<sub>2</sub> at different C and N concentrations–Response surface methodology analysis. *Energy Conversion and Management*, 50(2), pp. 262-267.
- Binnal, P. and Babu, P.N. 2017. Statistical optimization of parameters affecting lipid productivity of microalga *Chlorella protothecoides* cultivated in photobioreactor under nitrogen starvation. *South African Journal of Chemical Engineering*, 23, pp. 26-37.
- Bischoff, H.C., 1963. Some soil algae from Enchanted Rock and related algal species. Phycological Studies IV. University of Texas Publication, pp.1-95.
- Bitog, J.P., Lee, I.B., Lee, C.G., Kim, K.S., Hwang, H.S., Hong, S.W., Seo, I.H., Kwon, K.S. and Mostafa, E. 2011. Application of computational fluid dynamics for modeling and designing photobioreactors for microalgae production: a review. *Computers and Electronics in Agriculture*, 76(2), pp.131-147.
- Bligh, E.G. and Dyer, W.J. 1959. A rapid method of total lipid extraction and purification. *Canadian Journal of Biochemistry and Physiology*, 37(8), pp.911-917.
- Bold, H.C. 1949. The morphology of *Chlamydomonas chlamydogama*, sp. nov. *Bulletin of the Torrey Botanical Club*, 72(2), pp.101-108.
- Bonenfant, D., Mimeault, M. and Hausler, R. 2003. Determination of the structural features of distinct amines important for the absorption of

CO<sub>2</sub> and regeneration in aqueous solution. *Industrial and Engineering Chemistry Research*, *42*(14), pp. 3179-3184.

- Borowitzka, M.A. and Borowitzka, L.J. 1988. Microalgal Biotechnology. Cambridge University Press, Cambridge, p.466.
- Brennan, L. and Owende, P. 2010. Biofuels from microalgae-A review of technologies for production, processing, and extractions of biofuels and co-products. *Renewable and Sustainable Energy Reviews*, 14(2), pp.557-577.
- Brown, L.M. 1996. Uptake of Carbon dioxide from flue gas by microalgae. *Energy Conversion and Management*, *37*(6-8), pp.1363-1367.
- Cape Grim Greenhouse Gas Data. https://www.csiro.au/en/Research/ OandA/Areas/Assessing-our-climate/Greenhouse-gas-data.
- Celekli, A. and Dönmez, G. 2006. Effect of pH, light intensity, salt and nitrogen concentrations on growth and β-carotene accumulation by a new isolate of *Dunaliella* sp. *World Journal of Microbiology and Biotechnology*, *22*(2), pp.183-189.
- Chae, S.R., Hwang, E.J. and Shin, H.S. 2006. Single cell protein production of *Euglena gracilis* and Carbon dioxide fixation in an innovative photo-bioreactor.*Bioresource Technology*, 97(2), pp.322-329.
- Chang, E.H. and Yang, S.S. 2003. Some characteristics of microalgae isolated in Taiwan for biofixation of Carbon dioxide. *Botanical Bulletin of Academia Sinica*, 44, pp.43-52.

- Cheah, W.Y., Show, P.L., Chang, J.S., Ling, T.C. and Juan, J.C. 2015. Biosequestration of atmospheric CO<sub>2</sub> and flue gas-containing CO<sub>2</sub> by microalgae. *Bioresource Technology*, 184, pp.190-201.
- Chen, C.Y. and Durbin, E.G. 1994. Effects of pH on the growth and carbon uptake of marine phytoplankton. *Marine Ecology Progress Series*,109, pp.83-94.
- Chen, C.Y., Yeh, K.L., Aisyah, R., Lee, D.J. and Chang, J.S. 2011. Cultivation, photobioreactor design and harvesting of microalgae for biodiesel production: A critical review. *Bioresource Technology*, 102(1), pp.71-81.
- Chen, C.Y., Yeh, K.L., Su, H.M., Lo, Y.C., Chen, W.M. and Chang, J.S. 2010. Strategies to enhance cell growth and achieve high-level oil production of a *Chlorella vulgaris* isolate. *Biotechnology Progress*, 26(3), pp.679-686.
- Chen, C.Y., Zhao, X.Q., Yen, H.W., Ho, S.H., Cheng, C.L., Lee, D.J., Bai, F.W. and Chang, J.S. 2013. Microalgae-based carbohydrates for biofuel production *Biochemical Engineering Journal*, 78, pp.1-10.
- Chen, L., Li, X., Wang, M., Huang, J., Chen, B., Li, X. and Zheng, X. 2016. Utilization of gaseous emissions from a methanol plant for cultivation of *Acutodesmus obliquus*. *Journal of applied phycology*, 28(5), pp.2661-2669.
- Cheng, J., Huang, Y., Feng, J., Sun, J., Zhou, J. and Cen, K. 2013. Improving CO<sub>2</sub> fixation efficiency by optimizing *Chlorella* PY-ZU1 culture conditions in sequential bioreactors. *Bioresource Technology*, 144, pp.321-327.

- Cheng, L., Zhang, L., Chen, H. and Gao, C. 2006. Carbon dioxide removal from air by microalgae cultured in a membranephotobioreactor. *Separation and Purification Technology*, 50(3), pp.324-329.
- Chinnasamy, S., Ramakrishnan, B., Bhatnagar, A. and Das, K.C. 2009.
   Biomass production potential of a wastewater alga *Chlorella vulgaris* ARC 1 under elevated levels of CO<sub>2</sub> and temperature. *International Journal of Molecular Sciences*, 10(2), pp.518-532.
- Chiranjeevi, P. and Venkata Mohan, S. 2016. Optimizing the critical factors for lipid productivity during stress phased heterotrophic microalgae cultivation. *Frontiers in Energy Research*, *4*, pp.1-10.
- Chisti, Y. 2007. Biodiesel from microalgae. *Biotechnology Advances, 25*, pp.294-306.
- Chiu, S.Y., Kao, C.Y., Chen, C.H., Kuan, T.C., Ong, S.C. and Lin, C.S. 2008.
  Reduction of CO<sub>2</sub> by a high-density culture of *Chlorella* sp. in a semicontinuous photobioreactor. *Bioresource Technology*, 99(9), pp.3389-3396.
- Chiu, S.Y., Kao, C.Y., Huang, T.T., Lin, C.J., Ong, S.C., Chen, C.D., Chang, J.S. and Lin, C.S. 2011. Microalgal biomass production and on-site bioremediation of Carbon dioxide, Nitrogen oxide and Sulfur dioxide from flue gas using *Chlorella* sp. cultures. *Bioresource Technology*, 102(19), pp.9135-9142.
- Chiu, S.Y., Kao, C.Y., Tsai, M.T., Ong, S.C., Chen, C.H. and Lin, C.S. 2009.
  Lipid accumulation and CO<sub>2</sub> utilization of *Nannochloropsis oculata* in response to CO<sub>2</sub> aeration. *Bioresource Technology*, 100(2), pp.833-838.

- Clement, G. 1975. Production and characteristic constituents of the algae *Spirulina platensis* and *maxima*. *Annales de la Nutrition et de L'alimentation*, 29, pp.477-488.
- Das, P., Aziz, S.S. and Obbard, J.P. 2011. Two phase microalgae growth in the open system for enhanced lipid productivity. *Renewable Energy*, 36(9), pp.2524-2528.
- Davis, M.J., Coote, P.J. and O'Byrne, C.P. 1996. Acid tolerance in *Listeria monocytogenes*: the adaptive acid tolerance response (ATR) and growth-phase-dependent acid resistance. *Microbiology*, *142*(10), pp.2975-2982.
- Del Campo, J.A., Moreno, J., Rodríguez, H., Vargas, M.A., Rivas, J. and Guerrero, M.G. 2000. Carotenoid content of chlorophycean microalgae: Factors determining lutein accumulation in *Muriellopsis* sp.(Chlorophyta). *Journal of Biotechnology*, 76(1), pp.51-59.
- Demidov, E. 2000. Short-term responses of photosynthetic reactions to extremely high-CO<sub>2</sub> stress in a 'High-CO<sub>2</sub>' tolerant green alga, *Chlorococcum littorale*, and an intolerant green alga *Stichococcus bacillaris*. *Russian Journal of Plant Physiology*, *47*, pp.622-631.
- de Morais, M.G. and Costa, J.A.V. 2007a. Isolation and selection of microalgae from coal fired thermoelectric power plant for biofixation of carbon dioxide. *Energy Conversion and Management*, 48(7), pp.2169-2173.
- de Morais, M.G. and Costa, J.A.V. 2007 b. Carbon dioxide fixation by *Chlorella kessleri*, *C. vulgaris*, *Scenedesmus obliquus* and *Spirulina* sp. cultivated in flasks and vertical tubular photobioreactors. *Biotechnology Letters*, 29(9), pp.1349-1352.

- Devgoswami, C.R., Kalita, M.C., Talukdar, J., Bora, R. and Sharma, P., 2011. Studies on the growth behavior of *Chlorella*, *Haematococcus* and *Scenedesmus* sp. in culture media with different concentrations of Sodium bicarbonate and Carbon dioxide gas. *African Journal of Biotechnology*, 10(61), pp.13128-13138.
- Dineshkumar, R., Narendran, R. and Sampathkumar, P. 2017. Cultivation and harvesting of micro-algae for bio-fuel production-A review. *Indian Journal of Geo Marine Sciences*, *46*(9), pp.1731-1742.
- Doan, T.T.Y., Sivaloganathan, B. and Obbard, J.P. 2011. Screening of marine microalgae for biodiesel feedstock. *Biomass and Bioenergy*, 35(7), pp.2534-2544.
- Dote, Y., Sawayama, S., Inoue, S., Minowa, T. and Yokoyama, S.Y. 1994. Recovery of liquid fuel from hydrocarbon-rich microalgae by thermochemical liquefaction. *Fuel*, *73*(12), pp.1855-1857.
- Doucha, J., Straka, F. and Lívanský, K. 2005. Utilization of flue gas for cultivation of microalgae *Chlorella* sp. in an outdoor open thin-layer photobioreactor. *Journal of Applied Phycology*, 17(5), pp.403-412.
- Douskova, I., Doucha, J., Livansky, K., Machat, J., Novak, P., Umysova, D., Zachleder, V. and Vitova, M. 2009. Simultaneous flue gas bioremediation and reduction of microalgal biomass production costs. *Applied microbiology and Biotechnology*,82(1), pp.179-185.
- Duarte, J.H., de Morais, E.G., Radmann, E.M. and Costa, J.A.V. 2017. Biological CO<sub>2</sub> mitigation from coal power plant by *Chlorella fusca* and *Spirulina* sp. *Bioresource Technology*, *234*, pp.472-475.
- Dubinsky, Z. and Rotem, J.M.1974. Relations between algal populations and the pH of their media. *Oecologia*, *16*(1), pp.53-60.

- Dubois, M., Gilles, K.A., Hamilton, J.K., Rebers, P.T. and Smith, F. 1956. Colorimetric method for determination of sugars and related substances. *Analytical chemistry*, 28(3), pp.350-356.
- Ebi, K.L., Mearns, L.O. and Nyenzi, B. 2003. Weather and climate: changing human exposures. Climate Change and Health: Risks and Responses (McMichael AJ, Campbell-Lendrum DH, Corvalan CF, Ebi KL, Githeko A, et al., eds). Geneva: World Health Organization.
- Eloka-Eboka, A.C. and Inambao, F.L. 2017. Effects of CO<sub>2</sub> sequestration on lipid and biomass productivity in microalgal biomass production. *Applied Energy*, *195*, pp.1100-1111.
- Eriksen, N.T. 2008. The technology of microalgal culturing. *Biotechnology Letters*, *30*(9), pp.1525-1536.
- Fan, L.H., Zhang, Y.T., Zhang, L. and Chen, H.L. 2008. Evaluation of a membrane-sparged helical tubular photobioreactor for Carbon dioxide biofixation by *Chlorella vulgaris*. Journal of Membrane Science, 325(1), pp.336-345.
- Faria, G.R., Caroline, R.P.S., Paes, D.J.F.A., Castro, N.A.B., Tinoco, E.B. and Sergio, O.L. 2012. Effects of the availability of CO<sub>2</sub> on growth, nutrient uptake, and chemical composition of the marine microalgae *Chlorella* sp. and *Nannochloropsis oculata*, two potentially useful strains for biofuel production. *International Research Journal of Biotechnology*, 5, pp.65-75.
- Francisco, E.C., Neves, D.B., Jacob-Lopes, E. and Franco, T.T. 2010. Microalgae as feedstock for biodiesel production: Carbon dioxide sequestration, lipid production and biofuel quality. *Journal of Chemical Technology and Biotechnology*, 85(3), pp.395-403.

- Fuggi, A., Pinto, G., Pollio, A. and Taddei, R. 1988. The role of glycerol in osmoregulation of the acidophilic alga *Dunaliella acidophila* (Volvocales, Chlorophyta): Effect of solute stress on photosynthesis, respiration and glycerol synthesis. *Phycologia*, 27(4), pp.439-446.
- Fulke, A.B., Krishnamurthi, K., Giripunje, M.D., Devi, S.S. and Chakrabarti, T. 2015. Biosequestration of Carbon dioxide, biomass, calorific value and biodiesel precursors production using a novel flask culture photobioreactor. *Biomass and Bioenergy*, 72, pp.136-142.
- Fulke, A.B., Mudliar, S.N., Yadav, R., Shekh, A., Srinivasan, N., Ramanan, R., Krishnamurthi, K., Devi, S.S. and Chakrabarti, T. 2010. Biomitigation of CO<sub>2</sub>, calcite formation and simultaneous biodiesel precursors production using *Chlorella* sp. *Bioresource Technology*, 101(21), pp.8473-8476.
- García, M.C., Sevilla, J.F., Fernández, F.A., Grima, E.M. and Camacho, F.G. 2000. Mixotrophic growth of *Phaeodactylum tricornutum* on glycerol: Growth rate and fatty acid profile. *Journal of Applied Phycology*, *12*(3-5), pp.239-248.
- Gardner, R., Peters, P., Peyton, B. and Cooksey, K.E. 2011. Medium pH and nitrate concentration effects on accumulation of triacylglycerol in two members of the Chlorophyta. *Journal of Applied Phycology*, 23(6), pp.1005-1016.
- Gehl, K.A., Cook, C.M. and Colman, B. 1987. The effect of external pH on the apparent CO<sub>2</sub> affinity of *Chlorella saccharophila*. *Journal of Experimental Botany*, 38(7), pp.1203-1210.
- Gensemer, R.W., Smith, R.E. and Duthie, H.C. 1993. Comparative effects of pH and aluminum on silica-limited growth and nutrient uptake in

Asterionella ralfsii var. americana (Bacillariophyceae). Journal of Phycology, 29(1), pp.36-44.

- Gerloff-Elias, A.N.T.J.E., Spijkerman, E. and Pröschold, T. 2005. Effect of external pH on the growth, photosynthesis and photosynthetic electron transport of *Chlamydomonas acidophila* Negoro, isolated from an extremely acidic lake (pH 2.6). *Plant, Cell and Environment, 28*(10), pp.1218-1229.
- Giordano, M., Beardall, J. and Raven, J.A. 2005. CO<sub>2</sub> concentrating mechanisms in algae: Mechanisms, environmental modulation, and evolution. *Annual Review Plant Biology*, 56, pp. 99-131.
- Goldman, J.C. 1979. Outdoor algal mass cultures-II. Photosynthetic yield limitations. *Water Research*, 13(2), pp.119-136.
- Goldman, J.C. 1980. Physiological aspects in algal mass cultures. In: Shelef,G. and Soeder, C.J. (Eds.), Algae Biomass, Production and Use.Elsevier, pp.343-359.
- Goldman, J.C. and Ryther, J.H. 1976. Temperature-influenced species competition in mass cultures of marine phytoplankton. *Biotechnology and Bioengineering*, *18*(8), pp.1125-1144.
- Gong, Q., Feng, Y., Kang, L., Luo, M. and Yang, J. 2014. Effects of light and pH on cell density of *Chlorella vulgaris*. *Energy Procedia*, 61, pp.2012-2015.
- Gordillo, F.J., Jiménez, C., Figueroa, F.L. and Niell, F.X. 1998. Effects of increased atmospheric CO<sub>2</sub> and N supply on photosynthesis, growth and cell composition of the cyanobacterium *Spirulina platensis* (Arthrospira). *Journal of Applied Phycology*, 10(5), pp.461-469.

- Gross, W. 2000. Ecophysiology of algae living in highly acidic environments. *Hydrobiologia*, 433(1-3), pp.31-37.
- Guckert, J.B. and Cooksey, K.E. 1990. Triglyceride accumulation and fatty acid profile changes in *Chlorella* (Chlorophyta) during high pH induced cell cycle inhibition 1. *Journal of Phycology*, *26*(1), pp.72-79.
- Guedes, A.C., Amaro, H.M. and Malcata, F.X. 2011. Microalgae as sources of carotenoids. *Marine Drugs*, *9*(4), pp.625-644.
- Guillard, R.R. and Ryther, J.H. 1962. Studies of marine planktonic diatoms: I.
   Cyclotella nana Hustedt, and Detonula confervacea (Cleve)
   Gran. *Canadian journal of microbiology*, 8(2), pp.229-239.
- Guo, Z., Phooi, W.B.A., Lim, Z.J. and Tong, Y.W. 2015. Control of CO<sub>2</sub> input conditions during outdoor culture of *Chlorella vulgaris* in bubble column photobioreactors. *Bioresource Technology*, 186, pp.238-245.
- Gupta, H. and Fan, L.S. 2002. Carbonation-calcination cycle using high reactivity Calcium oxide for Carbon dioxide separation from flue gas. *Industrial and Engineering Chemistry Research*, 41(16), pp.4035-4042.
- Gupta, P.L., Lee, S.M. and Choi, H.J. 2015. A mini review: Photobioreactors for large scale algal cultivation. World Journal of Microbiology and Biotechnology, 31(9), pp.1409-1417.
- Halter, D., Casiot, C., Heipieper, H.J., Plewniak, F., Marchal, M., Simon, S., Arsène-Ploetze, F. and Bertin, P.N. 2012. Surface properties and intracellular speciation revealed an original adaptive mechanism to arsenic in the acid mine drainage bio-indicator *Euglena mutabilis*. *Applied Microbiology and Biotechnology*, 93(4), pp.1735-1744.

- Hanagata, N., Takeuchi, T., Fukuju, Y., Barnes, D.J. and Karube, I. 1992.
  Tolerance of microalgae to high CO<sub>2</sub> and high temperature. *Phytochemistry*, *31*(10), pp.3345-3348.
- Hansen, J., Sato, M., Kharecha, P., Beerling, D., Berner. R., Mansson-Delmotte. V., Pagani, M., Raymo. M., Royer, D. L.and Zachos, J. C. 2008. Target atmospheric CO<sub>2</sub> where should humanity aim. *Open Atmospheric Science Journal*, 2, pp. 217-231.
- Hansen, P.J. 2002. Effect of high pH on the growth and survival of marine phytoplankton: Implications for species succession. *Aquatic Microbial Ecology*,28(3), pp.279-288.
- Harder, R. and Witsch, V. H. 1942. Uber Massenkultur von Diatomeen. *Aquatic Microiology and Ecology, 60*, pp.146-152.
- Hargreaves, J.W. and Whitton, B.A. 1976. Effect of pH on growth of acid stream algae. *British Phycological Journal*, 11(3), pp.215-223.
- Herzog, H., Drake, E. and Adams, E.1997. CO<sub>2</sub> Capture, re-use and storage technologies for mitigating global climate change. White Paper Final Report, Energy Laboratory. Massachusetts Institute of Technology.US Department of Energy Order No: DE-AF22-96PC01257.
- Hirata, S., Hayashitani, M., Taya, M. and Tone, S. 1996. Carbon dioxide fixation in batch culture of *Chlorella* sp. using a photobioreactor with a sunlight-cellection device. *Journal of Fermentation and Bioengineering*, 81(5), pp.470-472.
- Ho, S.H., Chan, M.C., Liu, C.C., Chen, C.Y., Lee, W.L., Lee, D.J. and Chang,
   J.S. 2014. Enhancing lutein productivity of an indigenous microalga
   *Scenedesmus obliquus* FSP-3 using light-related
   strategies. *Bioresource Technology*, 152, pp.275-282.

- Ho, S.H., Chen, C.Y., Lee, D.J. and Chang, J.S. 2011. Perspectives on microalgal CO<sub>2</sub>-Emission mitigation systems-A review. *Biotechnology Advances*, 29(2), pp.189-198.
- Ho, S.H., Chen, W.M. and Chang, J.S. 2010. Scenedesmus obliquus CNW-N as a potential candidate for CO<sub>2</sub> mitigation and biodiesel production. *Bioresource Technology*, 101(22), pp.8725-8730.
- Ho, S.H., Kondo, A., Hasunuma, T. and Chang, J.S. 2013. Engineering strategies for improving the CO<sub>2</sub> fixation and carbohydrate productivity of *Scenedesmus obliquus* CNW-N used for bioethanol fermentation. *Bioresource Technology*, 143, pp.163-171.
- Hsueh, H.T., Li, W.J., Chen, H.H. and Chu, H. 2009. Carbon bio-fixation by photosynthesis of *Thermosynechococcus* sp. CL-1 and *Nannochloropsis oculta*. *Journal of Photochemistry and Photobiology B: Biology*, 95(1), pp.33-39.
- Hu, Q., Kurano, N., Kawachi, M., Iwasaki, I. and Miyachi, S. 1998.
  Ultrahigh-cell-density culture of a marine green alga *Chlorococcum littorale* in a flat-plate photobioreactor. *Applied Microbiology and Biotechnology*, 49(6), pp.655-662.
- Huang, Y.T. and Su, C.P. 2014. High lipid content and productivity of microalgae cultivating under elevated carbon dioxide. *International Journal of Environmental Science and Technology*, 11(3), pp.703-710.
- Huntley, M.E. and Redalje, D.G. 2007. CO<sub>2</sub> mitigation and renewable oil from photosynthetic microbes: A new appraisal. *Mitigation and Adaptation strategies for Global change*, *12*(4), pp.573-608.
- IPCC (2007) The physical science basis. In: Solomon SD, Qin D, Manning M, Chen Z, Marquie M, Averyt KB, Tignor M, Miller HL (eds)

Contribution of Working Group I to the Forth assessment report of the IPCC on climate change. Cambridge University Press, Cambridge.

- Ismaiel, M.M.S., El-Ayouty, Y.M. and Piercey-Normore, M. 2016. Role of pH on antioxidants production by *Spirulina (Arthrospira) platensis*. *Brazilian journal of microbiology*, 47(2), pp.298-304.
- Jana, A., Bhattacharya, P., Guha, S., Ghosh, S. and Majumdar, S. 2017. Application of a new ceramic hydrophobic membrane for providing CO<sub>2</sub> in algal photobioreactor during cultivation of *Arthrospira* sp. *Algal Research*, 27, pp.223-234.
- Jeong, M.L., Gillis, J.M. and Hwang, J.Y. 2003. Carbon dioxide mitigation by microalgal photosynthesis. *Bulletin of the Korean Chemical Society*, 24, pp.1763-1766.
- Jia, X., Yan, H., Wang, Z., He, H., Xu, Q., Wang, H., Yin, C. and Liu, L. 2011. Carbon dioxide fixation by *Chlorella* sp. USTB-01 with a fermentor-helical combined photobioreactor. *Frontiers of Environmental Science and Engineering in China*, 5(3), pp.402-408.
- Jiang, L., Luo, S., Fan, X., Yang, Z. and Guo, R. 2011. Biomass and lipid production of marine microalgae using municipal wastewater and high concentration of CO<sub>2</sub>. *Applied Energy*, 88(10), pp.3336-3341.
- Jiang, Y., Peng, X., Zhang, W. and Liu, T. 2012. Enhancement of acid resistance of *Scenedesmus dimorphus* by acid adaptation. *Journal of Applied Phycology*, 24,pp.1637–1641.
- Jiang, Y., Zhang, W., Wang, J., Chen, Y., Shen, S. and Liu, T. 2013. Utilization of simulated flue gas for cultivation of *Scenedesmus dimorphus*. *Bioresource Technology*, 128, pp.359-364.

- Jian-Rong, X.I.A. and Kun-shan, G.A.O. 2002. Effects of enrichment on microstructure and ultrastructure of two species of fresh water green algae. *Acta Botanica Sinica*, 44(5), 527-531.
- Jiménez, C. and Niell, F.X. 1991. Growth of *Dunaliella viridis* Teodoresco: Effect of salinity, temperature and nitrogen concentration. *Journal of Applied Phycology*, 3(4), pp.319-327.
- Juneja, A., Ceballos, R.M. and Murthy, G.S. 2013. Effects of environmental factors and nutrient availability on the biochemical composition of algae for biofuels production: A review. *Energies*, 6(9), pp.4607-4638.
- Kadam, K.L. 1997. Power plant flue gas as a source of CO<sub>2</sub> for microalgae cultivation: Economic impact of different process options. *Energy Conversion and Management*, 38, pp.505-510.
- Kamjunke, N., Tittel, J., Krumbeck, H., Beulker, C. and Poerschmann, J. 2005. High heterotrophic bacterial production in acidic, iron-rich mining lakes. *Microbial Ecology*, 49(3), pp.425-433.
- Kao, C.Y., Chen, T.Y., Chang, Y.B., Chiu, T.W., Lin, H.Y., Chen, C.D., Chang, J.S. and Lin, C.S. 2014. Utilization of Carbon dioxide in industrial flue gases for the cultivation of microalga *Chlorella* sp. *Bioresource Technology*, *166*, pp. 485-493.
- Kao, C.Y., Chiu, S.Y., Huang, T.T., Dai, L., Hsu, L.K. and Lin, C.S. 2012. Ability of a mutant strain of the microalga *Chlorella* sp. to capture carbon dioxide for biogas upgrading. *Applied Energy*, 93, pp.176-183.
- Kargupta, W., Ganesh, A. and Mukherji, S. 2015. Estimation of Carbon dioxide sequestration potential of microalgae grown in a batch photobioreactor. *Bioresource Technology*, 180, pp.370-375.

- Kaštánek, F., Šabata, S., Šolcová, O., Maléterová, Y., Kaštánek, P., Brányiková, I., Kuthan, K. and Zachleder, V. 2010. In-field experimental verification of cultivation of microalgae *Chlorella* sp. using the flue gas from a cogeneration unit as a source of carbon dioxide. *Waste Management and Research*, 28(11), pp.961-966.
- Keffer, J.E. and Kleinheinz, G.T. 2002. Use of *Chlorella vulgaris* for CO<sub>2</sub> mitigation in a photobioreactor. *Journal of Industrial Microbiology* and Biotechnology, 29(5), pp.275-280.
- Ketchum, B.H., Lillick, L. and Redfield, A.C. 1949. The growth and optimum yields of unicellular algae in mass culture. *Journal of Cellular Physiology*, 33(3), pp.267-279.
- Khairy, H.M., Shaltout, N.A., El-Naggar, M.F. and El-Naggar, N.A. 2014.
   Impact of elevated CO<sub>2</sub> concentrations on the growth and ultrastructure of non-calcifying marine diatom (*Chaetoceros gracilis* F. Schütt). *The Egyptian Journal of Aquatic Research*, 40(3), pp.243-250.
- Khalil, Z.I., Asker, M.M., El-Sayed, S. and Kobbia, I.A. 2010. Effect of pH on growth and biochemical responses of *Dunaliella bardawil* and *Chlorella ellipsoidea*. World Journal of Microbiology and Biotechnology, 26(7), pp.1225-1231.
- Khatoon, H., Rahman, N.A., Banerjee, S., Harun, N., Suleiman, S.S., Zakaria, N.H., Lananan, F., Hamid, S.H.A. and Endut, A. 2014. Effects of different salinities and pH on the growth and proximate composition of *Nannochloropsis* sp. and *Tetraselmis* sp. isolated from South China Sea cultured under control and natural condition. *International Biodeterioration and Biodegradation*, 95, pp.11-18.

- Kodama, M. 1993. A new species of highly CO<sub>2</sub> tolerant fast growing marine microalga suitable for high-density culture. *Journal of Marine Biotechnology*, 1, pp.21-25.
- Kondili, E.M. and Kaldellis, J.K. 2007. Biofuel implementation in East Europe: Current status and future prospects. *Renewable and Sustainable Energy Reviews*, 11(9), pp.2137-2151.
- Krinsky, N.I. and Johnson, E.J. 2005. Carotenoid actions and their relation to health and disease. *Molecular aspects of medicine*, *26*(6), pp.459-516.
- Kumar, A., Ergas, S., Yuan, X., Sahu, A., Zhang, Q., Dewulf, J., Malcata, F.X. and Van Langenhove, H., 2010. Enhanced CO<sub>2</sub> fixation and biofuel production via microalgae: recent developments and future directions. *Trends in biotechnology*, 28(7), pp.371-380.
- Kumar, K., Banerjee, D. and Das, D. 2014. Carbon dioxide sequestration from industrial flue gas by *Chlorella sorokiniana*. *Bioresource Technology*, 152, pp.225-233.
- Kumar, K., Dasgupta, C.N., Nayak, B., Lindblad, P. and Das, D. 2011. Development of suitable photobioreactors for CO<sub>2</sub> sequestration addressing global warming using green algae and cyanobacteria. *Bioresource Technology*, 102(8), pp.4945-4953.
- Kuo, C.M., Jian, J.F., Lin, T.H., Chang, Y.B., Wan, X.H., Lai, J.T., Chang, J.S. and Lin, C.S. 2016. Simultaneous microalgal biomass production and CO<sub>2</sub> fixation by cultivating *Chlorella* sp. GD with aquaculture wastewater and boiler flue gas. *Bioresource Technology*, 221, pp.241-250.
- Kurano, N., Ikemoto, H., Miyashita, H., Hasegawa, T., Hata, H. and Miyachi,S. 1995. Fixation and utilization of Carbon dioxide by microalgal

photosynthesis. *Energy Conversion and Management*, *36*(6-9), pp.689-692.

- Lakaniemi, A.M., Hulatt, C.J., Thomas, D.N. and Puhakka, J.A. 2012. Carbon dioxide utilization in gas-sparge microalgal photobioreactors. *Conference paper presented: Asian Biohydrogen, Bioproducts Symposium. Chongqing, China.*
- Lam, M.K. and Lee, K.T. 2011. Renewable and sustainable bioenergies production from palm oil mill effluent (POME): win-win strategies toward better environmental protection. *Biotechnology Advances*, 29(1), pp.124-141.
- Lam, M.K., Lee, K.T. and Mohamed, A.R. 2012. Current status and challenges on microalgae-based carbon capture. *International Journal of Greenhouse Gas Control*, *10*, pp.456-469.
- Lane, A.E. and Burris, J.E. 1981. Effects of environmental pH on the internal pH of Chlorella pyrenoidosa, Scenedesmus quadricauda and Euglena mutabilis. Plant Physiology, 68(2), pp.439-442.
- Lee, J.H., Lee, J.S., Shin, C.S., Park, S.C. and Kim, S.W. 2000. Effects of NO and SO<sub>2</sub> on growth of highly-CO<sub>2</sub>-tolerant microalgae. *Journal of Microbiology and Biotechnology*, 10(3), pp.338-343.
- Lee, J.S. and Lee, J.P. 2003. Review of advances in biological CO<sub>2</sub> mitigation technology. *Biotechnology and Bioprocess Engineering*, 8(6), pp.354-359.
- Lee, J.Y., Hong, M.E., Chang, W.S. and Sim, S.J. 2015. Enhanced Carbon dioxide fixation of *Haematococcus pluvialis* using sequential operating system in tubular photobioreactors. *Process Biochemistry*, 50(7), pp.1091-1096.

- Lee, Y.K. and Tay, H.S. 1991. High CO<sub>2</sub> partial pressure depresses productivity and bioenergetic growth yield of *Chlorella pyrenoidosa* culture. *Journal of applied phycology*, *3*(2), pp.95-101.
- Li, F.F., Yang, Z.H., Zeng, R., Yang, G., Chang, X., Yan, J.B. and Hou, Y.L.
   2011. Microalgae capture of CO<sub>2</sub> from actual flue gas discharged from a combustion chamber. *Industrial and Engineering Chemistry Research*, 50(10), pp.6496-6502.
- Liu, B.H. and Lee, Y.K. 2000. Secondary carotenoids formation by the green alga *Chlorococcum* sp. *Journal of Applied Phycology*, *12*(3-5), pp.301-307.
- Liu, Z.G. and Lu, G.L. 1990. The sterilizing studies of flagellate and ciliate in marine unicellular algae liquid. *Zhanjiang Aquacult. Coll*, *2*, pp.36-41.
- López, C.G., Fernández, F.A., Sevilla, J.F., Fernández, J.S., García, M.C. and Grima, E.M. 2009. Utilization of the cyanobacteria *Anabaena* sp. ATCC 33047 in CO<sub>2</sub> removal processes. *Bioresource Technology*, 100(23), pp.5904-5910.
- Lorenz, R.T. and Cysewski, G.R. 2000. Commercial potential for *Haematococcus* microalgae as a natural source of Astaxanthin. *Trends in Biotechnology*, 18(4), pp.160-167.
- Lowry, O.H., Rosebrough, N.J., Farr, A.L. and Randall, R.J. 1951. Protein measurement with the Folin phenol reagent. *Journal of Biological Chemistry*, 193(1), pp.265-275.
- Ma, M., Yuan, D., He, Y., Park, M., Gong, Y. and Hu, Q. 2017. Effective control of *Poterioochromonas malhamensis* in pilot-scale culture of *Chlorella sorokiniana* GT-1 by maintaining CO<sub>2</sub>-mediated low culture pH. *Algal Research*, 26, pp. 436-444.

- Maity, J.P., Bundschuh, J., Chen, C.Y. and Bhattacharya, P. 2014. Microalgae for third generation biofuel production, mitigation of greenhouse gas emissions and wastewater treatment: Present and future perspectives-A mini review. *Energy*, 78, pp.104-113.
- Malis-Arad, S. and McGowan, R.E. 1982. A "Point of No Return" in the Cell Cycle of *Chlorella*. *Plant and Cell Physiology*, *23*(3), pp.397-401.
- Malis-Arad, S. and McGrowan, R.E. 1982. Alkalinity-induced aggregation in *Chlorella vulgaris* II. Changes in the cell wall during the cell cycle. *Plant and Cell Physiology*, 23(1), pp.11-17.
- Mandotra, S.K., Kumar, P., Suseela, M.R., Nayaka, S. and Ramteke, P.W. 2016. Evaluation of fatty acid profile and biodiesel properties of microalga *Scenedesmus abundans* under the influence of Phosphorus, pH and light intensities. *Bioresource technology*, 201, pp. 222-229.
- Manjre, S.D. and Deodhar, M.A. 2013. Screening of thermotolerant microalgal species isolated from western ghats of Maharashtra, India for CO<sub>2</sub> sequestration. *Journal of Sustainable Energy and Environment*, 4(2).pp.61-67.
- Maraskolhe, V.R., Warghat, A.R. and Charan, G. 2012. Carbon sequestration potential of *Scenedesmus* species (Microalgae) under the fresh water ecosystem. *African Journal of Agricultural Research*, 7(18), pp.2818-2823.
- Matsumoto, H., Shioji, N., Hamasaki, A., Ikuta, Y., Fukuda, Y., Sato, M., Endo, N. and Tsukamoto, T., 1995. Carbon dioxide fixation by microalgae photosynthesis using actual flue gas discharged from a boiler. *Applied Biochemistry and Biotechnology*, 51(1), pp.681-692.

- Mayo, A.W. 1997. Effects of temperature and pH on the kinetic growth of unialga *Chlorella vulgaris* cultures containing bacteria.*Water EnvironmentResearch*,69(1), pp.64-72.
- McGinn, P.J., Dickinson, K.E., Bhatti, S., Frigon, J.C., Guiot, S.R. and O'Leary, S.J. 2011. Integration of microalgae cultivation with industrial waste remediation for biofuel and bioenergy production: Opportunities and limitations. *Photosynthesis Research*, 109(1-3), pp.231-247.
- Miao, X. and Wu, Q. 2006. Biodiesel production from heterotrophic microalgal oil. *Bioresource Technology*, *97*(6), pp. 841-846.
- Milledge, J.J. 2011. Commercial application of microalgae other than as biofuels: A brief review. *Reviews in Environmental Science and Bio/Technology*, 10(1), pp.31-41.
- Minillo, A., Godoy, H.C. and Fonseca, G.G. 2013. Growth Performance of Microalgae Exposed to CO<sub>2</sub>. Journal of Clean Energy Technologies, 1(2), pp.110-114.
- Minowa, T., Yokoyama, S.Y., Kishimoto, M. and Okakura, T. 1995. Oil production from algal cells of *Dunaliella tertiolecta* by direct thermochemical liquefaction. *Fuel*, *74*(12), pp.1735-1738.
- Miura, Y., Yamada, W., Hirata, K., Miyamoto, K. and Kiyohara, M. 1993. Stimulation of hydrogen production in algal cells grown under high CO<sub>2</sub> concentration and low temperature. *Applied Biochemistry and Biotechnology*, 39(1), pp.753-761.
- Miyachi, S., Iwasaki, I. and Shiraiwa, Y. 2003. Historical perspective on microalgal and cyanobacterial acclimation to low and extremely high-CO<sub>2</sub> conditions. *Photosynthesis research*, 77(2-3), pp.139-153.

- Miyachi, S., Tsuzuki, M., Maruyama, I., Gantar, M., Miyachi, S. and Matsushima, H. 1986. Effects of CO<sub>2</sub> concentration during growth on the intracellular structure of *Chlorella* and *Scenedesmus* (Chlorophyta) 1. *Journal of Phycology*, 22(3), pp.313-319.
- Miyairi, S. 1995. CO<sub>2</sub> assimilation in a thermophilic cyanobacterium. *Energy conversion and management*, *36*(6-9), pp.763-766.
- Moazami, N., Ashori, A., Ranjbar, R., Tangestani, M., Eghtesadi, R. and Nejad, A.S. 2012. Large-scale biodiesel production using microalgae biomass of *Nannochloropsis*. *Biomass and Bioenergy*, 39, pp.449-453.
- Moheimani, N.R. 2013. Inorganic carbon and pH effect on growth and lipid productivity of *Tetraselmis suecica* and *Chlorella* sp. (Chlorophyta) grown outdoors in bag photobioreactors. *Journal of Applied Phycology*, 25(2), pp.387-398.
- Molina, E., Fernández, J., Acién, F.G. and Chisti, Y. 2001. Tubular photobioreactor design for algal cultures. *Journal of Biotechnology*, 92(2), pp.113-131.
- Moreno-Garrido, I. and Canavate, J.P. 2001. Assessing chemical compounds for controlling predator ciliates in outdoor mass cultures of the green algae *Dunaliella* salina. *Aquacultural Engineering*, *24*(2), pp.107-114.
- Moss, B. 1973. The Influence of environmental factors on the distribution of freshwater algae: An experimental study: II. The role of pH and the Carbon dioxide-bicarbonate system. *The Journal of Ecology*, *61*, pp.157-177.
- Munir, N., Imtiaz, A., Sharif, N. and Naz, S. 2015. Optimization of growth conditions of different algal strains and determination of their lipid contents. *Journal of Animal and Plant Sciences*, 25(2), pp.546-553.

- Muradyan, E.A., Klyachko-Gurvich, G.L., Tsoglin, L.N., Sergeyenko, T.V. and Pronina, N.A. 2004. Changes in lipid metabolism during adaptation of the *Dunaliella salina* photosynthetic apparatus to high CO<sub>2</sub> concentration. *Russian Journal of Plant Physiology*, 51(1), pp.53-62.
- Mutanda, T., Ramesh, D., Karthikeyan, S., Kumari, S., Anandraj, A. and Bux, F. 2011. Bioprospecting for hyper-lipid producing microalgal strains for sustainable biofuel production. *Bioresource Technology*, 102(1), pp. 57-70.
- Muthuraj, M., Kumar, V., Palabhanvi, B. and Das, D. 2014. Evaluation of indigenous microalgal isolate *Chlorella* sp. FC2 IITG as a cell factory for biodiesel production and scale up in outdoor conditions. *Journal* of *Industrial Microbiology and Biotechnology*, 41(3), pp.499-511.
- Myers, J. and Clark, U.L. 1944. Culture conditions and the development of the photosynthetic mechanism: II. An apparatus for the continuous culture of *Chlorella*. *The Journal of General Physiology*, 28(2), pp.103-112.
- Naderi, G., Tade, M.O. and Znad, H. 2015. Modified photobioreactor for biofixation of carbon dioxide by *Chlorella vulgaris* at different light intensities. *Chemical Engineering and Technology*, 38(8), pp.1371-1379.
- Nagase, H., Eguchi, K., Yoshihara, K.I., Hirata, K. and Miyamoto, K. 1998.
   Improvement of microalgal NO<sub>x</sub> removal in bubble column and airlift reactors. *Journal of Fermentation and Bioengineering*, 86(4), pp.421-423.

- Nalewajko, C., Colman, B. and Olaveson, M. 1997. Effects of pH on growth, photosynthesis, respiration, and copper tolerance of three *Scenedesmus* strains. *Environmental and Experimental Botany*, 37(2-3), pp.153-160.
- Nayak, M., Rath, S.S., Thirunavoukkarasu, M., Panda, P.K., Mishra, B.K. and Mohanty, R.C. 2013. Maximizing biomass productivity and CO<sub>2</sub> biofixation of microalga, *Scenedesmus* sp. by using Sodium hydroxide. *Journal of Microbiolology and Biotechnology*, 23(9), pp.1260-1268.
- Negoro, M., Shioji, N., Miyamoto, K. and Micira, Y. 1991. Growth of microalgae in high CO<sub>2</sub> gas and effects of SO<sub>x</sub> and NO<sub>x</sub>. *Applied Biochemistry and Biotechnology*, *28*(1), pp.877-886.
- Nixdorf, B., Krumbeck, H., Jander, J. and Beulker, C. 2003. Comparison of bacterial and phytoplankton productivity in extremely acidic mining lakes and eutrophic hard water lakes. *Acta Oecologica*, *24*, pp.281-288.
- Nogia, P., Sidhu, G.K., Mehrotra, R. and Mehrotra, S. 2013. Capturing atmospheric carbon: Biological and non-biological methods. *International Journal of Low-Carbon Technologies*, 11(2), pp.266-274.
- Ohta, H., 1997. CO<sub>2</sub> separation and removal technologies in measures to cope with global warming. *The Separation Technology*, *27*(5), pp.312-319.
- Olaizola, M. 2003. Microalgal removal of CO<sub>2</sub> from flue gases: Changes in medium pH and flue gas composition do not appear to affect the photochemical yield of microalgal cultures. *Biotechnology and Bioprocess Engineering*, 8(6), pp.360-367.
- Olivier, J.G.J., Janssens-Maenhout, G., Muntean, M. and Peters, J.A.H.W., 2015. Trends in global CO2 emissions: 2015 Report. PBL Netherlands

Environmental Assessment Agency, The Hague; European Commission, Joint Research Centre (JRC).

- Olaizola, M., Bridges, T., Flores, S., Griswold, L., Morency, J. and Nakamura, T. 2004. Microalgal removal of CO<sub>2</sub> from flue gases: CO<sub>2</sub> capture from a coal combustor Third Annual Conference on Carbon Capture and Sequestration. *Alexandria, VA*.
- Ono, E. and Cuello, J.L. 2006. Feasibility assessment of microalgal Carbon dioxide sequestration technology with photobioreactor and solar collector. *Biosystems Engineering*, 95(4), pp.597-606.
- Ono, E. and Cuello, J.L. 2007. Carbon dioxide mitigation using thermophilic cyanobacteria. *Biosystems Engineering*, *96*(1), pp.129-134.
- Ota, M., Kato, Y., Watanabe, H., Watanabe, M., Sato, Y., Smith Jr, R.L. and Inomata, H. 2009. Fatty acid production from a highly CO<sub>2</sub> tolerant alga, *Chlorocuccum littorale*, in the presence of inorganic carbon and nitrate. *Bioresource Technology*, 100(21), pp.5237-5242.
- Packer, M. 2009. Algal capture of Carbon dioxide; biomass generation as a tool for greenhouse gas mitigation with reference to New Zealand energy strategy and policy. *Energy Policy*, 37(9), pp.3428-3437.
- Pancha, I., Chokshi, K., Maurya, R., Trivedi, K., Patidar, S.K., Ghosh, A. and Mishra, S. 2015. Salinity induced oxidative stress enhanced biofuel production potential of microalgae *Scenedesmus* sp. CCNM 1077. *Bioresource Technology*, 189, pp.341-348.
- Pandey, J.P., Pathak, N. and Tiwari, A. 2010. Standardization of pH and Light Intensity for the Biomass Production of *Spirulina platensis*. *Journal of Algal Biomass Utilization*, 1(2), pp.93-102.

- Papazia, A., Makridis, P., Divanach, P. and Kotzabasis, K. 2008. Bioenergetic changes in the microalgal photosynthetic apparatus by extremely high CO<sub>2</sub> concentrations induce an intense biomass production. *Physiologia Plantarum*, 132(3), pp.338-349.
- Patil, L. and Kaliwal, B. 2017. Effect of CO<sub>2</sub> concentration on growth and biochemical composition of newly isolated indigenous microalga *Scenedesmus bajacalifornicus* BBKLP-07. *Applied Biochemistry and Biotechnology*, 182(1), pp.335-348.
- Pegallapati, A.K. and Nirmalakhandan, N. 2011. Energetic evaluation of an internally illuminated photobioreactor for algal cultivation. *Biotechnology Letters*, *33*(11), pp.2161-2167.
- Pelizer, L.H., Carvalho, J.C.M., Sato, S. and de Oliveira Moraes, I. 2002. *Spirulina platensis* growth estimation by pH determination at different cultivations conditions. *Electronic Journal of Biotechnology*, 5(3), pp.17-18.
- Peng, L., Lan, C.Q., Zhang, Z., Sarch, C. and Laporte, M. 2015. Control of protozoa contamination and lipid accumulation in *Neochloris oleoabundans* culture: Effects of pH and dissolved inorganic carbon. *Bioresource Technology*, 197, pp.143-151.
- Philipose, M.T.1967. *Chlorococcales* Monograph. Indian Council of Agricultural Research, New Delhi, p.365
- Piazzi, L. and Ceccherelli, G. 2002. Effects of competition between two introduced Caulerpa. *Marine Ecology Progress Series*, 225, pp.189-195.
- Poerschmann, J., Spijkerman, E. and Langer, U. 2004. Fatty acid patterns in *Chlamydomonas* sp. as a marker for nutritional regimes and

temperature under extremely acidic conditions. *Microbial Ecology*, *48*(1), pp.78-89.

- Ponnuswamy, I., Madhavan, S., Shabudeen, S. and Shoba, U.S. 2014. Resolution of lipid content from algal growth in carbon sequestration studies. *International Journal of Advanced Science and Technology*, 67, pp.23-32.
- Pope, D.H. 1975. Effects of light intensity, oxygen concentration, and Carbon dioxide concentration on photosynthesis in algae. *Microbial Ecology*, 2(1), pp.1-16.
- Posten, C. 2009. Design principles of photo-bioreactors for cultivation of microalgae. *Engineering in Life Sciences*, *9*(3), pp.165-177.
- Poza-Carrión, C., Fernández-Valiente, E., Piñas, F.F. and Leganés, F. 2001. Acclimation of photosynthetic pigments and photosynthesis of the cyanobacterium *Nostoc* sp. strain UAM206 to combined fluctuations of irradiance, pH, and inorganic carbon availability. *Journal of Plant Physiology*, 158(11), pp.1455-1461.
- Prescott, G.W. 1951. Algae of the Western Great Lakes area. W.C. Brown Co., Dubuque, Iowa. p.977.
- Pronina, N., Kodama, M. and Miyachi, S. 1993. Changes in intracellular pH values in various microalgae induced by raising CO<sub>2</sub> concentrations.
  In: Furuya, M. (Ed.), XV International Botanical Congress, Yokohama, p. 419.
- Radmann, E.M., Camerini, F.V., Santos, T.D. and Costa, J.A.V. 2011. Isolation and application of  $SO_X$  and  $NO_X$  resistant microalgae in biofixation of  $CO_2$  from thermoelectricity plants. *Energy Conversion and Management*, 52(10), pp.3132-3136.
- Ragauskas, A.J., Williams, C.K., Davison, B.H., Britovsek, G., Cairney, J., Eckert, C.A., Frederick, W.J., Hallett, J.P., Leak, D.J., Liotta, C.L. and Mielenz, J.R. 2006. The path forward for biofuels and biomaterials. *Science*, *311*(5760), pp. 484-489.
- Rai, M.P., Gautom, T. and Sharma, N. 2015. Effect of salinity, pH, light intensity on growth and lipid production of microalgae for bioenergy application. *Online Journal of Biological Sciences*, 15(4), pp. 260-267.
- Rai, S.V. and Rajashekhar, M. 2014. Effect of pH, salinity and temperature on the growth of six species of marine phytoplankton. *Journal of Algal Biomass Utilization*, 5(4), pp. 55-59.
- Ramanan, R., Kannan, K., Deshkar, A., Yadav, R. and Chakrabarti, T. 2010.
   Enhanced algal CO<sub>2</sub> sequestration through calcite deposition by *Chlorella* sp. and *Spirulina platensis* in a mini-raceway pond. *Bioresource Technology*, 101(8), pp.2616-2622.
- Ramanathan, V. 1988. The greenhouse theory of climate change: A test by an inadvertent global experiment. *Science*, *240* (4850), pp.293-299.
- Ramos, G.J.P., Bicudo, C.E.D.M., Góes Neto, A. and Moura, C.W.D.N. 2012. *Monoraphidium* and *Ankistrodesmus* (Chlorophyceae, Chlorophyta) from Pantanal dos Marimbus, Chapada Diamantina, Bahia State, Brazil. *Hoehnea*, 39(3), pp.421-434.
- Raven, J.A. 1981. Nutrient transport in microalgae. In: Rose, A.H., Morris, J.G. (Eds.), Advances in Microbial Physiology, Academic Press, London. p.226.
- . Razzak, S.A., Ilyas, M., Ali, S.A.M. and Hossain, M.M. 2015. Effects of CO<sub>2</sub> concentration and pH on mixotrophic growth of *Nannochloropsis*

oculata. Applied Biochemistry and Biotechnology, 176(5), pp.1290-1302.

- Resnik, K.P., Yeh, J.T. and Pennline, H.W. 2004. Aqua ammonia process for simultaneous removal of CO<sub>2</sub>, SO<sub>2</sub> and NOx. *International Journal of Environmental Technology and Management*, 4(1-2), pp.89-104.
- Richmond, A. and Grobbelaar, J.U. 1986. Factors affecting the output rate of *Spirulina platensis* with reference to mass cultivation. *Biomass*, 10(4), pp.253-264.
- Rocha, J.M., Garcia, J.E. and Henriques, M.H. 2003. Growth aspects of the marine microalga *Nannochloropsis gaditana*. *Biomolecular Engineering*, 20(4-6), pp.237-242.
- Rotatore, C. and Colman, B. 1991. The acquisition and accumulation of inorganic carbon by the unicellular green alga *Chlorella ellipsoidea*. *Plant, Cell and Environment, 14*(4), pp.377-382.
- Ryu, H.J., Oh, K.K. and Kim, Y.S. 2009. Optimization of the influential factors for the improvement of CO<sub>2</sub> utilization efficiency and CO<sub>2</sub> mass transfer rate. *Journal of Industrial and Engineering Chemistry*, 15(4), pp.471-475.
- Sadeghizadeh, A., Moghaddasi, L. and Rahimi, R. 2017. CO<sub>2</sub> capture from air by *Chlorella vulgaris* microalgae in an airlift photobioreactor. *Bioresource Technology*, 243, pp.441-447.
- Sakarika, M. and Kornaros, M. 2016. Effect of pH on growth and lipid accumulation kinetics of the microalga *Chlorella vulgaris* grown heterotrophically under sulfur limitation. *Bioresource Technology*, 219, pp. 694-701.

- Salih, F.M. 2011. Microalgae tolerance to high concentrations of Carbon dioxide: A review. *Journal of Environmental Protection*, 2(05),pp. 648-654.
- Salim, M.A. 2013. The Growth of *Ankistrodesmus* sp in response to CO<sub>2</sub> induction. *Journal of Asian Scientific Research*, *3*(1), p.75-84.
- Sasaki, T., Pronina, N.A., Maeshima, M., Iwasaki, I., Kurano, N. and Miyachi, S. 1999. Development of vacuoles and vacuolar H<sup>+</sup>-ATPase activity under extremely high CO<sub>2</sub> conditions in *Chorococcum littorale* cells. *Plant Biology*, 1(1), pp.68-75.
- Satoh, A., Kurano, N., Senger, H. and Miyachi, S. 2002. Regulation of energy balance in photosystems in response to changes in CO<sub>2</sub> concentrations and light intensities during growth in extremely-high-CO<sub>2</sub>-tolerant green microalgae. *Plant and Cell Physiology*, 43(4), pp.440-451.
- Sawayama, S., Inoue, S., Dote, Y. and Yokoyama, S.Y. 1995. CO<sub>2</sub> fixation and oil production through microalga. *Energy Conversion and Management*, 36(6-9), pp.729-731.
- Seckbach, J., Baker, F.A. and Shugarman, P.M. 1970. Algae thrive under pure CO<sub>2</sub>. *Nature*, *227*(5259), pp.744-745.
- Sergeenko, T.V., Muradyan, E.A., Pronina, N.A., Klyachko-Gurvich, G.L., Mishina, I.M. and Tsoglin, L.N. 2000. The effect of extremely high CO<sub>2</sub> concentration on the growth and biochemical composition of microalgae. *Russian Journal of Plant Physiology*, 47(5), pp.632-638.
- Shah, S.M.U., Radziah, C.C., Ibrahim, S., Latiff, F., Othman, M.F. and Abdullah, M.A. 2014. Effects of photoperiod, salinity and pH on cell growth and lipid content of *Pavlova lutheri*. *Annals of Microbiology*, 64(1), pp.157-164.

- Sharma, K.K., Schuhmann, H. and Schenk, P.M. 2012. High lipid induction in microalgae for biodiesel production. *Energies*, *5*(5), pp.1532-1553.
- Sharmila, K., Ramya, S. and Ponnusami, A.B. 2014. Carbon sequestration using microalgae-a review. *International Journal of ChemTech Research*, 6(9), pp. 4128-4134.
- Shi, S.Y., Liu, Y.D., Shen, Y.W., Li, G.B., Li, D.H. 2006. Lysis of Aphanizomenon fols-aquae (Cyanobacteria) by a bacteria Bacillus cereus Biological Control, 39, pp.345-351.
- Shoaf, W.T. and Lium, B.W. 1976. Improved extraction of chlorophyll a and b from algae using Dimethyl sulfoxide. Limnology and Oceanography, 21(6), pp.926-928.
- Siefermann-Harms, D. 1985. Carotenoids in photosynthesis. I. Location in photosynthetic membranes and light-harvesting function. *Biochimica et Biophysica Acta (BBA)-Reviews on Bioenergetics*, 811(4), pp.325-355.
- Sierra, E., Acién, F.G., Fernández, J.M., García, J.L., González, C. and Molina, E. 2008. Characterization of a flat plate photobioreactor for the production of microalgae. *Chemical Engineering Journal*, 138(1-3), pp.136-147.
- Silva, H.J. and Pirt, S.J. 1984. Carbon dioxide inhibition of photosynthetic growth of *Chlorella*. *Microbiology*, *130*(11), pp.2833-2838.
- Singh, U.B. and Ahluwalia, A.S. 2013. Microalgae: A promising tool for carbon sequestration. *Mitigation and Adaptation strategies for Global change*, 18(1), pp.73-95.

- Sirisansaneeyakul, S., Singhasuwan, S., Choorit, W., Phoopat, N., Garcia, J.L. and Chisti, Y. 2011. Photoautotrophic production of lipids by some *Chlorella* strains. *Marine Biotechnology*, 13(5), pp. 928-941.
- Skjånes, K., Lindblad, P. and Muller, J. 2007. Bio CO<sub>2</sub>–A multidisciplinary, biological approach using solar energy to capture CO<sub>2</sub> while producing H<sub>2</sub> and high value products. *Biomolecular Engineering*, 24(4), pp.405-413.
- Slegers, P.M., Wijffels, R.H., Van Straten, G. and Van Boxtel, A.J.B. 2011. Design scenarios for flat panel photobioreactors. *Applied Energy*, 88(10), pp.3342-3353.
- Snow, J.W. 1902. The plankton algae of Lake Erie, with special reference to the Chlorophyceae. *US Fish Com Bull*, pp.371-394.
- Sobczuk, T.M., Camacho, F.G., Rubio, F.C., Fernández, F.A. and Grima, E.M. 2000. Carbon dioxide uptake efficiency by outdoor microalgal cultures in tubular airlift photobioreactors. *Biotechnology and Bioengineering*, 67(4), pp.465-475.
- Solovchenko, A. and Khozin-Goldberg, I. 2013. High-CO<sub>2</sub> tolerance in microalgae: possible mechanisms and implications for biotechnology and bioremediation. *Biotechnology letters*, *35*(11), pp.1745-1752.
- Somchai, W. Chalor, L. and Niti, C.H. 2015. Effect of pH on blue green algae isolated from pacific while shrimp pond, Kasetart University Fisheries Research Bulletin, 32(1), pp.1-9.
- Šoštarič, M., Klinar, D., Bricelj, M., Golob, J., Berovič, M. and Likozar, B. 2012. Growth, lipid extraction and thermal degradation of the microalga *Chlorella vulgaris*. *New Biotechnology*, 29(3), pp.325-331.

- Spolaore, P., Joannis-Cassan, C., Duran, E. and Isambert, A. 2006. Commercial applications of microalgae. *Journal of Bioscience and Bioengineering*, 101(2), pp.87-96.
- Stewart, C. and Hessami, M.A., 2005. A study of methods of carbon dioxide capture and sequestration-the sustainability of a photosynthetic bioreactor approach. *Energy Conversion and Management*, 46(3), pp.403-420.
- Suali, E., Sarbatly, R., Anisuzzaman, S.M., Lahin, F.A., Asidin, M.A. and Jusnukin, T. 2017. Effect of membrane on carbonation and carbon dioxide uptake of *Chlorella* sp. *MATEC Web of Conferences*, 111, pp.02004.
- Sukenik, A., Zmora, O. and Carmeli, Y. 1993. Biochemical quality of marine unicellular algae with special emphasis on lipid composition. II. *Nannochloropsis* sp. *Aquaculture*, 117(3-4), pp.313-326.
- Sydney, E.B., Sturm, W., de Carvalho, J.C., Thomaz-Soccol, V., Larroche, C., Pandey, A. and Soccol, C.R. 2010. Potential Carbon dioxide fixation by industrially important microalgae. *Bioresource Technology*, 101(15), pp.5892-5896.
- Takano, H. and Matsunaga, T. 1995. CO<sub>2</sub> fixation by artificial weathering of waste concrete and coccolithophorid algae cultures. *Energy Conversion and Management*, 36(6-9), pp.697-700.
- Takeuchi, T., Utsunomiya, K., Kobayashi, K., Owada, M. and Karube, I., 1992. Carbon dioxide fixation by a unicellular green alga *Oocystis* sp. *Journal of biotechnology*, 25(3), pp.261-267.
- Tamiya, H. 1957. Mass culture of algae. *Annual Review of Physiology.8*, pp.309-334.

- Tang, D., Han, W., Li, P., Miao, X. and Zhong, J. 2011. CO<sub>2</sub> biofixation and fatty acid composition of *Scenedesmus* obliquus and *Chlorella pyrenoidosa* in response to different CO<sub>2</sub> levels. *Bioresource Technology*, 102(3), pp.3071-3076.
- Taraldsvik, M. and Myklestad, S.M. 2000. The effect of pH on growth rate, biochemical composition and extracellular carbohydrate production of the marine diatom *Skeletonema costatum*. *European Journal of Phycology*, 35(2), pp.189-194.
- Tatsuzawa, H., Takizawa, E., Wada, M. and Yamamoto, Y. 1996. Fatty acid and lipid composition of the acidophilic green alga *Chlamydomonas* sp.1. *Journal of Phycology*, 32(4), pp.598-601.
- Thampy, K.G. and Wakil, S.J. 1985. Activation of acetyl-CoA carboxylase. Purification and properties of a Mn<sup>2+</sup> dependent phosphatase. *Journal of Biological Chemistry*, 260(10), pp.6318-6323.
- Toerien, D.F and Grobbelaar, U. 1980. Algal mass production experiments in South Africa. In: Shelef,G. and Soeder, C.J. (Eds.), *Algae Biomass, Production and Use*, Elsevier. pp.73-80.
- Toledo-Cervantes, A., Morales, M., Novelo, E. and Revah, S. 2013. Carbon dioxide fixation and lipid storage by *Scenedesmus* obtusiusculus. Bioresource Technology, 130, pp.652-658.
- Touloupakis, E., Cicchi, B. and Torzillo, G. 2015. A bioenergetic assessment of photosynthetic growth of *Synechocystis* sp. PCC 6803 in continuous cultures. *Biotechnology for Biofuels*, 8(1), pp.1-10.
- Tripathi, R., Singh, J. and Thakur, I.S. 2015. Characterization of microalga Scenedesmus sp. ISTGA1 for potential CO<sub>2</sub> sequestration and biodiesel production. Renewable Energy, 74, pp.774-781.

- Trivedi, R.K., Goel, P.K. and Trisal, C.L. 1987. Practical methods in Ecology and Environmental sciences. Environ. Public. Karad, (India).
- Tsuzuki, M. and Miyachi, S. 1989. The function of carbonic anhydrase in aquatic photosynthesis. *Aquatic botany*, *34*(1-3), pp.85-104.
- Tsuzuki, M., Gantar, M., Aizawa, K. and Miyachi, S. 1986. Ultrastructure of *Dunaliella tertiolecta* cells grown under low and high CO<sub>2</sub> concentrations. *Plant and Cell Physiology*, 27(4), pp.737-739.
- Tubea, B., Hawxby, K. and Mehta, R. 1981. The effects of nutrient, pH and herbicide levels on algal growth. *Hydrobiologia*, *79*(3), pp.221-227.
- Van der Westhuizen, A.J. and Eloff, J.N. 1983. Effect of culture age and pH of culture medium on the growth and toxicity of the blue-green alga *Microcystis aeruginosa*. Zeitschrift für Pflanzenphysiologie, 110(2), pp.157-163.
- Van Wagenen, J., Miller, T.W., Hobbs, S., Hook, P., Crowe, B. and Huesemann, M. 2012. Effects of light and temperature on fatty acid production in *Nannochloropsis salina*. *Energies*, 5(3), pp.731-740.
- Varshney, P., Sohoni, S., Wangikar, P.P. and Beardall, J. 2016. Effect of high CO<sub>2</sub> concentrations on the growth and macromolecular composition of a heat and high-light-tolerant microalga. *Journal of Applied Phycology*, 28(5), pp.2631-2640.
- Vasudevan, P.T. and Briggs, M. 2008. Biodiesel production-current state of the art and challenges. *Journal of Industrial Microbiology and Biotechnology*, 35(5), pp.421-430.
- Velea, S., Dragos, N., Serban, S., Ilie, L., Stalpeanu, D., Nicoara, A. and Stepan, E. 2009. Biological sequestration of Carbon dioxide from

thermal power plant emissions, by absorbtion in microalgal culture media. *Romanian Biotechnological Letters*, *14*(4), pp.4485-4500.

- Verduin, J. 1964. Principles of primary productivity: Photosynthesis under completely natural conditions. In *Algae and man.* Springer, pp.221-238.
- Vidyashankar, S., Deviprasad, K., Chauhan, V.S., Ravishankar, G.A. and Sarada, R. 2013. Selection and evaluation of CO<sub>2</sub> tolerant indigenous microalga *Scenedesmus dimorphus* for unsaturated fatty acid rich lipid production under different culture conditions. *Bioresource Technology*, 144, pp.28-37.
- Visviki, I. and Santikul, D. 2000. The pH tolerance of *Chlamydomonas* applanata (Volvocales, Chlorophyta). Archives of Environmental Contamination and Toxicology, 38(2), pp.147-151.
- Vonshak, A. 1997. Appendix III-Growth media and conditions for Spirulina. In :Vonshak, A. (Ed.), Spirulina platensis (Arthrospira): Physiology, Cell-biology and Biotechnology, Taylor and Francis, USA, pp.218-219.
- Wang, B., Li, Y., Wu, N. and Lan, C.Q. 2008. CO<sub>2</sub> bio-mitigation using microalgae. *Applied microbiology and Biotechnology*, 79(5), pp.707-718.
- Wang, H., Zhang, W., Chen, L., Wang, J. and Liu, T. 2013. The contamination and control of biological pollutants in mass cultivation of microalgae. *Bioresource technology*, 128, pp.745-750.
- Wassink, E.C., Kok, B. and Oorschot, J.L.P. 1953. Experiments on photosynthesis by *Chlorella* in flushing light. *Algal Culture from Laboratory to Pilot Plain*, 600, pp.55-62.

- Watanabe, Y. and Hall, D.O. 1995. Photosynthetic CO<sub>2</sub> fixation technologies using a helical tubular bioreactor incorporating the filamentous cyanobacterium *Spirulina platensis*. *Energy Conversion and Management*, 36(6-9), pp.721-724.
- Watanabe, Y., Ohmura, N. and Saiki, H., 1992. Isolation and determination of cultural characteristics of microalgae which functions under CO<sub>2</sub> enriched atmosphere. *Energy Conversion and Management*, 33(5-8), pp.545-552.
- Wijffels, R.H. and Barbosa, M.J. 2010. An outlook on microalgal biofuels. *Science*, 329(5993), pp.796-799.
- Wilson, M.H., Groppo, J., Placido, A., Graham, S., Morton, S.A., Santillan-Jimenez, E., Shea, A., Crocker, M., Crofcheck, C. and Andrews, R. 2014. CO<sub>2</sub> recycling using microalgae for the production of fuels. *Applied Petrochemical Research*, 4(1), pp.41-53.
- Xiao, R., Chen, R., Zhang, H.Y. and Li, H. 2011. Microalgae Scenedesmus quadricauda grown in digested waste water for simultaneous CO<sub>2</sub> fixation and nutrient removal. Journal of Biobased Materials and Bioenergy, 5(2), pp.234-240.
- Xu, X., Song, C., Andresen, J.M., Miller, B.G. and Scaroni, A.W. 2002.
   Novel polyethylenimine-modified mesoporous molecular sieve of MCM-41 type as high-capacity adsorbent for CO<sub>2</sub> capture. *Energy & Fuels*, *16*(6), pp.1463-1469.
- Yadav, G., Karemore, A., Dash, S.K. and Sen, R. 2015. Performance evaluation of a green process for microalgal CO<sub>2</sub> sequestration in closed photobioreactor using flue gas generated in-situ. *Bioresource technology*, 191, pp.399-406.

- Yang, Y. and Gao, K. 2003. Effects of CO<sub>2</sub> concentrations on the freshwater microalgae, *Chlamydomonas reinhardtii*, *Chlorella pyrenoidosa* and *Scenedesmus obliquus* (Chlorophyta). *Journal of Applied Phycology*, 15(5), pp.379-389.
- Yeh, J.T., Pennline, H.W. and Resnik, K.P. 2001. Study of CO<sub>2</sub> absorption and desorption in a packed column. *Energy and Fuels*, 15(2), pp.274-278.
- Yeh, K.L., Chang, J.S. and chen, W.M. 2010. Effect of light supply and carbon source on cell growth and cellular composition of a newly isolated microalga *Chlorella vulgaris* ESP-31. *Engineering in Life Sciences*, 10(3), pp.201-208.
- Ying, K., Zimmerman, W.B. and Gilmour, D.J. 2014. Effects of CO<sub>2</sub> and pH on growth of the microalga Dunaliella *salina*. *Journal of Microbial and Biochemical Technology*, 6(3), pp. 167-173.
- Yoo, C., Jun, S.Y., Lee, J.Y., Ahn, C.Y. and Oh, H.M. 2010. Selection of microalgae for lipid production under high levels carbon dioxide. *Bioresource Technology*, 101(1), pp.71-74.
- Yue, L. and Chen, W. 2005. Isolation and determination of cultural characteristics of a new highly CO<sub>2</sub> tolerant fresh water microalgae. *Energy Conversion and Management*, 46(11-12), pp.1868-1876.
- Yun, H.S., Ji, M.K., Park, Y.T., Salama, E.S. and Choi, J. 2016. Microalga, *Acutodesmus obliquus* KGE 30 as a potential candidate for CO2 mitigation and biodiesel production. *Environmental Science and Pollution Research*, 23(17), pp.17831-17839.

- Yun, Y.S. and Park, J.M. 1997. Development of gas recycling photobioreactor system for microalgal Carbon dioxide fixation. *Korean Journal of Chemical Engineering*, 14(4), pp.297-300.
- Zeiler, K.G., Heacox, D.A., Toon, S.T., Kadam, K.L. and Brown, L.M. 1995. The use of microalgae for assimilation and utilization of Carbon dioxide from fossil fuel-fired power plant flue gas. *Energy Conversion* and Management, 36(6-9), pp.707-712.
- Zeng, X., Danquah, M.K., Chen, X.D. and Lu, Y. 2011. Microalgae bioengineering: from CO<sub>2</sub> fixation to biofuel production. *Renewable* and Sustainable Energy Reviews, 15(6), pp.3252-3260.
- Zhang, D.H., Ng, Y.K. and Phang, S.M. 1997. Composition and accumulation of secondary carotenoids in *Chlorococcum* sp. *Journal of Applied Phycology*, 9(2), pp. 147-155.
- Zhang, K., Miyachi, S. and Kurano, N. 2001. Evaluation of a vertical flatplate photobioreactor for outdoor biomass production and Carbon dioxide bio-fixation: Effects of reactor dimensions, irradiation and cell concentration on the biomass productivity and irradiation utilization efficiency. *Applied Microbiology and Biotechnology*, 55(4), pp.428-433.
- Zhang, Q., Wang, T. and Hong, Y. 2014. Investigation of initial pH effects on growth of an oleaginous microalgae *Chlorella* sp. HQ for lipid production and nutrient uptake. *Water Science and Technology*, 70(4), pp.712-719.
- Zhao, B. and Su, Y. 2014. Process effect of microalgal-carbon dioxide fixation and biomass production: A review. *Renewable and Sustainable Energy Reviews*, *31*, pp.121-132.

- Zhao, B., Su, Y., Zhang, Y. and Cui, G. 2015. Carbon dioxide fixation and biomass production from combustion flue gas using energy microalgae. *Energy*, *89*, pp.347-357.
- Zhou, W.L., Qiao, X.T., Sun, J.F., Xing, K.Z. and Tang, X.X. 2011. Ecological effect of z-qs01 strain on *Chlorella vulgaris* and its response to UV-B radiation stress. *Procedia Environmental Sciences*, 11, pp.741-748.

#### PLATE 6B Photobioreactor



#### PLATE 1 Pure cultures of microalgae

# 1b. Culture of Chlamydomonas globosa 1a. Culture of Chlamydomonas grovei



1d. Culture of Monoraphidium contortum 1c. Culture of Desmodesmus opoliensis



1e. Culture of Acutodesmus obliquus

1f. Culture rack



#### PLATE 3 Experimental setup



### PLATE 4

# Response of microalgae to varing range of pH

### Chlamydomonas globosa



3a. First day

3b. Seventh day

### Acutodesmus obliquus



3c. First day

3d. Seventh day

### PLATE 5

# Figures representing the growth responses of Chlamydomonas globosa (a) and Acutodesmus obliquus (b) to varying pH ranges.



## Technical specification of the PBR facility used for the present study PLATE 6A



- a. CO<sub>2</sub> supply facility with flow regulator b. Culture medium containing
  - microalgae
    - c. Illumination source

- d. Outflow gas regulator
  e. Outflow gas monitoring facility (reservoir)
  f. Pure Bolds Basal medium